

## **GRADUATE STUDY PROGRAM**

## "METALLURGY"

NOTE: revised study program in application from academic year 2017/2018

## **PROGRAM OF THE COURSES**

Sisak, July, 2017

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1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE: 169729				
1.1. Course teacher	Assoc.Prof. Ljerka Slokar	1.6. Year of study	1.	
1.2. Name of the course	PHYSICAL METALLURGY	1.7. Credit value (ECTS)	5	
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	45+0+30+0	
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30	
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%	
2. COURSE DESCRIPTION				
2.1. Course objectives	Developing the ability to connect theoretical Deepening the theoretical understanding of Programming desired properties of metals	al knowledge and properties of metals. of reactions between metals and in metals.		
2.2. Enrolment requirements and required entry competences for the course				
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in Design the properties of metallic materials Compare the procedures of material treatmeters	engineering practice. nent with microstructure and useful properties.		
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Explain the classification, formation, structure and morphology of the important group of steels. Analyze and describe the procedures for designing the desired properties of metallic materials. Explain models and growth kinetics of selected metal materials. Describe the fatigue and fracture of metals from the standpoint of microstructure.			
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (45): Introduction. Classification of iron and steel according to the composition, manufacturing process, generic type, microstructure, final processing, the use of (4). Ferrite; the formation, structure and morphology. Ferritic steels. The influence of various elements on the properties of ferritic steels. The kinetics and mechanism of transformation. (4) Austenite; formation, kinetics, morphology. Austenitization in one-and two-phase area. Size and form of austenitic grain. The transformation of austenite during cooling. (4) Cementite; Primary, secondary and tertiary. The structure and morphology. Perlite, formation and forms. Interlamellar space and properties. (2) Ferrite-pearlite and pearlite steels. Hypo-and hyper eutectoid steels. Growth kinetics and structure (s) of pearlite. Intermediate precipitation. Spheroidization of pearlite. (4) Bainite. Mechanism of bainite forming. The upper and lower bainite. The morphology of specific types of bainite. Alloying elements impact. Bainitic steels. (4) Martensite. Characteristics and conditions of martensitic transformation. A crystallographic model of martensitic formation. Needle, thermoelastic martensite. Athermic and isothermal martensite. (4)			

	nucleation and growth in martensitic transformation. Memory forms in martensitic transformation. (4) Recovery and recrystallization of metals. (3) The precipitation hardening (or hardening) of metals. (2) Segregation of metals. Macro-and microsegregation types and causes of segregation. Distribution law. Zone refining of metals. Fatigue and fracture of metals. (4) Composite metals. Particle and fiber-reinforced composites, structural composites. Compatibility; physical, chemical, mechanical and electrical properties. (3) Amorphous metals and whiskery. Synthesis, structure and properties. Nanometals. Production, properties, use (3) EXERCISES (30): Auditory exercises: Phase diagram Fe-Fe <sub>3</sub> C, determination of microconstituents in specific types of steel, computational tasks in the field of composite materials (6). Laboratory exercises: metallographic preparation of various types of steel, composites, amorphous metals (12), and their						
2.6. Type of instruction	<ul> <li>Interview of the second second</li></ul>		indep multir labora	endent study nedia and the atory with the mento (other)	internet or	2.7. Comments:	
2.8. Student responsibilities	Attending lectures and auditory	/ exercises	(min.70%	b), committed i	eports from lab	poratory exercises.	
2.9. Screening of student's work (specify the proportion of ECTS credits for each activity so that the total number of CTS credits is equal to	Class attendance Experimental work Essay Tests	0.5 0.5 2.0	R R S C	esearch eport eminar essay oral exam	1.0	Practical training (Otherdescribe) (Other describe)	
the credit value of the course)):	Written exam	1.0	Р	roject		(Other— describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	The final grade is determined b	by score of t	three colle	oquiums and a	assessments of	written and oral exams resp	ectively.
	Title		Number of copies at the library		Availability via other me	dia	
2.11. Required literature (available at	R. W. Cahn, Physical Metallurg Publ. Comp., Amsterdam, 1997	gy, Nort-Hol ′.	lland	1	CD		
the library and via other media)	R. E. Smallman, R. J. Bishop, Metallurgy and Materials Engin Butterworth, Oxford, 1999.	Modern Phy neering,	ysical	1	CD		
	T. Matković, P. Matković, Lj. Sl metalima – Zbirka rješenih zad	okar, Znano ataka, Sisa	ost o k, 2010.		https://www.si sveucilisni-stu preddiplomsk	imet.unizg.hr/nastava/predav idij-metalurgija/2-godina- og/Zadaci%20iz%20FM%20r	anja/preddiplomski- e-TNR-Boja-

			<u>1.pdf/view</u>		
	T. Matković, P. Matković; Fizikalna metalurgija I (skripta), Metalurški fakultet, Sisak, 2009.		https://www.simet.unizg.hr/nastava/predavanja/preddiplomski- sveucilisni-studij-metalurgija/2-godina- preddiplomskog/Fizikalna%20metalurgija%201.pdf/view		
2.12. Optional literature (at the time of	W. D. Callister, Materials Science and Engineering,	J. Wiley& So	ns, New York, 1996.		
the submission of the study programme proposal)	W. C. Leslie, Physical Metallurgy of Steels, McGraw	/-Hill, Tokyo,	1982.		
2.13. Methods of monitoring quality that	Anonymous survey on the level of the Faculty and University.				
ensure acquisition of exit	Analysis provided by system of quality assurance institutions.				
competences	Analyses provided by quality assurance system and	l authorized c	offices of the University.		

Ordinal number	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning outcomes
1	Explain the classification, formation, structure and morphology of the important group of steels.	1st colloquium, auditory exercises, laboratory exercises, written and oral exam
2	Analyze and describe the procedures for designing the desired properties of metallic materials.	1st colloquium, auditory exercises, laboratory exercises, written and oral exam
3	Explain models and growth kinetics of selected metal materials.	2nd colloquium, auditory exercises, written and oral exam
4	Describe the fatigue and fracture of metals from the standpoint of microstructure.	2nd colloquium, laboratory exercises, written and oral exam
5	Explain the characteristics and structural features of composite, nano and amorphous metals.	3rd colloquium, auditory exercises, laboratory exercises, written and oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE: 169730				
1.1. Course teacher	Full Prof. Ladislav Lazić, PhD	1.6. Year of study	1	
1.2. Name of the course	INDUSTRIAL FURNACES	1.7. Credit value (ECTS)	5	
1.3. Associate teachers	Ivan Jandrlić, PhD	1.8. Type of instruction (number of hours L+S+E+e-learning)	45+0+30+0	
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30	
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%	
2. COURSE DESCRIPTION				
2.1. Course objectives	<ol> <li>Acquire knowledge about the furnaces</li> <li>Acquire knowledge about the special f</li> <li>Acquire knowledge about the furnaces</li> <li>Acquire knowledge about the electric f</li> <li>Inform students about the current statule</li> </ol>	for production of metal materials. urnaces for production of metal materials. and devices for heating of metal materials. urnaces and devices for heating of metal mater is and directions in development of industrial fu	rials. Irnaces.	
2.2. Enrolment requirements and required entry competences for the course	The acquired knowledge from the courses of undergraduate study: Metallurgy of iron, Steel metallurgy, Refractory and carbon materials, Engineering thermodynamics and Heat and mass transfer.			
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Suggest new and improved technical and technological solutions.</li> <li>Analyse the development and application of new technologies.</li> <li>Analyse the production processes by applying thermodynamic laws.</li> <li>Plan and manage metallurgical processes.</li> <li>Analyse the material and thermal balance of metallurgical processes.</li> <li>Formulate and suggest measures for increasing energy officiency.</li> </ol>			
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Analyze and identify problems in the work of the concrete industrial furnace and propose technical solutions.</li> <li>Determine computationally the profile and dimensions of the furnace space.</li> <li>Create individual structural elements of the furnace and select materials for their construction.</li> <li>Calculate the individual periods and regimes of heating of metal materials.</li> <li>Analyze and propose solutions for increasing the energy efficiency of a furnace.</li> <li>Analyze and propose solutions for reducing emissions of pollutants into the environment.</li> <li>Propose technical solutions for cleaning the waste gases.</li> </ol>			
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ol> <li>LECTURES (45):</li> <li>General considerations about the request of process industries. Furnaces of mare elements of the furnace. (3)</li> <li>Furnaces of process industries. Blast for raw materials in the blast furnace. A systems. (3)</li> <li>Cupola furnaces. The structural elemetair nozzles. Pre-small furnaces. Air pre-small furnaces. A</li></ol>	irements that the furnace should satisfy. The d nufacturing industry. Classification of furnaces a urnaces. The structural elements of the blast fu ir nozzles and slag outlet. Cleaning of furnace nts of cupola furnaces. Steel sheet, chimney an eheating, conventional and radiation recuperato	ivision of industrial furnaces. Furnaces and individual features. Structural urnace. Devices for filling-in (loading) gas. Air preheat, cowpers and cowper and spark chaser. Annular chamber and ors. (3)	

	<ul> <li>tilting. Spear for oxygen. Spear cooling. Cleaning of converter gas. Supplying of technical oxygen. (3)</li> <li>Electric arc furnaces. The structural elements of electric arc furnaces. Portal support of vault. The device for carrying and lowering the electrodes. Electric regime of furnace. Highly efficient electric arc furnaces. Plasma-arc furnaces for remelting of metals. (3)</li> <li>Induction electric furnace. High-frequency electric furnace. Electric resistance furnace. Furnaces with electron beam. Furnaces with plasma jet. (3)</li> <li>Movable and immovable mixers. Construction of mixers. Mixer supplied with burners. Lining of mixers and their use. (3) 1st colloquium</li> <li>Furnaces of manufacturing industry. Rolling mill soaking pits and their development. Contemporary soaking pits. Operating efficiency of furnace and fuel consumption. Rolling mill pusher furnace and their development. Constructional characteristics of the furnace. (3)</li> <li>Thermal-technical zones of pusher furnaces. Temperature and thermal regime of furnace. The possibility of changing of dimensions and shape of the load. Furnaces with one-sided heating and two-sided heating (bottom firing). Thermal-technical zones of furnaces and their conditionality. (3)</li> <li>Walking-beam furnaces and their conditionality. (3)</li> <li>Rotary hearth furnaces for rolling mills of seamless pipes. Plate and ring furnaces. Thermal-technical zones of furnaces and their mediate space. Profiled rollers for transporting the load. Heat exchangers. Electric furnaces. (3)</li> <li>Construction of segments and cylinder intermediate space. Profiled rollers for transporting the load. Heat exchangers. Electric furnaces for preheating of metal materials. The application of inductive heating in colling mills. (3)</li> <li>Forging furnaces in plants for metal forging. Directly and indirectly heating. Dualchamber furnaces of preheating and heating of material. Furnaces with annealing conditions. (3)</li> <li>Construction of ceramic radiant tubes. Bell</li></ul>					nism for rrying and or remelting eeam. use. (3) s. Operating cteristics of changing of mal- urnaces and ngers. g and he use of ing the
			2nd colloquium			
	EXERCISES (30): The unders projects are selected so that the the selected so that the selected so that the selected so that the selected so that the selected selected so that the selected sel	tanding of the standing of the	he material exposed in lecture the presented theory and illust	s is facilitated trate the appli	by solving the practical proble cation of theory to real probler	ems. The ns.
2.6. Type of instruction	<ul> <li>➢ lectures</li> <li>➢ seminars and workshops</li> <li>➢ exercises</li> <li>➢ online in entirety</li> <li>➢ mixed e-learning</li> <li>➢ field work</li> </ul>		<ul> <li>independent study</li> <li>multimedia and the intern</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>	et	2.7. Comments:	
2.8. Student responsibilities	Attendance to Lectures and Ex	xercises > 7	0 %			
2.9. Screening of student's work (specify	Class attendance	0.5	Research		Practical training	
the proportion of ECTS credits for	Experimental work		Report			
each activity so that the total number	Essay		Seminar essay		(Otherdescribe)	
of CTS credits is equal to the credit	Tests	1.25	Oral exam	2.0	(Other-describe)	

value of the course)):	Written exam	Project	1.25	(Other-describe	e)
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Projects – 30% Class attendance – 5% Written exam – 30% Oral exam – 35%				
0.44 Demined literature (oveileble et	Title			Number of copies at the library	Availability via other media
2.11. Required literature (available at	J. Črnko, Industrijske peći, Metalurški fakultet, Sisak, 2010.			5	
the library and via other media)	J. Črnko, Zbirka zadataka iz projektiranja industrijskih peći, Metalurški fakultet, Sisak, 2008.			2	
1.12. Optional literature (at the time of the submission of the study programme proposal)	P. Mullinger, B. Jenkins, Indust J. Ward, R. Collins, Industrial fu	rial and Process Furnaces, Elsevi urnace technology, Centro de ene	er, 2008. rgia e technologia, Ric	o Tinto, 2003.	
1.13. Methods of monitoring quality that	Survey at the level of faculty ar	nd University.			
ensure acquisition of exit	Analyses provided in the system of quality assurance of the institution.				
competences	Analyses provided in the system	m of quality assurance and an aut	horized office of the U	niversity.	

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Analyze and identify problems in the work of the concrete industrial furnace and propose	1st colloquium, oral exam
	technical solutions.	
2	Determine computationally the profile and dimensions of the furnace space.	1st colloquium, oral exam
3	Create individual structural elements of the furnace and select materials for their construction.	1nd colloquium, oral exam
4	Calculate the individual periods and regimes of heating of metal materials.	2nd colloquium, project, oral exam
5	Analyze and propose solutions for increasing the energy efficiency of a furnace.	2nd colloquium, oral exam
6	Analyze and propose solutions for reducing emissions of pollutants into the environment.	2nd colloquium, oral exam
7	Propose technical solutions for cleaning the waste gases.	2nd colloquium, oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE: 169731					
1.1. Course teacher	Full Prof. Stoja Rešković, PhD	1.1. Year of study	1		
1.2. Name of the course	THEORY OF METAL FORMING	1.2. Credit value (ECTS)	5		
1.3. Associate teachers	Tin Brlić, mag.ing.met.	1.3. Type of instruction (number of hours L+S+E+e-learning)	30+30+0+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.4. Expected enrolment in the course	30		
1.5. Status of the course	compulsory	1.5. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	2., 10%		
2. COURSE DESCRIPTION					
2.1. Course objectives	<ol> <li>Introduce students to the physical-chen</li> <li>Introduce students to the mechanical-m</li> <li>Introducing students to the scientific pri</li> <li>Acquired knowledge applied to metal for</li> </ol>	nical theory of deformation nathematical theory of deformation nciples of deformation prming processes			
2.2. Enrolment requirements and required entry competences for the course	-				
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Design and apply the modelling of metallurgical and other processes.</li> <li>Recognize and apply scientific principles important in the field of metallurgy.</li> <li>Use the acquired theoretical knowledge in engineering practice.</li> </ol>				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Set a hypothesis on metal forming,</li> <li>Create stress - strain diagram at the forming process,</li> <li>Create dependence diagrams of stress and strain on process parameters,</li> <li>Analyze the influential parameters on forming process,</li> <li>Predict the behaviour of different metals during deformation,</li> <li>Examine deformation resistance of metal materials,</li> <li>Analyze process parameters at different deformation processes,</li> <li>Valorise the deformation parameters at different deformation processes,</li> <li>Apply theoretical knowledge to solve engineering problems in practice,</li> <li>Set hypothesis on influence of individual factors of deformation process, design and conduct an experiment, analyze and</li> </ol>				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ul> <li>be trypeness of individual factors of deformation process, design and conduct an experiment, analyze and present the results.</li> <li>LECTURES (30) AND SEMINARS (30): <ol> <li>Introduction. 1</li> <li>Physical and chemical theory. 1</li> <li>Deformation of Monocrystal. Mechanisms of metal plastic flow (slip and twinning). 2</li> <li>Deformation of polycrystalline. 1</li> <li>Cold plastic deformation, hardening curves of I, II and III order. 2</li> <li>Hot plastic deformation. Strengthening and softening mechanisms during deformation. 1</li> </ol> </li> </ul>				

	9. Seminar: Different	topics from ph	nysical theory of deformation.	4			
	1 Colloquium, chapters 1-0						
	11 Stresses on an incl	lined plane 1	of deformation. State of site	55, 50°55 Z			
	12 Stresses in the boo	dv that rotates	around its axis 1				
	13. Seminar - determin	nation of stress	s. 2				
	14. Main normal and m	nain shear stre	esses. 2				
	15. Stress tensor. 1	. Stress tensor. 1					
	16. Seminar - determir	nation of the m	nain normal and shear stress	es for differer	t stress state conditions. 4		
	17. Graphic method of	7. Graphic method of determination of stress. 2					
	18. Seminar - graphica	al determinatio	n of stress. 4				
	19. Stress scheme and	d deformation	scheme. The law of equilibriu	um. 1			
	20. Deformation theory	/. 1					
	21. Conditions of flow.	The law of flo	w. 1				
	22. The strain rate. 1						
	23. Seminar - determin	hing the conne	ection between stress and str	ain for variou	s conditions of stress state. 4		
	24. Non-uniformity of a	elormation.	as Slip line method Deform	otion work M	athed Method of modelling 2		
	25. Determination of de	potion of dofor	rmation force for different cor	ditions of str	errod. Method of modelling. 2		
	20. Seminal – determin				ess state. 4		
	28 Seminar – determi	ne the deform:	ation work for different condit	tions of stress	s state 4		
	II Colloquium char	oters 10-27					
	29. Theory of deformat	tion process. 7	Theoretical basis of rolling. 1				
	30. Theoretical basics	of forging and	pressing. 1				
	31. Theoretical basis o	of extrusion. Th	neoretical basics of drawing.	1			
	32. Seminar - defining	stress state a	nd calculation of deformation	forces and d	eformation work for different technology	ogies of	
	shaping by deformation	n. 4				-	
			independent study		2.7. Comments:		
	Seminars and works	shops	multimedia and the internet				
2.6. Type of instruction			<ul> <li>Iaboratory</li> <li>work with the mentor</li> </ul>				
			(other)				
	Attendance at lectures	min.70%, ind	ividual work on all exercises	and preparati	on and submission of reports from field	eld of	
2.8. Student responsibilities	colloguium before writi	ng the collogu	ium or the written exam.				
2.9. Screening of student's work (specify	Class attendance		Research		Practical training		
the proportion of ECTS credits for	Experimental work	0.5	Report				
each activity so that the total	Essay		Seminar essay	0.5	(Otherdescribe)		
number of CTS credits is equal to	Tests	2.5	Oral exam	1.0	(Other—describe)		
the credit value of the course)):	Written exam		Project	0.5	(Other-describe)		
2.10. Grading and evaluation of student	During the classes are	evaluated the	presence and activity of stu	dents on clas	Ses.		
work over the course of instruction	Students score particip	pation on proje	ects and their work on specifi	c project.			

and at a final exam	Score of written colloquium trough continuous monitoring (or written exam) and oral exam. Score of seminar paper.			
	Title	Number of copies at the library	Availability via other media	
2.11. Required literature (available at the library and via other media)	S. Rešković, Teorija oblikovanja deformiranjem, Sveučilište u Zagrebu, Metalurški fakultet, Sisak 2014., peer reviewed lessons		https://www.simet.unizg.hr/nastava/predavanja/dipl omski-sveucilisni-studij-metalurgija/1-godina- diplomskog- studija/S%20Reskovic%20TEORIJA%20OBLIKOV ANJA%20DEFORMIRANJEM.pdf/view	
	I. Mamuzić, Teorija plastične deformacije metala, MF, Sisak, 2000.	10		
	M. Čaušević, Teorija plastične prerade, Svjetlost, Sarajevo, 1979.	6		
2.12. Optional literature (at the time of the submission of the study programme proposal)	B. Grizelj, Oblikovanje metala deformiranjem, Sveučilište J. J. Strossmayera u Osijeku, Strojarski fakultet u Slavonskom Brodu. Professional journals, articles from this area.			
2.13. Methods of monitoring quality that	Survey on the level of faculty and University.			
ensure acquisition of exit	Analyses provided by quality assurance syste	em of the institution.		
competences	Analyses provided by quality assurance syste	em and authorized office of the	University.	

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Set a hypothesis on metal forming.	1st colloquium, seminar paper, written and oral exam
2	Create stress - strain diagram at the forming process.	Seminar paper, independent task
3	Create dependence diagrams of stress and strain on process parameters.	Seminar paper, independent task
4	Analyze the influential parameters on forming process.	2nd colloquium, written and oral exam
5	Predict the behaviour of different metals during deformation.	2nd colloquium, seminar paper, written and oral exam
6	Examine deformation resistance of metal materials.	Independent task
7	Analyze process parameters at different deformation processes.	Seminar paper, oral exam
8	Valorise the deformation parameters at different deformation processes.	Seminar paper, independent task, oral exam
9	Apply theoretical knowledge to solve engineering problems in practice.	Project task
10	Set hypothesis on influence of individual factors of deformation process, design and conduct an	3rd colloquium, seminar paper, project task
	experiment, analyze and present the results.	

1. COURSE DECRIPTION - GENERAL	INFORMATION	ISV	<b>/U CODE:</b> 169732		
1.1. Course teacher	Assoc.Prof. Vladimir Grozdanić, PhD Assoc.Prof. Anita Begić Hadžipašić, PhD	1.6. Year of study	1		
1.2. Name of the course	THEORY OF METALLURGICAL PROCESSES	1.7. Credit value (ECTS)	5		
1.3. Associate teachers	-	<ol> <li>Type of instruction (number of hours L+S+E+e-learning)</li> </ol>	30+0+30+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30		
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	Introduction to application of thermodynami	ics and kinetics to metallurgical processes.			
2.2. Enrolment requirements and required entry competences for the course	-				
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Analyse the production processes by applying thermodymanic laws.</li> <li>Recognize and apply scientific principles important in the field of metallurgy.</li> <li>Plan and manage metallurgical processes</li> </ol>				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Analysing of oxidation reactions of impurities.</li> <li>Explain reactions of desulfurization, deoxidation and degazation.</li> <li>Comparison and estimate of defects during solidification.</li> <li>Explain kinetics of metallurgical processes</li> </ol>				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30) AND EXERCISES (30): 1. Introduction and review of subjects. 2.Classification of metallurgical processes. 3. Metals in liquid state. 4. Principles of refining of pig iron. 5. Oxidation of carbon. 6. Oxidation of silicon. 7. Oxidation of manganese. 8. Oxidation of phosphor. 9. Desulfurization. 10. Deoxidation and reoxidation. 11. Degazation. 12. Porosity and shrinkage cavity. 13. Kinetics of metallurgical processes. 14. Crystalization. 15. Sample theory and equilibrium constant	ts of reactions.			

	Image: Sector ses       Image: Image: Sector ses       Image: Image: Image: Sector ses       Image: Image: Sector ses       Image: Sector		independent study	independent study2.7.multimedia and the internetlaboratory		2.7. Comments:	
2.6. Type of instruction			multimedia and the int laboratory				
2.8. Student responsibilities	Conditions for signature: at Conditions for taking: -	Itendance	to lectures and exercises mi	n. 70 %.			
2.9. Screening of student's work (specify	Class attendance	0.5	Research		Practical training		
the proportion of ECTS credits for	Experimental work		Report				
each activity so that the total	Essay		Seminar essay		(Otherdescribe)		
number of CTS credits is equal to	Tests	2.0	Oral exam	1.5	(Other-describe)		
the credit value of the course)):	Written exam	1.0	Project		(Other-describe)		
2.10. Grading and evaluation of student work over the course of instruction	Written exam: 50% Oral exam: 50%						
and at a final exam					Normali and C		
	Title			copies at the library	Availability via other media		
2.11. Required literature (available at	A. Rosina, Teorija metalurških procesov, NTF, Ljubljana, 1994.			1			
the library and via other media)	F. Oeters, Metalurgie der Stahlerstellung, Springer, Berlin, 1989.			1			
	T. Rosenquist, Principles of Extractive Metalurgy, 2nd ed., McGraw-Hill, Singapore, 1986.			'e, 1			
2.12. Optional literature (at the time of the submission of the study	B. Dobovšek, Metalurške ž	lindre, NT	F, Ljubljana, 1989.				
programme proposal)							
2.13. Methods of monitoring quality that	Internal student survey, an	alyses pro	ovided by quality assurance s	system.			
2.13. Methods of monitoring quality that ensure acquisition of exit	Internal student survey, an Survey on the level of facu	alyses pro	ovided by quality assurance s niversity.	system.			
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Internal student survey, an Survey on the level of facu Analyses provided by quali	alyses pro Ity and Ur Ity assurat	ovided by quality assurance s niversity. nce system of the institution.	system.			

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of		
number		learning outcomes		
1	Analysing of oxidation reactions of impurities.	Auditory exercises, written and oral exam		
2	Explain reactions of desulfurization, deoxidation and degazation.	Auditory exercises, written and oral exam		
3	Comparison and estimate of defects during solidification.	Oral exam		
4	Explain kinetics of metallurgical processes.	Auditory exercises, written and oral exam		

1. COURSE DECRIPTION – GENERAL	INFORMATION	ISVU CODE: 169733			
1.1. Course teacher	Assoc.Prof. Natalija Dolić, PhD	1.6. Year of study	1		
1.2. Name of the course	NON-FERROUS METALS AND THEIR ALLOYS	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e- learning)	30+15+15+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30		
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	<ul> <li>1. Acquiring knowledge about the properties of non - ferrous metals (lead, zinc, cobalt, nickel, vanadium, molybdenum, mercury titanium, beryllium, calcium, sodium, lithium, germanium, silver and gold), raw materials for their obtaining and application areas</li> <li>2. Acquisition and understanding the basic theoretical knowledge of modern production processes non-ferrous metals (lead zinc, cobalt, nickel, vanadium, molybdenum, mercury, titanium, beryllium, calcium, sodium, lithium, germanium, silver and gold)</li> <li>3. Acquiring knowledge about the most important alloys of non-ferrous metals (lead, zinc, cobalt, nickel, molybdenum, titanium beryllium, silver and gold), their phase diagrams, properties and application areas.</li> <li>4. Training students through computational tasks to establish the basic elements of technology estimates in the production of lead zinc, and nickel</li> </ul>				
2.2. Enrolment requirements and required entry competences for the course	-				
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Combine the skills necessary for lifelong learning, including continued professional training. Plan and manage metallurgical processes.				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Explain the basic properties of non-ferrous metals (lead, zinc, cobalt, nickel, vanadium, molybdenum, mercury, titanium, beryllium, calcium, sodium, lithium, germanium, silver and gold), raw materials for their obtaining and application areas.</li> <li>Recommend a given alloy of non-ferrous metals (lead, zinc, cobalt, nickel, molybdenum, titanium, beryllium, gold and silver) depending on the required properties and applications.</li> <li>Describe modern processes for obtaining non-ferrous metals (lead, zinc, cobalt, nickel, vanadium, molybdenum, mercury, titanium, beryllium, calcium, sodium, lithium, germanium, silver and gold).</li> <li>Calculate of rational composition of zinc and lead concentrates.</li> <li>Present a seminar paper.</li> </ol>				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction to the plan course and the t ferrous metals. Obtaining non-ferrous met Heavy non-ferrous metals: Metallurgy of lead and its alloys (4) Lead and its alloys: application, mineral	ime schedule for the colloquium. The basic classification and als throughout history and their uses (1). raw materials, uses. Pyrometallurgycal and hydrometallurg	characteristics of non-		

production. Refining of crude lead. Modern processes for production lead. Lead alloys (Pb – Sb, Pb – Ca, Pb – Sn, Pb – Cu).
Metallurgy of zinc and its alloys (4) Zinc and its alloys application mineral raw materials uses. Pyrometallurgycal and hydrometallurgical processes for zinc
production. ISP process. Refining of zinc. Zinc alloys (Zn - Al, Zn - Al – Cu).
Metallurgy of cobalt and its alloys (2)
production. Refining of cobalt. Cobalt alloys (Co – Cr, Co – Cr – Mo, Cobalt L-605, Sm – Co).
Metallurgy of nickel and its alloys (2) Nickel and its alloys: application, mineral raw materials, uses. Production and refining of nickel. Nickel alloys.
Metallurgy of vanadium and its alloys (1)
Vanadium and its alloys: application, mineral raw materials, uses. Production vanadium from iron and titanum magnetite ores. Refining of vanadium. Vanadium alloys (ferrovanadium).
Metallurgy of molybdenum and its alloys (2)
Molybdenum and its alloys: application, mineral raw materials, uses. Production and refining of molybdenum. Molybdenum alloys.
<i>Metallurgy of mercury (1)</i> Properties of mercury and its compounds, mineral raw materials for production mercury. Production of mercury.
TESTSI
Light non-ferrous metals:
Metallurgy of titanium and its alloys (4) Titanium and its alloys: application, mineral raw materials, uses. Production of the metal titanium. Refining of titanium.
Production of compact, ductile titanium from sponge or powder. Titanium alloys (Ti - Al - $\alpha$ -alloys, Ti - Al - Cr, V, Cu, Mo - $\alpha$ + $\beta$ - alloys, Ti - Al - Mo, Cr, Zr - $\beta$ -alloys).
Metallurgy of beryllium and its alloys (2)
Refining of beryllium. Beryllium alloys.
<i>Metallurgy of calcium (1)</i> Calcium: application, mineral raw materials, uses. Production of calcium: electrolysis of melt, aluminothermy.
Metallurgy of sodium (1) Proportion of adjum and its compounds, minoral row materials for production adjum. Broduction of adjums electrolysis of
NaOH, electrolysis of NaCl.

	Metallurgy of lithium and its alloys Lithium and its alloys: application, and carbonate. Refining of lithium.	<i>(1)</i> mineral rav Lithium all	v materials, uses. Proc oys (Al - Li, Cu - Li, Zr	duction of lithiun - Li, Mg - Li, M	n: electrolysis of chloride, g - Al - Li, Li - Be).	electrolysis of oxide
	<i>Metallurgy of germanium</i> Germanium and its alloys: application, mineral raw materials, uses. Production of germanium. Germanium – semiconduc					
	<b>Noble metals:</b> <i>Metallurgy of silver and gold</i> Silver and gold and their alloys: application, mineral raw materials, uses. Production of silver and gold. Refining of silver and gold. Silver and gold alloys (Au - Ag-, Au - Cu, Au - Ni, Au - Pd, Au - Zn, Ag - Zn, Ag - Cd).					
	TEST II					
	<b>SEMINAR (15):</b> How to properly write seminar paper and make the best possible presentation!? (1). Phase diagrams of basic alloys (4). Preparation and presentation of seminar papers (8). Test I, II (2)					
	<b>EXERCISES (15):</b> Calculation of charge for agglomeration roasting of lead concentrates. Processes of melting lead (5). Calculation of the rational composition of zinc concentrates. Roasting process of zinc concentrates, <i>distillation</i> of agglomerate of zinc (5). Agglomeration of concentrates of nickel. Electro melting agglomerates of nickel (3). Basic elements of technology budgets for the production of magnesium by electrolysis (2).					
			independent stud	lv	2.7. Comments:	
2.6. Type of instruction	☑ seminars and workshops       ☐ independent study         ☑ exercises       ☐ multimedia and the internet         ☐ online in entirety       ☐ laboratory         ☐ mixed e-learning       ☑ field work					
2.8. Student responsibilities	Conditions for signature: regular a	ttendance (	(> 70 %), successful a	nd timely writter	and exposed seminar pa	per (ppt).
2.9. Screening of student's work	Class attendance		Research		Practical training	
(specify the proportion of ECTS	Experimental work		Report			
credits for each activity so that the	Essay		Seminar essay	1		
total number of CTS credits is equal	Tests		Oral exam	1	(Other-describe)	
to the credit value of the course)):	Written exam	2	Project		(Other-describe)	
2.10.Grading and evaluation of student	Continuous monitoring and eva	luation of	student:			
work over the course of instruction and	The exam could be passed through two TESTS (written + oral). In case it is not passed one of the two tests, the student has the					

at a final exam	right to take not passed test one more time. Both positive evaluation tests release the student of laying the final exam. At each tests student can achieve a maximum of 10 points for the question, number of questions is 5. For satisfactory accomplishment in each test student must collect more than 30 % of points for each question. If student fails the examination by tests, laying the final exam (written + oral). <i>Continuous monitoring and evaluation of student:</i> Tests (I + II), written + oral: 3 ECTS Seminar paper: 1 ECTS					
	Title	Number of copies at the library	Availability via other media			
2.11.Required literature (available at the library and via other media)	Z. Lenhard, Metalurgija obojenih metala I, Sveučilište u Zagrebu Metalurški fakultet, Sisak, 2008.		https://www.simet.unizg.hr/nastava/predavanja/preddiplo mski-sveucilisni-studij-metalurgija/2-godina- preddiplomskog/metalurgija-obojenih-metala-i/view			
	Z. Lenhard, Metalurgija obojenih metala II, Sveučilište u Zagrebu Metalurški fakultet, Sisak, 2008.		https://www.simet.unizg.hr/nastava/predavanja/diplomski- sveucilisni-studij-metalurgija/1-godina-diplomskog- studija/metalurgija-obojenih-metala-ii/view			
2.1. Optional literature (at the time of the submission of the study programme proposal)	<ul> <li>A. Vignes, Extractive Metallurgy 1, Basic Thermodynamics and Kinetics, ISTE Ltd UK and John Wiley &amp; Sons, Inc. SAD, 2011.</li> <li>R. Ž. Vračar, Teorija i praksa dobivanja obojenih metala, Savez inženjera metalurgije Srbije, Beograd, 2010.</li> <li>N. Štrbac, Ž. Živković, I. Mihajlović, Zbirka zadataka iz metalurgije obojenih metala, University of Belgrade Technical Faculty in Bor, Bor, 2004.</li> <li>R. Vračar, Ekstraktivna metalurgija cinka, Naučna knjiga, Beograd, 1997.</li> <li>R. Vračar, B. Nikolić, Ekstraktivna metalurgija olova, Naučna knjiga, Beograd, 1995.</li> <li>Handbook of Extractive Metallurgy, Volume II: Primary Metals, Secondary Metals, Light Metals, ur. F. Habashi, WILEY-VCH, Weinheim – Chichester - New York - Toronto - Brisbane - Singapore, Germany, 1997.</li> <li>C. B. Gill, Nonferrous Extractive Metallurgy, Robert E. Krieger Publisbing Company, Malabar, Florida, 1988.</li> </ul>					
2.2. Methods of monitoring quality that ensure acquisition of exit competences	Examination of students who have finished stud Survey on the faculty and University level. Analysis predicted by systems for insurance of in Analysis predicted by systems for insurance qua	y. nstitution quality. ality from authorized	University office.			

Ordinal number	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning outcomes
1	Explain the basic properties of non-ferrous metals (lead, zinc, cobalt, nickel, vanadium, molybdenum, mercury, titanium, beryllium, calcium, sodium, lithium, germanium, silver and gold), raw materials for their obtaining and application areas.	1st colloquium, 2nd colloquim, written and oral exam, seminar paper
2	Recommend a given alloy of non-ferrous metals (lead, zinc, cobalt, nickel, molybdenum, titanium, beryllium, gold and silver) depending on the required properties and applications.	1st colloquium, 2nd colloquim, written and oral exam, seminar paper
3	Describe modern processes for obtaining non-ferrous metals (lead, zinc, cobalt, nickel, vanadium, molybdenum, mercury, titanium, beryllium, calcium, sodium, lithium, germanium, silver and gold).	1st colloquium, 2nd colloquim, written and oral exam, seminar paper
4	Calculate of rational composition of zinc and lead concentrates.	1st colloquium, 2nd colloquim, written exam, auditory exercises
5	Present a seminar paper.	Seminar paper

1. COURSE DECRIPTION - GENERAL	INFORMATION		ISVU CODE: 169734		
1.1. Course teacher	Assist.Prof. Ivan Ivec, PhD	1.6. Year of study	1		
1.2. Name of the course	ENGINEERING MATHEMATICS	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30		
1.5. Status of the course	compulsory	<ul><li>1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)</li></ul>	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	<ol> <li>Use the tools of multivarable calculus in the engineering quantitative analysis.</li> <li>Use basic statistical techniques to analyse, process and display data from engineering analysis, handling correctly the accuracy of measurement.</li> <li>Apply techniques and tools of Excel in statistical analysis of data obtained by measurement or simple physical models.</li> </ol>				
2.2. Enrolment requirements and required entry competences for the course	-				
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Plan and manage the competences of analysis and synthesis.</li> </ol>				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Collect, analyse and interprete data by statistical methods.</li> <li>Explain the notion, properties and graphical interpretation of functions of several real variables.</li> <li>Use directional derivatives of functions of several variables to analyse their local behaviour.</li> <li>Use and interprete interpolation and approximation of given data with a mathematical model.</li> <li>Use basic methods of numerical integration to solve simple differential equations.</li> <li>Explain and discuss iterations in the process of solving equations.</li> <li>Explain and use methods of chosen computer tool to solve nonlinear equations.</li> </ol>				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ol> <li>Partial and directional derivative</li> <li>Local extrema of function of sev</li> <li>Conditional extrema - Lagrange</li> <li>Curves and line integrals of diffe</li> <li>Numerical solutions of nonlinear</li> <li>Least squares approximation me</li> <li>Interpolation by polynomials - La</li> <li>Numerical integration - trapezoid</li> <li>Numerical solutions of differentiat</li> <li>Graphical representation of data</li> <li>Measures of central tendency -</li> <li>Measures of dispersion - variant</li> </ol>	eral variables. multiplier method. erentials. requations - bisection method, secant ethod. agrange and Newton method. dal rule, Simpson's rule. al equations - Euler method. a - histogram, frequency polygon. mean, median, mode. ce. standard deviation.	gradient. method, tangent method.		

	<ul> <li>13) Basics of probability - probability space, conditional probability, random variables.</li> <li>14) Discrete random variables - binomial and Poisson distribution.</li> <li>15) Continuous random variables - normal distribution.</li> </ul>					
2.6. Type of instruction	Image: Section of the section of th			2.7. Comments:		
2.8. Student responsibilities	Conditions for signature: attendance	e to lecture	s and exercises m	nin. 70%.		
2.9. Screening of student's work (specify	Class attendance	1	Research		Practical training	
the proportion of ECTS credits for	Experimental work		Report			
each activity so that the total	Essay		Seminar essay		(Otherdescribe)	
number of CTS credits is equal to	Tests		Oral exam	1	(Other-describe)	
the credit value of the course)):	Written exam		Project	2	(Other-describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Project: 60% Oral exam: 40%					
	Title			Number of copies at the library	Availability via ot	her media
0.44 Demuired literature (considering to	Ivan Slapničar, Matematika 2, Fakultet elektrotehnike, strojarstva i brodogradnje u Splitu, Split, 2008.			0	http://lavica.fesb.hr/mat2/P	DF/predavanja.pdf
the library and via other media)	Ivan Slapničar, Josipa Barić i Marina Ninčević, Matematika 1 – zbirka zadataka, Fakultet elektrotehnike, strojarstva i brodogradnje u Splitu, Split, 2012.			20		
	B. P. Demidovič: Zadaci i riješeni primjeri iz više matematike s primjenom na tehničke nauke, Tehnička knjiga, Zagreb, 1986.			7		
2.12. Optional literature (at the time of the submission of the study programme proposal)	A. Jazbec, Osnove statistike, Sveuč F. M. Brückler, R. Pezer, Inženjersk	čilišna tiska a matemat	ra, Zagreb, 2007. ika – skripta (onlir	ne).		
2.13. Methods of monitoring quality that	Survey on the level of faculty and University. Analyses provided by quality assurance system of the institution. Analyses provided by quality assurance system and authorized office of the University.					

Ordinal number	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning outcomes
1	Collect, analyse and interprete data by statistical methods.	Project task, oral exam
2	Explain the notion, properties and graphical interpretation of functions of several real variables.	Project task, oral exam
3	Use directional derivatives of functions of several variables to analyse their local behaviour.	Oral exam
4	Use and interprete interpolation and approximation of given data with a mathematical model.	Project task, oral exam
5	Use basic methods of numerical integration to solve simple differential equations.	Project task
6	Explain and discuss iterations in the process of solving equations.	Project task, oral exam
7	Explain and use methods of chosen computer tool to solve nonlinear equations.	Project task

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE: 169735								
1.1. Course teacher	Full Prof. Damir Hršak, PhD		1.6. Year of study		1			
1.2. Name of the course	HYDROMETALLURGY		1.7. Credit value (ECTS)		3			
1.3. Associate teachers	-		1.8. Type of instruction (nur L+S+E+e-learning)	mber of hours	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate		1.9. Expected enrolment in	the course	30			
1.5. Status of the course	compulsory		1.10. Level of use of e-lear level), percentage of in course on line (20% ma	ning (1, 2, 3 struction in the aximum)	1., 5%			
2. COURSE DESCRIPTION								
2.1. Course objectives	Course objective is that stud materials suitable for hydror	dents learn netallurgica	the theoretical basis of hydro al leaching. Achieving leaders	ometallurgical pro ship skills for hyd	cesses and systematization rometallurgical processes.	of mineral raw		
2.2. Enrolment requirements and required entry competences for the course	Knowledge of basic techniq	ues of cher	nical analysis.					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Plan and manage metallurgical processes. Use the acquired theoretical knowledge in engineering practice.							
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Explain the theoretical basis Categorize leaching of mine Appraise hydrometallurgical Use purification and concen	of hydrom ral raw ma technologi tration of le	etallurgy. terials. es. eaching solutions.					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Theory of hydrometallurgy (4), Hydrometallurgical technologies (5), First colloquium (1), Hydrometallurgical leaching of mineral raw materials (9), Second colloquium (1), Purification and concentration of leaching solutions (5), Precipitation of metals and metal compounds (4), Third colloquium (1). EXERCISES (15): Stoichiometry in hydrometallurgy (6), Reactivity of hydrometallurgical raw materials (3), Use of hydrometallurgical leaching sources (6).							
	A lectures	<u> </u>	independent study	2.	.7. Comments:			
2.6. Type of instruction	exercises     online in entirety     mixed e-learning     field work	mops I multimedia and the internet I laboratory I work with the mentor (other)						
2.8. Student responsibilities	Attendence a minimum of 70	0% lectures	s. Successfully finished labora	atory exercises.				
2.9. Screening of student's work	Class attendance		Research	P	ractical training			
(specify the proportion of ECTS	Experimental work	1	Report					
credits for each activity so that the	Essay		Seminar essay		(Otherdescribe)			
total number of CTS credits is equal	Tests	1	Oral exam	1	(Other-describe)			

to the credit value of the course)):	Written exam Project (Other—describe)						
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	t Evaluation of student activity in class and laboratory, evaluation of laboratory exercises, exam through continuous monitoring (three colloquiums and oral exam) or final exam.						
2.11. Required literature (available at		Title		Number of copies at the library	Availability via other media		
the library and via other media)	D. Hršak, Hydrometallurgy,	10					
2.12. Optional literature (at the time of the submission of the study programme proposal)	F. Habashi, Metals from Ore	es, Metallurgie Extractive Quebec, Sa	inte-Foy, 2003.				
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Analysis of colloquiums and Survey on the level of facult Analyses provided by qualit Analyses provided by qualit	l exams. y and University. y assurance system of the institution. y assurance system and authorized o	ffice of the University				

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Explain the theoretical basis of hydrometallurgy.	1st colloquium, oral exam
2	Categorize leaching of mineral raw materials.	2nd colloquium, laboratory exercises, oral exam
3	Appraise hydrometallurgical technologies.	1st colloquium, oral exam
4	Use purification and concentration of leaching solutions.	3rd colloquium, laboratory exercises, oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE: 169736						
1.1. Course teacher	Assoc.Prof. Anita Begić Hadžipašić, PhD	1.6. Year of study	1			
1.2. Name of the course	METAL CORROSION AND PROTECTION	1.7. Credit value (ECTS)	5			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+30+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30			
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION						
2.1. Course objectives	Introducing to the basic principles of chemic Introducing students with mechanisms of di Introducing to the principles of corrosion pro	cal and electrochemical degradation of struc ifferent corrosion types of engineering metal otection.	ctural metals. lic materials in real operating conditions.			
2.2. Enrolment requirements and required entry competences for the course	-					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Suggest new and improved technical and technological solutions. Combine social, ethical and business principles and norms in the professional field.					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Distinguish the corrosion behaviour of metallic materials in exploitation. Calculate corrosion rate on the basis of data obtained by measuring in the laboratory and in the practice. Evaluate and compare the individual forms of corrosion defects of metallic materials. Investigate the resistance of structural steels to hydrogen embrittlement on the basis of determination of hydrogen physical- chemical parameters. Recognize the corrosion conditions and recommend an optimal engineering material for real service conditions.					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ul> <li>LECTURES (30):</li> <li>1. Introduction. 2h</li> <li>2. Temperature dependence of corrosion rate. Corrosion in gasses. 2h</li> <li>3. Electrochemical corrosion. 2h</li> <li>4. Kinetics of electrochemical processes. The polarization curves. Depolarization processes. The types of anode and cathode polarization. 2h</li> <li>5. Kinetics of electrochemical processes. Hydrogen depolarization in acid medium. Hydrogen depolarization in alkaline/neutral medium. Oxygen depolarization. 1h</li> <li>6. The external and internal factors that determine the rate of electrochemical corrosion. 1h</li> <li>7. 1<sup>st</sup> colloquium. 1h</li> <li>8. The passivity of metals. The passivity theories. The mechanism of the formation and maintenance of the passive layer. 2b</li> </ul>					

	9. The corrosion inhibitors: anode, cathode, organic, inorganic and VCI. 2h									
	10. The corrosion stimulators: anode and cathode stimulators. 2h									
	11. Electrochemical p	11. Electrochemical protection of metals. The anode and cathode electrochemical protection of metals. 2h								
	12. Electrochemical protection of metals with non-metallic coatings. 1h									
	13. 2 <sup>nd</sup> colloquium. 1h	13. 2 <sup><sup>ru</sup> colloquium. 1h</sup>								
	14. The types of corro	14. The types of corrosion testings and application of appropriate standards. 2h								
	15. Atmospheric corro	osion. The	controlling processes o	f atmos	spheric	corrosion of me	etals. 2h			
	16. Corrosion in the s	oil. The eff	ect of microorganisms of	on the	corrosic	on in the soil. Th	ne corrosior	h by stray currents. 1h		
	17. The protection coa	atings. 3h								
	18. 3 <sup>rd</sup> colloquium. 1h									
	LABORATORY EXERCIS	SES (30):								
	1. Chemical (hightem	perature, g	aseous) corrosion. 4h							
	2. Electrochemical co	rrosion and	d galvanic couples. 2h							
	3. The influence of im	purities in	zinc on the corrosion ra	ite in a	n electro	olyte solution. 3	3h			
	4. Stress corrosion cr	acking: coi	rosion of stressed meta	al in an	electro	lyte solution. 4	า			
	5. Hydrogen depolariz	zation and	hydrogen embrittlemen	t. 3h		-				
	6. Potentiodynamic po	olarization	of metal in solution. 2h							
	<ol><li>Cathode protection</li></ol>	of metals.	2h							
	8. Obtaining of nonme	etal coating	s by chemical method:	brown	ing of s	teel. 3h				
	<ol><li>Testing of galvanic</li></ol>	baths. 2h								
	10. Obtaining of meta	l coatings	by chemical method. 2h	า						
	11. Galvanization and	l metal coa	tings: nickel plating of s	steel. 3	h					
	⊠ lectures			,	2	.7. Comments	•			
	seminars and workshops		multimedia and the interr		not		-			
2.6 Type of instruction	🔀 exercises				liet					
	online in entirety		$\square$ work with the mentor							
	mixed e-learning									
	ield work									
2.8. Student responsibilities	Attendance min. 70 %, co	mpleted la	boratory exercises, sub	mitted	work di	ary of laborator	y exercises			
2.9. Screening of student's work	Class attendance		Research			Practical train	ning			
(specify the proportion of ECTS	Experimental work	1.5	Report	0.5						
credits for each activity so that the	Essay		Seminar essay			(Other-	-describe)			
total number of CTS credits is equal	Tests	3.0	Oral exam			(Other-	-describe)			
to the credit value of the course)):	Written exam		Project			(Other-	-describe)			
2.10. Grading and evaluation of student	Passed three colloquiums	through co	ontinuous monitoring or	final e	xamina	tion by oral exa	ım.			
work over the course of instruction										
and at a final exam										
		Title			Numb	per of copies	۸vai	lability via other media		
2.11 Pequired literature (available at		The			at	the library	Avai	ability via other media		
the library and via other media)	E. Stupnišek-Lisac, Koroz	ija i zaštita	konstrukcijskih materija	ala,	1					
	Fakultet kemijskog inženje	<u>erstva i te</u> h	nologije, Zagreb, 2007.							
	I. Esih, Z. Dugi, Tehnologija zaštite od korozije I. Školska knjiga									

	Zagreb, 1990.					
	I. Esih, Z. Dugi, Tehnologija zaštite od korozije II, Sveučilište u	1				
	Zagrebu, Zagreb, 1992.	1				
	J. Malina, Vježbe iz korozije i zaštite metala, interna skripta,		Morlin custov za o učonio			
	Metalurški fakultet, Sisak, 2004.		Merlin Sustav za e-ucenje			
	F. Sebenji, L. Hakl, Korozija metala u teoriji i praksi, prijevod s	0				
	engleskog, Tehnička knjiga, Beograd, 1980.	0				
	S. Mladenović, Korozija materijala, Tehnološko-metalurški	1				
	fakultet, Beograd, 1978.	4				
	B. Jarić, A. Rešetić, Korozija i katodna zaštita, Korexpress,	1				
	Zagreb, 2003.	1				
2.12 Optional literature (at the time of	S. Martinez, I. Štern, Korozija i zaštita-eksperimentalne metode, HINUS, Zagreb, 1999.					
the submission of the study	M. Gojić, Površinska obradba materijala, Metalurški fakultet Sveuč	čilišta u Zagrebu, Denona d.o.o., 2010.				
	T. Filetin, K. Grilec, Postupci modificiranja i prevlačenja površina, Hrvatsko društvo za materijale i tribologiju, Zagreb, 2004.					
programme proposal)	I. Esih, Osnove površinske zaštite, Sveučilište u Zagrebu, Fakultet strojarstva i brodogradnje, Zagreb, 2003.					
2.13. Methods of monitoring quality that	The survey at the level of faculty and University.					
ensure acquisition of exit	Analysis predicted by the system of institution quality assurance.					
competences	Analysis predicted by the system of quality assurance and author	ized bureau of universi	ty.			

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Distinguish the corrosion behaviour of metallic materials in exploitation.	Oral exam
2	Calculate corrosion rate on the basis of data obtained by measuring in the laboratory and in the	Laboratory exercises
	practice.	
3	Evaluate and compare the individual forms of corrosion defects of metallic materials.	Laboratory exercises, oral exam
4	Investigate the resistance of structural steels to hydrogen embrittlement on the basis of	Laboratory exercises, oral exam
	determination of hydrogen physical-chemical parameters.	
5	Recognize the corrosion conditions and recommend an optimal engineering material for real	Laboratory exercises, oral exam
	service conditions.	
6	Recommend the measures of corrosion protection.	Oral exam

1. COURSE DECRIPTION - GENERAL	INFORMATION		ISVU CODE: 169737			
1.1. Course teacher	Assoc.Prof. Ljerka Slokar, PhD Assoc.Prof. Tamara Holjevac Grgurić, PhD	1.6. Year of study	1			
1.2. Name of the course	MATERIALS CHARACTERIZATION	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+30+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30			
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION						
2.1. Course objectives	Course objectives Acquisition of knowledge and acquirements for characterization of different kinds of materials. Introducing with experimental techniques for investigation of content and structure of materials, their implementation and interpretation. Overwhelm with experimental techniques for determination of thermal, mechanical, rheological and tribological properties. Acquirement of competences for selection of appropriate experimental technique, defining conditions for material preparatio					
2.2. Enrolment requirements and required entry competences for the course	-					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Plan and manage the competences of analysis and synthesis. Suggest appropriate methods for material quality analysis.					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Define and classify the experimental techniques. Determine different groups of materials. Analyze the chemical composition of different groups of materials. Prepare samples for specific experimental technique. Apply techniques of structure testing and analyze the results obtained from the morphology of different materials. Select the appropriate technique of analysis of specific material properties and interpret measurement results. Autonomously organize and manage the measurement under optimal conditions and interpretate the results.					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction to material characterization. Classification of measuring techniques. Classification of materials. (2) Analysis of chemical content and structure by Fourier Transform Infrared Spectroscopy (FTIR); sample preparation, methodology and interpretation. (3) Spectrometry: quantitative and qualitative analysis. Analysis of chemical composition by atomic absorption spectrometer (AAS), etc.; sample preparation, methodology, interpretation. (3) Nuclear magnetic resonance; sample preparation, methodology and interpretation. (2) Structure analysis by X-Ray Diffraction (XRD); sample preparation, methodology and interpretation. (3)					

	Microstructure analysis by Thermal analysis of mater	Microstructure analysis by electron microscopy (SEM) and optical microscopy (OM). Interpretation of results. (3) Thermal analysis of materials. Differential scanning calorimetry (DSC). Differential thermal analysis (DTA). Sample preparation,							
	methodology and interpret	tation. (3)		-		-	• • •		
	Thermogravimetry (TG). S	Simultaneou	is technique	s of the	rmal analysis	s. (3)			
	Dilatometry (DIL). Thermo	mechanical	l analysis (T	MA). (2	)	х.			
	Mechanical and dynamic-	echanical and dynamic-mechanical analysis. Mechanical testing (3)							
	Investigation of tribulogica	/estigation of tribological properties. Equipment for surface scratching (micro- and nano-scratch tester). Determination of							
	I ABORATORY EXERCIS	11 αθρίτι απα Ες (30)· Δr	nalveis of ch	ADIASION	n. (3)	by atomic absorption spectrometer (4)	) Analysis of the		
	microstructure of different	types of ma	aterial by or	stical mi	croscopy and	d scanning electron microscopy and ir	nterpretation of results		
	(7). Structure analysis by	X-Ray Diffra	action, resul	Its interr	pretation (4).	Analysis of the chemical composition	of bentonite,		
	polymers and other mater	ials by FTIF	R method (3	). Analy	sis of metal a	allovs by differential scanning calorime	etry: determination of		
	transformation temperatur	e, the deter	rmination of	the entl	nalpy (3). An	alysis of polymers, ceramics and com	posites by differential		
	scanning calorimetry: glas	s transition	temperatur	e, meltir	ng point, entl	halpy (3). Determination of weight cha	inge through oxidation		
	of metal by thermogravime	etry (3). Mor	nitoring of c	hanges	of module st	tiffness, loss modulus and tangent of t	the angle of loss by		
	dynamic -mechanical anal	lysis: metals	s, polymers	, compo	sites (3).				
			] independe	ent study	/	2.7. Comments:			
		multimedia and the inte		e internet					
2.6. Type of instruction	aboratory		7						
	mixed e-learning work with		the men	itor					
	ield work		_ (oth	ier)					
2.8. Student responsibilities	Attendance to lectures and	d exercises	min. 70 %.	Lab rep	orts.	1			
2.9. Screening of student's work	Class attendance	0.4	Research			Practical training			
(specify the proportion of ECTS	Experimental work	0.3	Report		0.3				
credits for each activity so that the	Essay	ļ	Seminar e	essay		(Otherdescribe)			
total number of CTS credits is equal	Tests	1.0	Oral exam	۱	1.0	(Other-describe)			
to the credit value of the course)):	Written exam	1.0	Project			(Other-describe)			
2.10. Grading and evaluation of student	The final grade is determined by score of two colloquiums and assessments of written and oral exams respectively.								
work over the course of instruction									
and at a final exam				NL	wher of				
	Title			nu	mper or	Availability via othe	r media		
				l	ibrarv		i illeula		
	J. B. Sibilia, Materials Cha	aracterizatio	on and		Jor an <b>J</b>	https://www.amazon.com/Guide-Ma	terials-		
	Chemical Analysis, Wiley-	VCH, 1996		l		Characterization-Chemical-Analysis	s/dp/0471186333		
2.11. Required literature (available at	, , , , , , , , , , , , , , , , , , ,			l					
the library and via other media)	P. J. Heines, Thermal Met	thods of Ana	alysis,	]	1				
	Principles and Aplication,	Blackie Aca	ademic &	l					
	Professional, 1995.			ļ					
	T. H. Grgurić, Eksperir	mentalne t	lehnike u	1		https://www.simet.unizg.hr/nastava/	predavanja/diplomski-		
	termodinamici materijala,	ti fakultet,	1		sveucilisni-studij-metalurgija/1-godir	na-diplomskog-			

	Sisak		studija/eksperimentalne-tehnike-u-termodinamici- materijala/view			
2.12. Optional literature (at the time of	E. F. Kaufmann, Characterization of Materials, Wiley-Interscience, 2003.					
the submission of the study	R. Divakar, P. J. Blau, Wear Testing of Advanced Materials, ASTM Committee G-2 on Erosion and Wear, 1992.					
programme proposal)	J. Goldstein et al., Scanning Electron Microscopy and X-Ray Microanalysis, Third Edition, Springer, USA, 2003.					
2.13. Methods of monitoring quality that	Anonymous survey on the level of the Faculty and University.					
ensure acquisition of exit	Analysis provided by system of quality assurance institutions.					
competences	Analyses provided by quality assurance system and authorized offices of the University.					

Ordinal number	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning outcomes
1	Define and classify the experimental techniques.	1st colloquium, laboratory exercises, written and oral
2	Determine different groupe of meterials	examined and collective to be retering exercises, written and
2	Determine different groups of materials.	oral exam
3	Analyze the chemical composition of different groups of materials.	1st and 2nd colloquium, independent task, laboratory
		exercises, written and oral exam
4	Prepare samples for specific experimental technique.	1st and 2nd colloquium, independent task, laboratory
		exercises, written and oral exam
5	Apply techniques of structure testing and analyze the results obtained from the morphology of	1st and 2nd colloquium, independent task, laboratory
	different materials.	exercises, written and oral exam
6	Select the appropriate technique of analysis of specific material properties and interpret	1st and 2nd colloquium, laboratory exercises, written and
	measurement results.	oral exam
7	Autonomously organize and manage the measurement under optimal conditions and	1st and 2nd colloquium, independent task, laboratory
	interpretate the results.	exercises, written and oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE: 169738					
1.1. Course teacher	Full Prof. Ladislav Lazić, PhD Assoc.Prof. Zdenka Zovko Brodarac, PhD	1.6. Year of study	1		
1.2. Name of the course	AUTOMATION AND COMPUTER CONTROL OF PRODUCTION PROCESSES	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+15+15+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30		
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	<ol> <li>Acquire knowledge of process automation, process control and computerization process.</li> <li>Acquire knowledge about the degree of automation and methods of application of process computers.</li> <li>Acquire knowledge on levels of process control and automation functions.</li> <li>Acquiring knowledge about the structures of automation.</li> <li>Acquire knowledge about the structures of automation.</li> </ol>				
2.2. Enrolment requirements and required entry competences for the course					
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Plan and manage metallurgical processes.</li> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Suggest new and improved technical and technological solutions.</li> <li>Analyse the development and application of new technologies.</li> <li>Suggest solutions for the optimization of metallurgical processes.</li> <li>Combine the skills necessary for lifelong learning, including continued professional training.</li> </ol>				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Use repeatable and predictable control of the process.</li> <li>Suggest and implement overall better control of the process.</li> <li>Implement the improvement of process efficiency.</li> <li>By controlling the process in order to achieve a higher quality of products.</li> <li>By controlling the process easier to detect and eliminate errors.</li> </ol>				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): 1. Production systems, industrial plant and its types. The task of managing the process and stratification of management tasks. Man-machine interface. (2) 2. Computer application for process control. Informatization and automation of the production system. The basic structure of the system for automatic process control. Examples from practice. (2) 3. The system for measurement and display of process variables. The system of automatic control. The advantages of digital regulators. (4) 4. PLC properties and their programming. (4)				

	<ol> <li>Connecting a process computer with the process. Process (operating) unit - central unit system for automatic process control. Structural units for simple and complex systems. (4)</li> <li>Central, decentral, hierarchical and distributed structures. Control unit - subsystem for communication operator-production system. Equipment for the implementation of process and control unit. (4)</li> </ol>						
	1. colloquium						
	<ol> <li>Communication systems in industry. Transfer technologies / standards of general purpose underlying some industrial communication standards. (2)</li> <li>Fieldbus communication technologies; ASI, PROFIBUS, CAN, BITBUS. (2)</li> <li>PLC specialized networks; Melsecnet, SINEC, DataHighway. Software in automation systems (SCADA). (2)</li> <li>Programming tools. PC as a monitoring unit. Integrating office packages / applications in system automation. (2)</li> <li>Designing and maintaining automation systems. (2)</li> </ol>						
	LABORATORY EXERCISES (15): Examples of computer application for process control and automatic control. SEMINAR (15): The understanding of the material exposed in lectures is facilitated by solving the given projects. The projects						
	A lectures					Comments:	
2.6. Type of instruction	<ul> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>	<ul> <li>Independent study</li> <li>multimedia and the internet</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>					
2.8. Student responsibilities	Attendance to Lectures and E	Attendance to Lectures and Exercises > 70 %					
2.9 Screening of student's work (specify	Class attendance 0.2 Resea		irch		Practical training		
the proportion of ECTS credits for	Experimental work		Repor	t			
each activity so that the total	Essay		Semin	aressav		(Otherdescribe)	
number of CTS credits is equal to	Tests	0.8	Oral e	xam	1.2	(Other—describe)	
the credit value of the course)):	Written exam	0.8	Proiec	:t	1.0	(Other—describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Projects – 30% Class attendance – 5% Written exam – 30% Oral exam – 35%						
	Title		Number of copies at the library		Availability via other media		
2.11. Required literature (available at the library and via other media)	N. Perić, Automatizacija postrojenja i procesa - predavanja, Zavodska skripta, FER, Zagreb, 2000.		5				
	N. Perić, I. Petrović, M. Vašak, Procesna automatizacija, Sveučilište u Zagrebu, Fakultet elektrotehnike i računarstva. Zagreb. 2013.				https://www.fer.unizg.hr/_download/repository/Skripta _Proaut_veljaca_2013.pdf		

	I. Gašparac, M. Vražić, Projektiranje i automatizacija industrijskih postrojenja, Sveučilište u Zagrebu, Fakultet elektrotehnike i računarstva, Zagreb, 2012.		https://www.fer.unizg.hr/_download/repository/skripta PIAIP_ver1.pdf
	I. Erceg, T. Šimić, Automatizacija industrijskih postrojenja, Uvod u PLC-ove.		http://www.ieee.hr/_download/repository/AIP _PLC_prezentacija.pdf
2.12. Optional literature (at the time of the submission of the study programme proposal)	A. J. Crispin, Programmable Logic Controllers and G. Smiljanić, Računala i procesi, Školska knjiga, Z	their Engineering App agreb, 1991.	lications, McGraw-Hill Publishing Company, 1997.
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Survey at the level of faculty and University. Analyses provided in the system of quality assurance of the institution. Analyses provided in the system of quality assurance and an authorized office of the University.		

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of		
number		learning outcomes		
1	Use repeatable and predictable control of the process.	1st colloquium, oral exam		
2	Suggest and implement overall better control of the process.	1st colloquium, project, oral exam		
3	Implement the improvement of process efficiency.	2nd colloquium, oral exam		
4	By controlling the process in order to achieve a higher quality of products.	2nd colloquium, oral exam		
5	By controlling the process easier to detect and eliminate errors.	2nd colloquium, oral exam		

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE: 169739							
1.1. Course teacher	Assoc.Prof. Zdenka Zovko Brodarac, PhD	1.6. Year of study	1				
1.2. Name of the course	THEORY OF METALS SOLIDIFICATION	1.7. Credit value (ECTS)	4				
1.3. Associate teachers	-	<ol> <li>Type of instruction (number of hours L+S+E+e-learning)</li> </ol>	30+15+0+0				
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30				
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%				
2. COURSE DESCRIPTION							
2.1. Course objectives	Understanding of the solidification mechani Definition of thermodynamics and solidification Knowledge of solidification model, crystal g Theory of Al-Si and Fe-C eutectic solidification	ism and development of the primary structure. tion phenomena. rowth and melt treatment. tion.					
2.2. Enrolment requirements and required entry competences for the course	-	-					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Suggest new and improved technical and technological solutions. Plan the production and casting processes of ferrous and non-ferrous metals.						
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Evaluate the thermodynamic parameters of materials in the process of solidification. Present the importance of heterogeneous nucleation and the related practice of melt treatment. Present the importance of eutectic system and the basic mechanisms of solidification and growth of Al-Si eutectic. Present the importance of eutectic system and the basic mechanisms of solidification and growth of Fe-C eutectic.						
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction to the subject curriculum, method of mastering the material. (1) The requirements of the market and trends in production of casting and importance of solidification related to the properties of casting - primary microstructure. (2) Interpretation of the microstructure of the casting alloy. (1) Interpretation of the microstructure of the casting alloys. Correlation of technology, primary microstructure and properties of casting (2) The mechanisms of heat transfer at the interface melt-mold. The transformation of physical condition. Characteristic transformations during solidification of metals and alloys. (2) The solidification of single-phase alloys. Thermodynamics of phase transformation of liquid-solid. The driving force for solidification, simple thermal analyzes. Recallescence. The kinetics of phase transitions. (2) The theory of nucleation. The homogeneous nucleation rate. Terms of homogeneous nucleation. Heterogeneous nucleation rate. (2) The characteristics of the grain refiners. Inoculation practice. Nucleation and inoculation of commercial casting alloys. The mechanisms of nucleation, grain refinement. The structure of crystals boundary surface, crystal growth, microstructure development. Graphite growth from cast iron melt, kinetics and growth forms of graphite. (4)						

	Solidification alloys. The distribution of the solute elements. The solidification in steady state. The growth of dendrites. Constitutional undercooling. (2) Segregation in ingots and castings. Macro and microsegregation. Gravitational segregation. Consequences of microsegregation - occurrence of unexpected phase. Development of the primary structure of the castings. The emergence of the structural zones. The transition from columnar to equiaxed crystallization. (2) Eutectic solidification, eutectic zones. Competitive growth of dendrites and eutectic. Solidification of alloys outside the eutectic composition. Modification of the eutectic. Eutectic growth in Fe-C and Al-Si alloys. The effect of inoculation and modification on the properties of eutectics. (4) Peritectic solidification. Mechanisms of peritectic solidification. Solidification, the development of primary microstructure and technological properties: shrinkage, blistering, castability. The use of simple and simultaneous thermal analysis in interpretation of solidification. Solidification interval, the sequence of phase transitions. (4)							
	SEMINAR (15): The study of relevant scier Presentation of knowledge	tific and teo from scien	chnical literature (10). tific and professional articles	s (5)				
	$\boxtimes$ lectures	☐ lectures ☐ independent study 2.7. Comments:						
2.6. Type of instruction	<ul> <li>Seminars and worksnops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>		ernet					
2.8. Student responsibilities	Attending the classes >70%. Seminar essay and presentation.							
2.9. Screening of student's work (specify	Class attendance 1 Research Prac				ractical tra	aining		
the proportion of ECTS credits for	Experimental work		Report	1				
each activity so that the total number	Essay		Seminar essay		(Oth	(Otherdescribe)		
of CTS credits is equal to the credit	Tests	2	Oral exam		(Oth	(Other—describe)		
value of the course)):	Written exam		Project		(Oth	(Other-describe)		
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	<ul> <li>Examination through continuous monitoring:</li> <li>1. colloquium after the unit Graphite growth from cast iron melt, kinetics and growth forms of graphite.</li> <li>2. colloquium after the unit Solidification interval, the sequence of phase transitions.</li> <li>Examination through final exam: written and oral.</li> <li>Make a project task.</li> </ul>							
			Title		N CO	umber of bies at the library	Av: of	ailability via ther media
2.11. Required literature (available at the library and via other media)	Metals Handbook, Volume 15, CASTING, ASM International, Metals Park, Ohio, 1988. Dopunska literatura (1 do 5) 1 2				1			
	W. Kurz, D. J. Fisher, Fundamentals of solidification, Trans Tech Publications LTD, Aedermannsdorf, 1986.				),	1		
	D. A. Porter, K. E. Easterling, Phase transformations in metals and allovs,					1		

	Chapman & Hall, London, 1992.			
	D. M. Stefanescu, Science and engineering of casting solidification, Kluwer Academic /Plenum Publishers, New York, 2002.	1		
2.12. Optional literature (at the time of	T. Nishizawa, Thermodynamics of microstructures. ASM International, Materials Park, Ohio, 2008.			
the submission of the study	J. Campbell, Castings, Butterworth Heinemann, Oxford, 1991.			
programme proposal)				
2.13. Methods of monitoring quality that	Survey at the Faculty and University level.			
ensure acquisition of exit	Analysis provided the quality assurance system of the institution.			
competences	Analysis provided the quality assurance system and authorized Office of the Universit	ty		

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of				
numper		learning outcomes				
1	Evaluate the thermodynamic parameters of materials in the process of solidification.	1st colloquium, written and oral exam				
2	Present the importance of heterogeneous nucleation and the related practice of melt treatment.	1st colloquium, written and oral exam				
3	Present the importance of eutectic system and the basic mechanisms of solidification and	2nd colloquium, seminar paper, written and oral exam				
	growth of Al-Si eutectic.					
4	Present the importance of eutectic system and the basic mechanisms of solidification and	2nd colloquium, seminar paper, written and oral exam				
	growth of Fe-C eutectic.					
1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE: 169740						
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1.1. Course teacher	Full Prof. Ladislav Lazić, PhD	1.6. Year of study	1			
1.2. Name of the course	HEATING TECHNOLOGY OF INDUSTRIAL FURNACES	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	Ivan Jandrlić, PhD	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30			
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION	·		•			
2.1. Course objectives	<ol> <li>Enable students to calculate energy balance, determine the energy efficiency and evaluate the rational use of energy.</li> <li>Enable students to determine the overall heat transfer coefficient in the furnace workspace.</li> <li>Enable students to distinguish individual cases of material heating and determine the calculating method of heating (cooling) in a particular case.</li> <li>Enable students to regulate the temperature of combustion gases, stop the furnace operation and perform a preliminary regions of the furnace.</li> </ol>					
2.2. Enrolment requirements and required entry competences for the course	Passed the exams of course Industrial Furnaces.					
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Suggest new and improved technical and technological solutions.</li> <li>Analyse the development and application of new technologies.</li> <li>Analyse the production processes by applying thermodynamic laws.</li> <li>Plan and manage metallurgical processes.</li> <li>Design professional elaborates and professional projects in metallurgy.</li> <li>Design and apply the modelling of metallurgical and other processes.</li> <li>Analyse the material and thermal balance of metallurgical processes.</li> <li>Suggest solutions for the optimization of metallurgical processes.</li> </ol>					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Calculate the heat transfer inside the furnace to heated material.</li> <li>Suggest conditions of heating metal materials, predict the amount of oxidation loss of material in these conditions, and choose the composition of the necessary protective or reaction gas.</li> <li>Calculate thermal stress of the heated material and choose the allowed rate of heating metal material.</li> <li>Calculate the required dimensions of the heating surfaces of the heat exchanger and boiler utilizer.</li> <li>Design the optimal thickness of walls furnaces.</li> <li>Suggest measures for reducing pollutant emissions into the environment.</li> </ol>					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ul> <li>LECTURES (30):</li> <li>1. Thermo-technical processes in industrial furnaces. Requirements to be met by the furnace. Melting, heating and heat treatment furnaces. (2)</li> <li>2. Processes of heating and cooling. Furnaces heated with fossil fuels, electric energy, and solar energy. (2)</li> </ul>					

	<ol> <li>Energy balance and energy</li> <li>Heat transfer by convection</li> </ol>	<ol> <li>Energy balance and energy efficiency of the furnace. Oxidation of the heated material. Protective and reaction gases. (2)</li> <li>Heat transfer by convection and radiation, radiation of flame, overall coefficient of heat transfer in the workspace of</li> </ol>						
	furnace. (2)							
	5. The heating of metal materials. Thin and massive bodies. Numerical calculation methods. (2)							
	6. Rules relating to the mas	6. Rules relating to the massive bodies. Graphical method for plate-shaped bodies. Other calculation methods for heating						
	7 Heating (cooling) of mas	sive hodies	at: a-const dT/dt-const Tw	-const(2)				
	7. Treating (cooling) of mas		1st colloquium	0 –const. (2)				
	8. Thermal stresses in heater	d body. Ter	sile and compressive stresses	. (2)				
	9. Classification of steel into	groups acco	ording to heating conditions. (2	)				
	10. Heat exchangers. Classif	ication and	types. Calculation of heating s	urface. (4)				
	11. Utilizer boller. Project cal	culation. (2)	) Indication of composition and amo	ount of combuct	on gooog. Clooping o	fwooto	(2)	
	13 Preheating of combustion	n air Efficie	ncy of combustion (2)		on gases. Cleaning o	i wasie	e yases. (2)	
	14. Assessment of furnace c	ondition. Pr	eliminary review of furnace. Re	pair and overha	ul of furnace. (2)			
			2nd colloquium	I				
							<b>T</b> L .	
	EXERCISES (15): The understanding of the material exposed in lectures is facilitated by solving the given projects. The							
	A lectures							
	seminars and workshops		independent study					
2.6 Type of instruction	exercises							
	online in entirety		$\boxtimes$ work with the mentor					
	field work		(other)					
2.8. Student responsibilities	Attendance to Lectures and I	Exercises >	70 %.					
2.9 Screening of student's work (specify	Class attendance	0.4	Research	F	Practical training			
the proportion of ECTS credits for	Experimental work		Report					
each activity so that the total number	Essay		Seminar essay		(Otherdescribe)	)		
of CTS credits is equal to the credit	Tests	1.0	Oral exam	1.6	(Other-describe	e)		
value of the course)):	Written exam		Project	1.0	(Other-describe	e)		
2.10. Grading and evaluation of student	Projects – 30%							
work over the course of instruction	Class attendance – 5%							
and at a final exam	Oral exam = 35%							
					Number of			
0.44 Demoired literature (available at the			Title		copies at the	Ava	llability via	
Library and via other media)					library	U		
library and via other modia;	W. Lehnert, Toplotehničke os	snove za inc	dustrijske peći, Metalurški faku	ltet, Sisak, 2001	. 3			
				· · · ·				
2.12. Optional literature (at the time of	P. Mullinger, B. Jenkins, Indu	Jstrial and p	process furnaces: principles, de	esign and operat	Ion, Elsevier, Oxford,	2008.		
	J. ward, R Collins, Industrial furnace technology, Centro de energia e technologia, Rio Tinto, 2003.							

programme proposal)	
2.13. Methods of monitoring quality that	Survey at the level of faculty and University.
ensure acquisition of exit	Analyses provided in the system of quality assurance of the institution.
competences	Analyses provided in the system of quality assurance and an authorized office of the University.

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Calculate the heat transfer inside the furnace to heated material.	1st colloquium, oral exam
2	Suggest conditions of heating metal materials, predict the amount of oxidation loss of material in	1st colloquium, oral exam
	these conditions, and choose the composition of the necessary protective or reaction gas.	
3	Calculate thermal stress of the heated material and choose the allowed rate of heating metal	2nd colloquium, oral exam
	material.	
4	Calculate the required dimensions of the heating surfaces of the heat exchanger and boiler	2nd colloquium, project, oral exam
	utilizer.	
5	Design the optimal thickness of walls furnaces.	2nd colloquium, oral exam
6	Suggest measures for reducing pollutant emissions into the environment.	2nd colloquium, oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE: 169741						
1.1. Course teacher	Full Prof. Ladislav Lazić, PhD Assist.Prof. Martina Lovrenić-Jugović, PhD	1.6. Year of study	1			
1.2. Name of the course	NUMERICAL MODELLING OF METALLURGICAL PROCESSES	4				
1.3. Associate teachers	-	30+0+15+0				
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30			
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION			-			
2.1. Course objectives	<ol> <li>Develop the ability to apply numerical methods in the analysis of metallurgical processes.</li> <li>Qualify students to formulate an independent computer programs based on the finite difference method</li> <li>Enabling students to use modern software packages based on finite difference or finite element methods of analysis some problem.</li> </ol>					
2.2. Enrolment requirements and required entry competences for the course	-					
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Design and apply the modelling of metallurgical and other processes.</li> <li>Plan and manage metallurgical processes.</li> <li>Suggest solutions for the optimization of metallurgical processes.</li> </ol>					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Apply appropriate modeling techniques in guiding, monitoring and optimization of metallurgical processes.</li> <li>Formulate numerical models for planning, process analysis, design and optimization of existing technologies in metallurgy.</li> <li>Use commercial software packages based on finite elements to determine the temperature distribution over the cross section of the body, to analyze the thermal stress and to determine the speed of heating and cooling systems in metallic materials.</li> <li>Analyze and choose the optimal mode of heating the material during metallurgical processes.</li> <li>Calculate the adiabatic temperature and equilibrium composition of the combustion products of well-known type of fuel, the ratio of fuel to air, temperature and pressure.</li> </ol>					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ul> <li>LECTURES (30):</li> <li>Basic division of models: physical modeling, mathematical modeling, Experimental plant (2).</li> <li>The types of mathematical models and their applicability for modeling metallurgical processes (4).</li> <li>Numerical methods: Finite difference, finite element method, finite volume method, the boundary element method (4).</li> <li>The application of finite difference modeling of steady and unsteady heat conduction (6).</li> <li>1st preliminary exam: includes the units 1-4</li> <li>The method of finite element method for modeling of steady and unsteady heat transfer analysis with thermal stress when heating or cooling a metal cartridge (6).</li> </ul>					

	<ol> <li>Combustion and thermochemistry: stoichiometry, Standard thermodynamic conditions, enthalpy of formation, absolute enthalpy, enthalpy of combustion and heat value, adiabatic flame temperature (4).</li> <li>Chemical equilibrium complex systems (2).</li> <li>A numerical model for calculation of adiabatic temperature and the equilibrium composition of the combustion products (2).</li> <li>2nd preliminary exam: includes the units 6-9</li> <li>EXERCISES (15):</li> <li>Application software package for thermal analysis of metallurgical processes.</li> <li>Application software package HPFLAME budget adiabatic temperature and the equilibrium composition of the combustion products, the known type of fuel, the ratio of fuel to air, temperature and pressure, in solving practical problems in order to increase energy efficiency and reduce the emissions of industrial fuel combustion devices.</li> </ol>							
2.6. Type of instruction	<ul> <li>lectures</li> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> <li>independent study</li> <li>multimedia and the inter</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>			rnet	2.7. Con	nments:		
2.8. Student responsibilities	Conditions for signatur - attendance on Lectur - program task Conditions for taking: -	e: es and Exe	ercises > 70	%				
2.9 Screening of student's work (specify	Class attendance	0.4		Research		Practica	l training	
the proportion of FCTS credits for	Experimental work			Report		Program	n task	1.2
each activity so that the total number	Essav			Seminar essav		((	Otherdescribe)	
of CTS credits is equal to the credit	Tests	12		Oral exam	12	((	Other_describe)	
value of the course)):	Written exam			Project		(0	Other—describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Projects (program tasks) – 30% Attendance – 10% Written exam – 30% Oral exam – 30%							
	Title			Number of copies at the library		Availability via other media		
2.11. Required literature (available at the	L. Lazić, Numeričke metode u toplinskoj analizi, Sveučilište u Zagrebu, Sisak, 2007.			izi, Sveučilište	10			-
library and via other media)	V. Hari i drugi, Numerička analiza, PMF-MO, Zagreb, 2003.			Zagreb, 2003.	0		https://web.math.pmf.unizg.hr/~rogina/ 2001096/num_anal.pdf	
2.12. Optional literature (at the time of the submission of the study	S. P. Ketkar, Numerica T. Young, M. J. Mohler	al Thermal /	Analysis, AS oduction to	SME Press, New Numerical Metho	York, 1999. Ids and Mathl	ab Progra	mming for Engir	eers, Ohio University,

programme proposal)	2011.
	D. Vučina, Primjena računala u inženjerskoj analizi, FESB, Split, 2007.
	I. Ivanušić, Numerička matematika, ISBN: 953-197-526-4, Element, Zagreb, 2002.
	Y. Jaluria, Computer Methods for Engineering, Allyn and Bacon Inc., Massachusetts, 1988.
	Internal:
2.12 Mothods of monitoring quality that	Student survey input. Numerical analysis of tests and exams according to scoring task by task at the level of course.
2.13. Methods of monitoring quality that	External:
	Survey at the level of faculty and University.
competences	Analyses provided in the system of quality assurance of the institution.
	Analyses provided in the system of quality assurance and an authorized office of the University.

Ordinal number	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning outcomes
1	Apply appropriate modelling techniques in guiding, monitoring and optimization of metallurgical processes.	Oral exam
2	Formulate numerical models for planning, process analysis, design and optimization of existing technologies in metallurgy.	Oral exam
3	Use commercial software packages based on finite elements to determine the temperature distribution over the cross section of the body, to analyze the thermal stress and to determine the speed of heating and cooling systems in metallic materials.	Program task, 1st colloquium
4	Analyze and choose the optimal mode of heating the material during metallurgical processes.	Oral exam
5	Calculate the adiabatic temperature and equilibrium composition of the combustion products of well-known type of fuel, the ratio of fuel to air, temperature and pressure.	Program task, 2nd colloquium
6	Formulate and use of mathematical models for the development of new technologies in metallurgy.	Oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE: 169742						
1.1. Course teacher	Full Prof. Mirko Gojić, PhD Assoc.Prof. Stjepan Kožuh, PhD	1.6. Year of study	1			
1.2. Name of the course	TECHNIQUES OF JOINING AND CUTTING	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	30			
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION	-					
2.1. Course objectives	Knowing of students with the basic procedures of joining and cutting of metallic materials. Training for the selection of the appropriate process joining or cutting depending on the particular case of use. Training to avoid basic mistakes during joining and cutting and appropriate selection and recommendation of safety at work.					
2.2. Enrolment requirements and required entry competences for the course	-					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Suggest new and improved technical and technological solutions. Suggest solutions for the optimization of metallurgical processes.					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Define of the most important procedures for fusion welding. Diferentiate of weld joints types of metallic materials. Suggest of optimal parameters for arc welding processes. Differentiate of base material for welding. Compare of bonding metal material with soldering and bonding.					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Week 1: Overview of joining processes: welding, soldering and bonding (2). Week 2: Power sources for welding. Welding positions and types of weld joints (2). Weeks 3-6: Gas welding, manual electro arc, TIG-tungsten inert gas, MIG-metalk inert gas, MAG-metal active gas, under the protection of powder, electroresistance welding (spot, seam, high frequency), other fusion welding processes (electron beam, laser beam, under slag, aluminotermic welding), pressure welding (cold, diffusion, explosion, under friction, ultrasound) (8). Week 7: Weldability (2). Weeks 8 and 9: The behavior of iron alloys (steel, cast iron) and non-ferrous metals and alloys during welding (4). Week 10: Soldering: types of joint, solder, fluxes. (2) Week 11: The procedures of soldering (gas, electro-resistance, induction, hand soldering and others) (2). Week 12: Methods of cutting: gas, oxygen, electroarc, plasma, laser, electron beam, water etc. (2). Week 13: Bonding. The types of adhesives and joints (2).					

	Week 15: Mistakes during joining and cutting. Safety at work (2).							
	LABORATORY EXERCISES (15): Individual and group doing of certain procedures of joining and cutting (MAW, TIG, and MIG / MAG procedures). Surface preparation for soldering and bonding. Hand soldering. Bonding metal.							
			independent study	2.	7. Comments:			
2.6. Type of instruction	<ul> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>		<ul> <li>multimedia and the internet</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>					
2.8. Student responsibilities	Students must attend over 70	% of lecture	es and exercises.					
2.9. Screening of student's work	Class attendance	0.5	Research	Pi	actical training			
(specify the proportion of ECTS	Experimental work	1.0	Report					
credits for each activity so that the	Essay	Essay Seminar essay				(Otherdescribe)		
total number of CTS credits is equal	Tests	2.5	Oral exam		(Other-describe)			
to the credit value of the course)):	Written exam		Project		(Other-describe			
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	<ul> <li>evaluation of students activit</li> <li>evaluation of written examin</li> <li>evaluation of exercises activ</li> </ul>	ties in cours ation (two c rity.	se, colloquiums) through continuous	s monitoring or fir	nal examination (writh	ten and	d oral),	
2.11 Required literature (available at	TitleNumber of copies at the libraryAvailability via other media					ilability via ner media		
the library and via other media)	M. Gojić, Tehnike spajanja i razdvajanja materijala, Metalurški fakultet, Sisak,2003.			10				
,	Z. Lukačević: Zavarivanje, Strojarski fakultet Slavonski Brod, Slavonski Brod, 1998.				3			
2.12. Optional literature (at the time of the submission of the study programme proposal)	I. Juraga, K. Ljubić, M. Živčić, Pogreške u zavarenim spojevima, HDTZ, Zagreb, 1998. S. Kralj, Š. Andrić, Osnove zavarivačkih i srodnih postupaka, Fakultet strojarstva i brodogradnje, Zagreb, 1992.							
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Input and output student survey. Survey at the level of faculty and University. Analyses provided in the system of quality assurance of the institution. Analyses provided in the system of quality assurance and an authorized office of the University.							

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning
number		outcomes
1	Define of the most important procedures for fusion welding.	1st colloquium, written and oral exam
2	Diferentiate of weld joints types of metallic materials.	1st colloquium, written and oral exam
3	Suggest of optimal parameters for arc welding processes.	1st colloquium, written and oral exam, laboratory exercises
4	Differentiate of base material for welding.	2nd colloquium, written and oral exam, laboratory exercises
5	Compare of bonding metal material with soldering and bonding.	2nd colloquium, written and oral exam
6	Analyze of defects in welded joints.	2nd colloquium, written and oral exam

1. COURSE DECRIPTION - GENERAL	INFORMATION	ISVU CODE:				
1.1. Course teacher	Full Prof. Mirko Gojić,PhD Assoc.Prof. Stjepan Kožuh, PhD	1.6. Year of study	2			
1.2. Name of the course	HEAT TREATMENT AND SPECIAL STEELS	1.7. Credit value (ECTS)	5			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	45+15+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	20			
1.5. Status of the course	compulsory	<ul><li>1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)</li></ul>	1., 5%			
2. COURSE DESCRIPTION						
2.1. Course objectives	Getting insight into the properties and microstructure of metal after heat treatment. Studying of the relationship between heat treatment and properties of metals. Deepening knowledge in important areas in relation to the properties, microstructure and application of special steels. Understanding and identifying the basic types of steel and their properties.					
2.2. Enrolment requirements and required entry competences for the course	-					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Analyse the development and application of new technologies. Design the properties of metallic materials.					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Predict of individual transformation products during cooling of austenite, Relate of recrystallization processes and the heat treatment, Compare certain procedures of thermo-chemical treatment processes, Compare different types of steel to the characteristics and functions, Develop thermomechanical treatment with the goal of designing the properties of special steels, Predict the application of tool steels for specific loading conditions					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (45): Week 1 and 2: Definition and classification of steel, Influence of alloying elements, Phase transformations. Decomposition of austenite during cooling (pearlite, bainite, martensite) (6 hours). Week 3: The protective atmospheres in heat treatment. Property of hardenability (3 hours). Week 4: Chemical-thermal treatment procedures (3 hours). Week 5: Special methods of heat treatment. Thermomechanical treatment (3 hours). Week 6: Heat treatment of non-ferrous metals and alloys (3 hours). Week 7: Construction steels: properties and classification, Basic unalloyed and low alloyed construction steels, General construction steels (bearing structures, machine), Designation of steel (3 hours). Week 8: Fine-grained microalloved steels: Basic processes for increasing of steel strength. Precipitation hardening. Hardening					

	<ul> <li>Week 9: Microalloyed normalized fine grain steels, Improved fine grain steels, Construction steels with low content of pearlite or without pearlite, Steels for springs (3 hours).</li> <li>Week 10: Steels with improved cutting performance (steels for automated machines), Steels for nitriding, Techniques for nitriding of steels, Special constructional steels (classification) (3 hours).</li> <li>Week 11: Corrosion resistant (stainless) steels, ferrite, austenite, austenite-ferrite (duplex), and martensite stainles steels (3 hours).</li> <li>Week 12: Steels resistant to wear, Steels for use at elevated and high temperatures, Heat-resistant steels, Steel for use at low temperatures (3 hours).</li> <li>Week 13: High-strength steels: Low-alloyed and low-tempered steels, Cold working unalloyed and low-alloyed steels, Maraging steels (3 hours).</li> <li>Week 14: Fundamentals of tool steels: classification, Unalloyed (carbon) tool steels, Alloyed tool steels, Low alloyed tools steels for cold working (W-V, W-Cr-(Si)-V, Cr-steels, Mn-Cr-V, Mn-Cr-W], High-alloyed tool steels for cold working, Tool steels for hot working (3 hours).</li> <li>Week 15: High-speed steels, Sintered tool steels. Trends in the development of special steels (3 hours).</li> <li>EXERCISES (15): Testing of hardenability (Jominy test). Determination of the austenite grain size. Determination parameters of recrystallization. Annealing, quenching and tempering. Cementing and nitriding. Thermomechanical treatment.</li> <li>SEMINAR (15): The selection of topics and seminar work in writing form by a mentor system (10 hours). Preparation and presentation of the seminar and discussions in relation to the topic of the present paper (5 hours).</li> </ul>					nt of pearlite or ques for es steels (3 for use at low ) steels, Maraging oyed tools g, Tool steels	
2.6. Type of instruction	☑ lectures       □ independent study       2.7. Comments:         ☑ seminars and workshops       □ multimedia and the internet       □ laboratory         ☑ online in entirety       □ laboratory       □ work with the mentor         □ field work       □ (other)       □ (other)				5:		
2.8. Student responsibilities	Students must attend over 70% o	f lectures and a	are requi	red to compl	lete a seminar in writing form and orally present.		
2.9 Screening of student's work (specify	Class attendance	0.5	Resear	ch		Practical training	
the proportion of FCTS credits for	Experimental work		Report				
each activity so that the total	Essav		Semina	aressav	1.5	(Otherdescribe)	
number of CTS credits is equal to	Tests	3.0	Oral ex	am		(Other-describe)	
the credit value of the course)):	Written exam		Project			(Other-describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	- evaluation of students activities in course,     - evaluation of written examination (two colloquiums) through continuous monitoring or final examination (written and oral),     - evaluation of seminar paper and presentation.					and oral),	
2.11. Required literature (available at the library and via other media)	Required literature (available at e library and via other media)			Number of copies at the library		Availability via other med	ia
	M. Novosel i dr., Posebni čelici, S	trojarski fakulte	et	9			

	Slavonski Brod, Slavonski Brod, 1998.			
	S. Kožuh, Specijalni čelici-skripta, Sveučilište u Zagrebu Metalurški fakultet, Sisak, 2010.		https://www.simet.unizg.hr/nastava/predavanja/diplomski- sveucilisni-studij-metalurgija/2-godina-diplomskog- studija/specijalni-celici/view	
	J. Pirš, Toplinska obrada metala, Tehnički fakultet Rijeka, Rijeka, 1992.	3		
	ASM Handbook.pdf		Electronic form	
2.12. Optional literature (at the time of the submission of the study programme proposal)	C. R. Brooks, Principles of the Heat Treatment of Plain M. Novosel, F. Cajner, D. Krumes, Alatni materijali, Stro Grupa autora: Inženjerski priručnik 4, prvi svezak: Mate M. Gojić, Metalurgija čelika, Sveučilište u Zagrebu Meta R. A. Lula, Stainless Steel, ASM, Metals Park, Ohio, 19 Y. Lakhtin, Engineering Physical Metallurgy and Heat-T	Carbon and ijarski fakulte rijali, Školska lurški fakulte 85. reatment, Mi	Low Alloy Steels, ASM International, Materials Park, 1996. et, Slavonski Brod, 1996. a knjiga, Zagreb, 1998. et, Sisak, 2006. r Publishers, Moskva, 1990.	
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Y. Lakhtin, Engineering Physical Metallurgy and Heat-Treatment, Mir Publishers, Moskva, 1990. Input and output student survey. Survey at the level of faculty and University. Analyses provided in the system of quality assurance of the institution. Analyses provided in the system of quality assurance and an authorized office of the University.			

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of				
number		learning outcomes				
1	Predict of individual transformation products during cooling of austenite.	1st colloquium, written and oral exam, laboratory				
		exercises				
2	Relate of recrystallization processes and the heat treatment.	1st colloquium, written and oral exam, seminar paper				
3	Compare certain procedures of thermo-chemical treatment processes.	1st colloquium, written and oral exam				
4	Compare different types of steel to the characteristics and functions.	2nd colloquium, written and oral exam, seminar paper				
5	Develop thermomechanical treatment with the goal of designing the properties of special steels.	2nd colloquium, written and oral exam				
6	Predict the application of tool steels for specific loading conditions.	2nd colloquium, written and oral exam				

1. COURSE DECRIPTION – GENERA	AL INFORMATION	ISVU CODE:			
1.1. Course teacher	Assoc.Prof. Zoran Glavaš, PhD Assoc.Prof. Anita Štrkalj, PhD	1.6. Year of study	2		
1.2. Name of the course	SECONDARY METALLURGY AND CONTINUOUS CASTING	1.7. Credit value (ECTS)	5		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	45+0+15+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	20		
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	Capability for managing the steelmaking pro Understanding of treatment processes of ste Understanding of the continuous casting pro	cesses. eel in the ladle. cess of steel.			
2.2. Enrolment requirements and required entry competences for the course	-				
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Plan and manage metallurgical processes. Plan the production and casting processes of ferrous and pon-ferrous metals.				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Explain the chemical reactions that occur during treatment of steel in the ladle. Describe and explain the secondary steelmaking processes in ladle. Analyze the results of the applied secondary steelmaking process and propose measures to increase its efficiency. Describe and explain the process of continuous casting of steel. Explain the solidification of continuously cast products. Analyze the quality of continuously cast products and propose measures to improve their quality.				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (45): Definition, importance an basic characteristics of secondary steelms chemical equilibrium, Gibbs free energy, ac structure and physicochemical properties of during tapping, the influence of furnace slag preheating, mixing, steel reheating, contr thermodynamics of simple and complex ox liquid steel. (2); Degassing and decarburisa decarburisation, fluid flow and mixing duri vacuum, the production of steel with ultra- chromium steel melts. (4); Desulphurizatio	d classification of secondary steelmaking p aking processes. (2); Fundamentals of th ctivity, activity – composition relationship, th the slag, basicity and capacities of slag. (2) on the quality of steel, the change in tempe rol of the steel flow from bottom of the idation, kinetics of deoxidation reaction, kin ation of liquid steel in a vacuum: thermody ing degassing and decarburisation, kinetic low carbon content and stainless steel, th n in secondary steelmaking: thermodynam	processes (with and without vacuum). The hermodynamics: laws of thermodynamics, he equilibrium constant of the reaction, the ); Tapping of the steel: reactions that occur erature of the steel. Ladle: refractory lining, e ladle. (2); Deoxidation of liquid steel: hetics of deoxidation product removal from mamics of reactions during degassing and cs of degassing and decarburisation in a hermodynamics of decarburisation of high- nics of desulfurization, desulfurization with		

	synthetic slag on top of liquid steel, desulfurization with powdered reagents. (5); Phosphorus control in secondary steelmaking. Alloy additions. Nonmetallic inclusions and cleanliness of steel: the origin of inclusions, the influence of inclusions on the properties of steel, methods for detection of inclusions and assessment of steel cleanliness, measures to improve the cleanliness of steel during secondary steelmaking, inclusion modification. (4); Continuous casting of steel: caster types, the basic components of caster and principle of operation. Solidification of steel during continuous casting: the structure of continuously cast products, analysis of macrostructure and microstructure of continuously cast products. (3); Tundish: construction, the refractory lining, methods for detection of slag from the ladle, protection of the steel stream from the ladle to the tundish, protection of the top surface of liquid steel in the tundish, flow control of steel level, nonsteady-state. (2); Mould: heat transfer, growth of solid shell, melt flow, mould powder (types, properties, consumption, melting rate, lubrication). (4); Mould: mould types, mould oscillation, mould level control, breakout. (4); Secondary cooling control. (3); Electromagnetic methods for continuous casting: electromagnetic stirrers for billet and bloom casters, electromagnetic equipment for slab casters. (2); Ductility of steel at high temperatures. Surface defects on continuously cast products. Internal defects in continuously cast products. (3). EXERCISES (15): Visits to plants for secondary steelmaking and continuous casting of steel - monitoring of the production process and the casting of steel.						
	□       lectures       □       independer         □       seminars and workshops       □       multimedia		independent	ent study ia and the internet		2.7. Comments:	
2.6. Type of instruction	☐ online in entirety ☐ lab ☐ mixed e-learning ☐ field work ☐		laboratory     work with the     (other)	laboratory     work with the mentor     (other)			
2.8. Student responsibilities	Conditions for signature: Stu	dents must	t attend lectures ar	nd exercises (	(> 70 %).		
2.9. Screening of student's work	Class attendance	0.5	Research			Practical training	
(specify the proportion of ECTS	Experimental work		Report				
credits for each activity so that	Essay		Seminar essay			(Otherdescribe)	
the total number of CTS credits is	Tests	4.5	Oral exam			(Other-describe)	
course)):	Written exam		Project			(Other—describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Exam of the course: Through continuous monitoring - student needs to pass 2 colloquiums. If the student has passed colloquiums, the final score is determined as the average score of the colloquiums. Through the final exam: written exam for students who have not passed the exam through continuous monitoring or are r satisfied with the success that are achieved through continuous monitoring or have not decided on this method of examination.					toring or are not examination.	
2.11 Required literature (available at	Title			Number of copies at the library		Availability via other me	dia
the library and via other media)	Z. Glavaš, A. Štrkalj, S kontinuirano lijevanje, text website of Faculty of Metall Sisak, 2015.	Sekundarna of the lec urgy, Facul	a metalurgija i tures placed on lty of Metallurgy,		https://www sveucil studi	w.simet.unizg.hr/nastava/preda isni-studij-metalurgija/2-godina ija/sekundarna-metalurgija-i-ko lijevanje/view	<u>avanja/diplomski-</u> <u>a-diplomskog-</u> ontinuirano-

	G. Stolte, Secondary Metallurgy – Fundamentals, Processes, Applications, Verlag Stahleisen GmbH, Düsseldorf, 2002.					
2.12. Optional literature (at the time	, The making, shaping and treating of steels, 11th edition, Steelmaking and refining volume The AISE Steel Foundation, 1988.					
of the submission of the study	, The making, shaping and treating of steels, 11th edition, Casting Volume The AISE Steel Foundation, 2003.					
programme proposal)	M. Gojić, Metalurgija čelika, Faculty of metallurgy, Sisak, 2006.					
	Z. Pašalić, Metalurgija čelika, Faculty for metallurgy and materials, Zenica, 2002.					
2.13. Methods of monitoring quality	Survey on the level of faculty and university.					
that ensure acquisition of exit	Analysis predicted in the quality assurance system of institution.					
competences	Analysis predicted in the quality assurance system and authorized office of the university.					

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Explain the chemical reactions that occur during treatment of steel in the ladle.	1st colloquium, written exam
2	Describe and explain the secondary steelmaking processes in ladle.	1st colloquium, written exam
3	Analyze the results of the applied secondary steelmaking process and propose measures to	1st colloquium, written exam
	increase its efficiency.	
4	Describe and explain the process of continuous casting of steel.	2nd colloquium, written exam
5	Explain the solidification of continuously cast products.	2nd colloquium, written exam
6	Analyze the quality of continuously cast products and propose measures to improve their	2nd colloquium, written exam
	quality.	
7	Select measures to increase steel cleanliness during the production process in the steelworks.	1st and 2nd colloquium, written exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:				
1.1. Course teacher	Full Prof. Ladislav Lazić, PhD	1.6. Year of study	2	
1.2. Name of the course	ENERGY MANAGEMENT	1.7. Credit value (ECTS)	4	
1.3. Associate teachers	-	<ol> <li>Type of instruction (number of hours L+S+E+e-learning)</li> </ol>	30+15+0+0	
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	20	
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%	
2. COURSE DESCRIPTION				
2.1. Course objectives	<ol> <li>Acquire knowledge about the particula</li> <li>Acquire knowledge about types of ene</li> <li>Acquire knowledge about importance a importance to thermal processes.</li> <li>Acquire knowledge about the possibilit</li> </ol>	ar forms of energy. rgy conversion from one form to another. and ways of achieving the optimal energy efficien ies of reducing emissions of polluting gases.	ency in industrial processes by giving	
2.2. Enrolment requirements and required entry competences for the course	The acquired knowledge from the courses Numerical modelling of metallurgical proce	of graduate study: Industrial furnaces, Heating esses, Theory of metal forming.	technology of industrial furnaces,	
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Suggest new and improved technical and technological solutions.</li> <li>Analyse the development and application of new technologies.</li> <li>Analyse the production processes by applying thermodynamic laws.</li> <li>Suggest solutions for the optimization of metallurgical processes.</li> <li>Formulate and suggest measures for increasing energy efficiency.</li> </ol>			
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Analyze and compare the ways of con</li> <li>Analyse and plan energy consumption</li> <li>Assess the impact of the combustion plan</li> <li>Apply techniques for improving the en</li> <li>Analyse and choose the methods for construction</li> <li>Evaluate the energy and environment</li> </ol>	nverting the individual of energy forms. In at individual aggregate or plant. process on the environment. lergy efficiency of the combustion process and optimization of energy processes in industry. tal efficiency of applied methods or procedure.	reducing emissions of polluting gases.	
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ol> <li>LECTURES (30):         <ol> <li>General information about significa</li> <li>Energy, basic forms of energy and</li> <li>Processes of energy conversions (</li> <li>Energy consumption in the world, ereducing the consumption of fossil</li> </ol> </li> <li>Possible ways of reducing the energy pumping systems, electric machine</li> <li>Contemporary trends in the improvi</li> </ol>	ance of energy management (2) their division, energy sources (4) (2) emissions of carbon dioxide caused by burning fuels and greenhouse gas emissions (4) 1st colloquium rgy consumption in industry (optimization of inc es) (8) vement of energy efficiency of the combustion p	the fossil fuels, general strategy for dustrial furnaces, steam generators, process and reducing the emissions of	

	polluting gases (4) 7. Optimization of combustion processes: regulation of the air-fuel ratio, combustion air preheating, oxygen enrichment of combustion air (4) 2nd colloquium SEMINAR (15): The understanding of the material exposed in lectures is facilitated by solving the given projects. The projects are selected so that they expand the presented theory and illustrate the application of theory to real problems.							
2.6. Type of instruction	Image: Second de und uney expand the presented theory and metry and metry expand the presented theory and metry and metry expand the presented theory and metry and metry independent study         Image: Second de Und uney expand the presented theory and metry expand the presented theory and metry independent study         Image: Second de Und uney expand the presented theory and metry expand.         Image: Second de Und uney expand the presented theory and metry expand.         Image: Second de Und uney expand. <td< td=""><td colspan="3">2.7. Comments:</td></td<>				2.7. Comments:			
2.8. Student responsibilities	Attendance to Lectures and Se	eminars > 7	0%	1	1			
2.9. Screening of student's work (specify	Class attendance	0.4	Research		Practical traini	ng		
the proportion of ECTS credits for	Experimental work		Report					
each activity so that the total number	Essay		Seminar essay	1.0	(Other	describe)		
of CTS credits is equal to the credit	Tests	1.0	0 Oral exam 1.6		(Other-	(Other-describe)		
value of the course)):	Written exam		Project		(Other-	(Other-describe)		
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Projects – 30% Class attendance – 5% Written exam – 30% Oral exam – 35%	Projects – 30% Class attendance – 5% Written exam – 30% Oral exam – 35%						
		Title		Nu	mber of copies at the library	Availability me	/ via other dia	
2.11. Required literature (available at	L. Lazić, Materials from lectures				0	Merlin sys learr	tem for e- ning	
the library and via other media)	H. Požar, Osnove energetike,	Školska Kn	jiga, Zagreb, 1992		2			
	B. Udovičić, Energetika, Škols	ka Knjiga, Z	Zagreb, 1993.		3			
2.12. Optional literature (at the time of the submission of the study programme proposal)	J. G. Wunning, A. Milani, Hand	book of Bu	rner Technology for Industrial	Furnace	s, Vulkan-Verlag Gm	bH, 2009.		
2.13. Methods of monitoring quality that	Survey at the level of faculty a	nd Universi	ty.					
ensure acquisition of exit	Analyses provided in the syste	em of quality	y assurance of the institution.					
competences	Analyses provided in the system of quality assurance and an authorized office of the University.							

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Analyze and compare the ways of converting the individual of energy forms.	1st colloquium, oral exam
2	Analyse and plan energy consumption at individual aggregate or plant.	1st colloquium, oral exam
3	Assess the impact of the combustion process on the environment.	2nd colloquium, oral exam
4	Apply techniques for improving the energy efficiency of the combustion process and reducing	2nd colloquium, seminar paper, oral exam
	emissions of polluting gases.	
5	Analyse and choose the methods for optimization of energy processes in industry.	2nd colloquium, oral exam
6	Evaluate the energy and environmental efficiency of applied methods or procedure.	2nd colloquium, oral exam

1. COURSE DECRIPTION - GENERAL	INFORMATION	ORMATION ISVU CODE:			
1.1. Course teacher	Assoc.Prof. Zoran Glavaš, PhD	1.6. Year of study	2		
1.2. Name of the course	CASTING OF FERROUS METALS	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	<ul><li>1.8. Type of instruction (number of hours L+S+E+e-learning)</li></ul>	30+0+15+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	20		
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	Introduce students with properties and typ Understanding the correlation of proc microstructural and service properties of for The ability to define the production of meta	es of ferrous casting alloys. ess parameters, melt quality, pouring a errous castings. al castings from unalloyed and alloyed cast ir	nd conditions during solidification with ons and cast steels.		
2.2. Enrolment requirements and required entry competences for the course	-				
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Plan and manage metallurgical processes. Plan the production and casting processes of ferrous and non-ferrous metals.				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Describe the casting and solidification processes of ferrous metal castings in expendable moulds. Relate microstructural and service properties of cast irons and cast steels. Relate chemical composition, metallurgical quality of melt and condition during solidification with resulting microstructural and service properties of castings. Select charge materials for melting aggregates according to casting type and required properties of castings. Select the melt treatment parameters based on chemical composition and metallurgical quality of melt. Select the appropriate quality of cast iron and cast steel depending on the conditions of application of castings.				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Classification of ferrous nucleation and growth of flake graphite composition on microstructure and prope quality control of the melt, quality control nucleation and growth of spheroidal grap composition on microstructure and prope heat treatment of castings, ausferritic duo control of treated melt, quality control of standards, nucleation and growth of com melt production, production processes of cast iron. (3); Risering gray, compacted g of graphitic cast irons, casting modulus, r	s casting alloys. (1); Gray cast iron: basic fe , types of flake graphite, transformations rties, melt production, inoculation, heat treat of castings. (6); Ductile cast iron: basic fe hite, irregular graphite forms, transformation rties, base melt production, nodularizers, ma ctile cast iron (ADI), alloyed ductile cast iron castings. (6); Compacted graphite cast iron pacted graphite, factors influencing properti compacted graphite cast iron, control of the raphite and ductile iron castings: volume cha isering techniques, increasing the efficiency	atures, properties, application, standards, in solid state, the influence of chemical ment of castings, alloyed gray cast irons, atures, properties, application, standards, is in solid state, the influence of chemical agnesium treatment methods, inoculation, s, quality control of the base melt, quality n: basic features, properties, application, es of compacted graphite cast iron, base production process of compacted graphite anges during the cooling and solidification of the riser. (4); Gating systems for gray,		

2.6. Type of instruction		compacted graphite and ductile iron castings: basic components, types of gating systems, the principles of designing a pressurized, non-pressurized and hybrid gating systems in horizontally parted moulds, the principles of designing a gating systems in vertically parted moulds, application of filters. (4); Malleable cast iron: basic features, properties, application, standards, chemical composition, production of whiteheart and blackheart malleable cast iron. (1); White cast irons: basic features, properties, application, standards, pearlitic abrasion resistant cast irons, nickel-chromium abrasion resistant cast irons (Ni-Hard 1, 2, 3 and 4), high-chromium abrasion resistant cast irons, specialty abrasion resistant cast irons, solidification, heat treatment of casting, production. (2); Cast steels: cast carbon steels (basic features, low-carbon cast steels, medium-carbon cast steels, high-carbon cast steels, properties, application), cast low-alloy steels (basic features, cast low-alloy steels for structural components with high strength, hardenability and toughness, cast low-alloy steels for components that need to have wear resistance, abrasion or corrosion resistant steels, cast heat-resistant steels, cast wear-resistant steels, properties, application), cast high-alloy steels (basic features, cast corrosion-resistant steels, cast heat-resistant steels, cast wear-resistant steels, properties, application), production of cast steel melt. (3). EXCERCISES (15): Designing a gating and risering system for gray, compacted graphite and ductile iron castings. (3); Analysis of casting microstructure. (2); Analysis of cooling curves of gray, compacted graphite and ductile cast iron. (2); Visits to cast irons and cast steels foundries. (8)						
2.8. Student responsibilities       Conditions for signature: Students must attend the lectures and exercises (> 70 %).         2.9. Screening of student's work (specify the proportion of ECTS credits for each activity so that the total number of CTS credits is equal to the credit value of the course)):       Class attendance       0.5       Research       Practical training         2.10. Grading and evaluation of student work over the course of instruction and at a final exam       Exam of the course: Through continuous monitoring - student needs to pass 2 colloquiums. Through the final exam: written exam for students who have not passed the exam through continuous monitoring or are not satisfied with the success that are achieved through continuous monitoring or have not decided on this method of examination. Conditions for access to the exam: -         2.11. Required literature (available at the library and via other media)       Z. Glavaš, F. Unkić, Lijevanje željeznih metala, text of the lectures placed on website of Faculty of Metallurgy, Faculty of Metallurgy, Sisak, 2008.       Number of copies at the library       Availability via other media https://www.simet.unizg.hr/nastava/predavanja/diplomski- syeuciligni-studji-metala/view	2.6. Type of instruction	<ul> <li>lectures</li> <li>seminars and workshop</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>	s independent study multimedia and the interr laboratory work with the mentor (other)		rnet	2.7. Comments:		
2.9. Screening of student's work (specify the proportion of ECTS credits for each activity so that the total number of CTS credits is equal to the credit value of the course)):       Class attendance       0.5       Research       Practical training         2.9. Screening of student's work (specify the proportion of ECTS credits for each activity so that the total number of CTS credits is equal to the credit value of the course)):       Class attendance       0.5       Research       Practical training         2.10. Grading and evaluation of student work over the course of instruction and at a final exam       The course: Through continuous monitoring - student needs to pass 2 colloquiums. If the student has passed all colloquiums, the final score is determined as the average score of the colloquiums. Through the final exam: written exam for students who have not passed the exam through continuous monitoring or are not satisfied with the success that are achieved through continuous monitoring or have not decided on this method of examination. Conditions for access to the exam: -         2.11. Required literature (available at the library and via other media)       Z. Glavaš, F. Unkić, Lijevanje željeznih metala, text of the lectures placed on website of Faculty of Metallurgy, Faculty of Metallurgy, Sisak, 2008. <u>https://www.simet.unizg.hr/nastava/predavanja/diplomski- studija/lijevanje-zeljeznih-metala/view</u>	2.8. Student responsibilities	Conditions for signature: Students must attend the lectures and exercises (> 70 %).						
Experimental work       Report         credits for each activity so that the total number of CTS credits is equal to the credit value of the course):       Essay       Seminar essay       (Otherdescribe)         Tests       3.5       Oral exam       (Otherdescribe)       (Otherdescribe)         Written exam       Project       (Otherdescribe)       (Otherdescribe)         2.10. Grading and evaluation of student work over the course of instruction and at a final exam       Exam of the course: Through continuous monitoring - student needs to pass 2 colloquiums. If the student has passed all colloquiums, the final score is determined as the average score of the colloquiums. Through the final exam: written exam for students who have not passed the exam through continuous monitoring or nave not decided on this method of examination. Conditions for access to the exam: -         2.11. Required literature (available at the library and via other media)       Z. Glavaš, F. Unkić, Lijevanje željeznih metala, text of the lectures placed on website of Faculty of Metallurgy, Sisak, 2008.       https://www.simet.unizg.hr/nastava/predavanja/diplomski-sveucilisni-studij-metala/view	2.9. Screening of student's work	Class attendance	0.5	Research			Practical training	
credits for each activity so that the total number of CTS credits is equal to the credit value of the course)):       Essay       Seminar essay       (Otherdescribe)         Viriten exam       Project       (Otherdescribe)         2.10. Grading and evaluation of student work over the course of instruction and at a final exam       Exam of the course: Through continuous monitoring - student needs to pass 2 colloquiums. If the student has passed all colloquiums, the final score is determined as the average score of the colloquiums. If the student has passed all colloquiums, the final exam: written exam for students who have not passed the exam through continuous monitoring or are not satisfied with the success that are achieved through continuous monitoring or have not decided on this method of examination. Conditions for access to the exam: -         2.11. Required literature (available at the library and via other media)       Z. Glavaš, F. Unkić, Lijevanje željeznih metala, text of the lectures placed on website of Faculty of Metallurgy, Sisak, 2008. <u>https://www.simet.unizg.hr/nastava/predavanja/diplomski-sveucilisni-studij-metalargija/2-godina-diplomskog-studija/lijevanje-zeljeznih-metala/view</u>	(specify the proportion of ECTS	Experimental work		Report				
total number of CTS credits is equal to the credit value of the course)):       Tests       3.5       Oral exam       (Other-describe)         Written exam       Project       (Other-describe)         2.10. Grading and evaluation of student work over the course of instruction and at a final exam       Exam of the course: Through continuous monitoring - student needs to pass 2 colloquiums. If the student has passed all colloquiums, the final score is determined as the average score of the colloquiums. Through the final exam: written exam for students who have not passed the exam through continuous monitoring or are not satisfied with the success that are achieved through continuous monitoring or have not decided on this method of examination. Conditions for access to the exam: -       Number of copies at the library       Availability via other media         2.11. Required literature (available at the library and via other media)       Z. Glavaš, F. Unkić, Lijevanje željeznih metala, text of the lectures placed on website of Faculty of Metallurgy, Sisak, 2008.       https://www.simet.unizg.hr/nastava/predavanja/diplomski-studij-metala/view	credits for each activity so that the	Essay		Seminar ess	ау		(Otherdescribe)	
to the credit value of the course)):Written examProject(Other-describe)2.10. Grading and evaluation of student work over the course of instruction and at a final examExam of the course: Through continuous monitoring - student needs to pass 2 colloquiums. If the student has passed all colloquiums, the final score is determined as the average score of the colloquiums. Through the final exam: written exam for students who have not passed the exam through continuous monitoring or are not satisfied with the success that are achieved through continuous monitoring or have not decided on this method of examination. Conditions for access to the exam: -2.11. Required literature (available at the library and via other media)Z. Glavaš, F. Unkić, Lijevanje željeznih metala, text of the lectures placed on website of Faculty of Metallurgy, Sisak, 2008.Number of copies at the library at uignet. studija/lijevanje-zeljeznih-metala/view	total number of CTS credits is equal	Tests	3.5	Oral exam			(Other-describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam       Exam of the course: Through continuous monitoring - student needs to pass 2 colloquiums. If the student has passed all colloquiums, the final score is determined as the average score of the colloquiums. If the student has passed all colloquiums, the final exam: written exam for students who have not passed the exam through continuous monitoring or are not satisfied with the success that are achieved through continuous monitoring or have not decided on this method of examination. Conditions for access to the exam: -         2.11. Required literature (available at the library and via other media)       Z. Glavaš, F. Unkić, Lijevanje željeznih metala, text of the lectures placed on website of Faculty of Metallurgy, Sisak, 2008.       https://www.simet.unizg.hr/nastava/predavanja/diplomski-sveucilisni-studij-metala/view	to the credit value of the course)):	Written exam		Project			(Other-describe)	
Number of copies at the library       Availability via other media         2.11. Required literature (available at the library and via other media)       Z. Glavaš, F. Unkić, Lijevanje željeznih metala, text of the lectures placed on website of Faculty of Metallurgy, Faculty of Metallurgy, Sisak, 2008.       https://www.simet.unizg.hr/nastava/predavanja/diplomski- sveucilisni-studij-metalurgija/2-godina-diplomskog- studija/lijevanje-zeljeznih-metala/view	2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Exam of the course: Throug colloquiums, the final score Through the final exam: wr satisfied with the success t Conditions for access to th	gh continuo is determir itten exam f hat are ach e exam: -	nus monitoring ned as the ave for students wh ieved through o	- student need rage score of t to have not pa continuous mo	s to pass 2 colle he colloquiums ssed the exam nitoring or have	oquiums. If the student has pa through continuous monitoring not decided on this method o	assed all g or are not of examination.
2.11. Required literature (available at the library and via other media)       Z. Glavaš, F. Unkić, Lijevanje željeznih metala, text of the lectures placed on website of Faculty of Metallurgy, Faculty of Metallurgy, Sisak, 2008. <a href="https://www.simet.unizg.hr/nastava/predavanja/diplomski-sveucilisni-studij-metalurgija/2-godina-diplomskog-studija/lijevanje-zeljeznih-metala/view">https://www.simet.unizg.hr/nastava/predavanja/diplomski-sveucilisni-studij-metalurgija/2-godina-diplomskog-studija/lijevanje-zeljeznih-metala/view</a>		Title		Number of copies at the library		Availability via other medi	a	
the library and via other media)       of the lectures placed on website of Faculty of Metallurgy, Faculty of Metallurgy, Sisak, 2008.       sveucilisni-studij-metalurgija/2-godina-diplomskog- studija/lijevanje-zeljeznih-metala/view	2.11. Required literature (available at	Z. Glavaš, F. Unkić, Lijevanie želieznih metala. text			https://www.sir	met.unizg.hr/nastava/predava	nja/diplomski-	
Metallurgy, Faculty of Metallurgy, Sisak, 2008. <u>studija/lijevanje-zeljeznih-metala/view</u>	the library and via other media)	of the lectures placed or	n website o	of Faculty of		sveucilisni-stu	dij-metalurgija/2-godina-diplor	<u>nskog-</u>
		Metallurgy, Faculty of Meta	Illurgy, Sisa	k, 2008.		studija/lijevanj	<u>e-zeljeznih-metala/view</u>	
, Metals Handbook, Volume 15, Casting, ASM CD		, Metals Handbook, Volu	me 15, Cas	sting, ASM			CD	

2.12. Optional literature (at the time of	, Cast Iron, ASM International, Materials Park, 199	9.			
the submission of the study	G. Laird, R. Gundlach, K. Röhrig, Abrasion-Resistant Cast Iron Handbook, American Foundry Society, 2000.				
programme proposal)	Steel Castings Handbook, 6th Edition, editors: N. Blair, T. L. Stevens, ASM International, 1995.				
2.13. Methods of monitoring quality that	Survey on the level of faculty and university.				
ensure acquisition of exit	Analysis predicted in the quality assurance system of	i institution.			
competences	Analysis predicted in the quality assurance system a	nd authorized	office of the university.		

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Describe the casting and solidification process of ferrous metal castings in expendable moulds.	1st and 2nd colloquium, written exam
2	Relate microstructural and service properties of cast irons and cast steels.	1st and 2nd colloquium, written exam
3	Relate chemical composition, metallurgical quality of melt and condition during solidification with	1st and 2nd colloquium, written exam
	resulting microstructural and service properties of castings.	
4	Select charge materials for melting aggregates according to casting type and required	1st and 2nd colloquium, written exam
	properties of castings.	
5	Select the melt treatment parameters based on chemical composition and metallurgical quality	1st and 2nd colloquium, written exam
	of melt.	
6	Select the appropriate quality of cast iron and cast steel depending on the conditions of	1st and 2nd colloquium, written exam
	application of castings.	
7	Solve complex problems in metallurgy of cast irons and cast steels.	1st and 2nd colloquium, written exam

1. COURSE DECRIPTION - GENERAL	INFORMATION		ISVU CODE:		
1.1. Course teacher	Assoc.Prof. Zdenka Zovko Brodarac, PhD	1.6. Year of study	2		
1.2. Name of the course	CASTING OF NON-FERROUS METALS	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	20		
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	Introduce students to properties and types of Understanding the correlation of process par microstructural and service properties oF no The ability to define the production of metal castings ((aluminum, magnesium, copper, r	of nonferrous casting alloys. arameters, melt quality, pouring and onferrous castings. I castings based on unalloyed and al metal matrix composites).	conditions during solidification with loyed cast irons and steels and nonferous metal		
2.2. Enrolment requirements and required entry competences for the course					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in e Plan the production and casting processes Compare the procedures of material treatm	engineering practice. of ferrous and non-ferrous metals. ent with microstructure and useful p	roperties.		
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Describe the casting and solidification process of nonferrous metal castings in expendable and permanent moulds. Relate microstructural and service properties of nonferrous metal castings. Relate the chemical composition, metallurgical quality of melt and solidification condition with resulting microstructural and usage properties of castings. Select charge materials for melting aggregates according to casting type and required properties of castings.				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction to the subject curriculum, method of mastering the material. (1) Classification of non-ferrous casting alloys. Aluminium casting alloys in general. Application of alloys. (5) Classification of alloys: norms. Solubility of elements. Metallurgical state: Heat treatment. Criterions for alloy selection. Properties of Al-casting alloys. Influence of alloying and trace elements. Microstructures overview. (4) Melt treatment of Al-casting alloys: degassing, filtration, circulation. (4) Methods of melt treatment of Al-casting alloys: grain refinement, modification of microstructure. (4) Solidification and development of cast microstructure. Fundamentals of nucleation. Crystal growth. Solidification types. Determination of nucleation potential and solidification kinetics. Classification of casting processes of aluminum alloys. (4) Implementation of new strategies and concepts in production projection. Disadvantages of conventional casting processes. Innovative casting processes. Recycling of aluminium alloys. Technology of recycling. Casting defects. (3) Magnesium and copper alloys: classification and designation properties and application production processes beat treatment				

	corrosion protection. (3) Casting of metal matrix composites. (2)						
	Visit to the relevant foundries casting of non-ferrous metals (15).						
	⊠ lectures		2.7. Comments:	2.7. Comments:			
2.6. Type of instruction	Seminars and worksr	Iminars and workshops       Imilars and workshops         icercises       Imilars and the internet         Inine in entirety       Iaboratory         ixed e-learning       (other)					
2.8. Student responsibilities	Attending the classes >70%.						
2.9. Screening of student's work (specify	Class attendance	1	Research		Practical training		
the proportion of ECTS credits for	Experimental work		Report	1			
each activity so that the total	Essay		Seminar essay		(Otherdescri	ibe)	
number of CTS credits is equal to	Tests	2	Oral exam		(Other-describe)		
the credit value of the course)):	Written exam		Project		(Other-desc	ribe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Class attendance. Examination through continuous monitoring: 1. Colloquium-after the unit Methods of melt treatment of Al-casting alloys: grain refinement, modification of microstructure, 2. Colloquium-after the unit Casting of metal matrix composites or final examination: written and oral.						
			Title		Number of copies at the library	Av	ailability via other media
	Metals Handbook, Volur Ohio, 1988.	1					
2.11. Required literature (available at the library and via other media)	J. G. Kaufman, E. L. Ro and Applications, ASM I	oy, Aluminu nternationa	um Alloy Castings Prop al, Metals Park, Ohio, 20	erties, Processes 005.	1		
	Aluminum and Aluminur International, ed. J. R. D	n Alloys. As Davis, Mater	SM Specialty Handbool rials Park, Ohio, USA, 2	k, ASM 2002.	1		
	Aluminum Casting Technology, American Foundrymen's Society, Illinois, 1 1997.						
2.12. Optional literature (at the time of the submission of the study programme proposal)	T. Nishizawa, Thermodynamics of microstructures. ASM International, Materials Park, Ohio, 2008 J. Campbell, Castings, Butterworth Heinemann, Oxford, 1991.						
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Survey at the Faculty and University level. Analysis provided the quality assurance system of the institution. Analysis provided the quality assurance system and authorized Office of the University						

Ordinal number	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning outcomes
1	Describe the casting and solidification process of nonferrous metal castings in expendable and permanent moulds.	1st colloquium, auditory exercises, written and oral exam
2	Relate microstructural and service properties of nonferrous metal castings.	1st colloquium, auditory exercises, written and oral exam
3	Relate the chemical composition, metallurgical quality of melt and solidification condition with	1st colloquium, auditory exercises, written and oral exam
	resulting microstructural and usage properties of castings.	
4	Select charge materials for melting aggregates according to casting type and required	2nd colloquium, auditory exercises/field work, written and
	properties of castings.	oral exam
5	Select the melt treatment parameters based on chemical composition and metallurgical quality	2nd colloquium, auditory exercises/field work, written and
	of melt.	oral exam
6	Describe the casting and solidification process of nonferrous metal castings in expendable and	2nd colloquium, auditory exercises/field work, written and
	permanent moulds.	oral exam

1. COURSE DECRIPTION - GENERAL	AL INFORMATION ISVU CODE:				
1.1. Course teacher	Assoc.Prof. Anita Štrkalj, PhD Assoc.Prof. Tamara Holjevac Grgurić, PhD	1.6. Year of study	2		
1.2. Name of the course	EXPERIMENTAL TECHIQUES IN METALLURGY	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	20		
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	To adopt a basic knowledge about expe To introduce new technological methods To train students for autonomous exper	rimetal techniques in the area of metallurgy s for investigation of materials in foundry and imental work.	d steel works.		
2.2. Enrolment requirements and required entry competences for the course					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Suggest appropriate methods for material quality analysis. Plan and manage metallurgical processes				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Plan the technological investigations of metallurgical materials. Determine radioactivity of the scrap. Analysis of flue gas/chimney content. Analyze the quality and stability of the material in the foundry industry. Define experimental techniques for measurement of pressure, temperature and flow. Select and apply appropriate experimental technique for monitoring of metallurgical processes				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Theoretical basics of experimental techniques. (3) Analysis of metallurgical slag, electric arc-furnace dust and foundry sand. (4) Gravimetry (the content of solid particles in flue gases) (4) Measuring of radioactivity in the steel scrap. (4) Measuring of oxygen activity in the melt. (3) Experimental techniques for measurement of pressure. (4) Experimental techniques for measurement of temperature. (4) Experimental techniques for measurement of flow. (4) Two colloquiums through continuos monitoring. LABORATORY EXERCISES (15): Measuring of control temperature.				

	Determination of acid Determination of chro	Determination of acid demand value of foundry sands. Determination of chromite content in silicon sands.							
	Determination of mas	s loss by	calcination of chromite.						
	Determination of cher	nical cont	ent of flue gas.						
	Measuring of tempera	Measuring of temperature (solid, fluid).							
	Mesuring in steel works.								
	Measuring in foundry.								
	Characterization of be	Characterization of bentonites and sands as well as thermal stability of metallurgical materials (DSC/TG).							
	☐ independent study 2.7. Comments:								
	seminars and work	kshops	multimedia and the	internet					
2.6. Type of instruction			laboratory						
			work with the mentor						
			(other)						
2.8 Student responsibilities	Attendence te lectures min 70 % Attendence te leb presties min 70 % Leb reports								
	Class attendance	Allendence to rectures min 70 %. Allendance to lab practice min. 70 %							
2.9. Screening of student's work (specify		1	Research						
the proportion of ECIS credits for each	Experimental work	1	Report						
activity so that the total number of CTS	Essay		Seminar essay		(Otherdescribe)				
credits is equal to the credit value of the	lests		Oral exam	1	(Other—describe)				
course)).	Written exam	1	Project		(Other—describe)				
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Two colloquiums thro	ugh contii	nuos monitoring and ora	l exam or final e	exam: written and oral.				
2.11. Required literature (available at			Title		Number of copies at the library	Availability via other media			
the library and via other media)	Mold & Core Test Har	ndbook, A	American Foundry Socie	ty, Inc. 2006.	1				
2.12. Optional literature (at the time of	R. Halmshaw, Non-de	estructive	Testing, Buttervorth-He	inemann Ltd., 19	991.				
the submission of the study	J. L. Taylor, Basic Me	etallurgy fo	or Non-Destructive Testi	ng, British Institu	ute for Non-dstructive Testing, 1	996.			
programme proposal)	Occurance and Testir	ng of Four	ndry Mouding Sands, L.I	H. Cole, Nabu P	ress, 2010.				
2.12 Methods of monitoring quality that	Student survey input	and outpu	It. Numerical analysis of	tests and exam	s according to scoring task by ta	ask at the level of course.			
2.13. Methods of monitoring quality that	Survey at the level of	faculty ar	nd University.						
	Analyses provided in	the system	m of quality assurance c	of the institution.					
competences	Analyses provided in the system of quality assurance and an authorized office of the University								

Ordinal number	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning outcomes
1	Plan the technological investigations of metallurgical materials.	1st and 2nd colloquium, laboratory exercises, written
		exam
2	Determine radioactivity of the scrap.	1st colloquium, written exam
3	Analysis of flue gas/chimney content.	1st and 2nd colloquium, written exam, oral exam
4	Analyze the quality and stability of the material in the foundry industry.	1st colloquium, laboratory exercises, written exam, oral
		exam
5	Define experimental techniques for measurement of pressure, temperature and flow.	1st and 2nd colloquium, laboratory exercises, written
		exam, oral exam
6	Select and apply appropriate experimental technique for monitoring of metallurgical processes.	1st and 2nd colloquium, laboratory exercises, written
		exam, oral exam

1. COURSE DECRIPTION - GENERAL	I – GENERAL INFORMATION ISVU CODE:				
1.1. Course teacher	Assoc.Prof. Zdenka Zovko Brodarac, PhD	1.6. Year of study	2		
1.2. Name of the course	OPTIMISATION OF CASTINGS FORMING	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	20		
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION	-		-		
2.1. Course objectives	Getting acquainted with modern concepts of design and development of metal castings using computer packages. Optimization of construction castings, tools, models and prototypes by application of informatics technology. Macro and micro modelling of commercial alloys solidification. The use of computers in planning and guality assurance. Product data management.				
2.2. Enrolment requirements and required entry competences for the course					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Analyse the development and application of new technologies. Plan the production and casting processes of ferrous and non-ferrous metals				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Elaborate technological phases of casting production. Select the method of rapid prototyping. Optimization of the casting, models and tools construction using informatics technology.				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction to the plan and syllabus of the course, and ways for successful overcoming. (1) Introduction in of contemporary conceptions of production of castings. (5) Optimization of castings, of tools, models constructions by applying informatics technology. Optimization of prototypes constructions by applying informatics technologies. (4) Rapid prototyping. (3) Simulations of mold filling and solidification processes. Stochastic modeling solidification. (5) Optimization melts quality via informatics technologies: expert systems. Optimization melts quality via informatics technologies: methods of artificial inteligency. Optimization quality melt via information technologies: computer aided thermal analysis. (5) Macro and micro modeling solidification of commercial alloys: commercial alloys. (5) Application of computers in planning and quality assurance. Product data management. (2) EXERCISES (15): Making stl model default casting (5). Numerical simulation of casting and solidification (7). Prediction and correction of errors (1).				

	☐ lectures		independent study 2.7. C		2.7. Co	mments:		
2.6. Type of instruction	<ul> <li>Seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>		<ul> <li>multimedia and the internet</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>					
2.8. Student responsibilities	Attending the classes. 1. colloquium Project and presentation.							
2.9 Screening of student's work (specify	Class attendance	1	Research		Practica	al training		
the proportion of ECTS credits for	Experimental work		Report	1				
each activity so that the total	Essay		Seminar essay		(	Otherdescribe)		
the credit value of the course).	Tests	2	Oral exam		(	Other—describe)		
	Written exam		Project		(	Other—describe)		
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Attending the classes. Project task. One tests through continuou	is monitorin	g or final examination (writt	en and oral).				
			Title			Number of copies at the library	Availabilit other me	ty via edia
2.11 Deguined literature (evolution of	Metals Handbook, Volume 15, CASTING, ASM International, Metals Park, Ohio, 1988			1				
2.11. Required interature (available at the library and via other media)	TMS, Modelling of casting, welding and advanced solidification processes, Illinois, 1998				1			
the library and via other media)	J. P. Womack, D. T. Jones, D. Roos, The machine that changed the world, New York, 1991.							
	1991.	D. R008, 11	he machine that changed th	e world, New \	′ork,	1		
	1991. M. Imaj, Kaizen, Ključ japans	skog poslov	ne machine that changed th nog uspjeha, Beograd, 200	e world, New \ 08.	ork,	1		
	1991. M. Imaj, Kaizen, Ključ japans	skog poslov	ne machine that changed th nog uspjeha, Beograd, 200	e world, New א 8.	′ork,	1		
2.12. Optional literature (at the time of the submission of the study programme proposal)	1991. M. Imaj, Kaizen, Ključ japan:	skog poslov	ne machine that changed th	e world, New \ 08.	/ork,	1		
<ul> <li>2.12. Optional literature (at the time of the submission of the study programme proposal)</li> <li>2.13. Methods of monitoring quality that</li> </ul>	1991.     M. Imaj, Kaizen, Ključ japan:     -     Survey at the Faculty and Ut	skog poslov	ne machine that changed th nog uspjeha, Beograd, 200 vel.	e world, New \ 08.	′ork,	1		
<ul> <li>2.12. Optional literature (at the time of the submission of the study programme proposal)</li> <li>2.13. Methods of monitoring quality that ensure acquisition of exit</li> </ul>	A solve a revided the quality	niversity lev	vel.	e world, New \ 18.	oroity	1		

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning
number		outcomes
1	Elaborate technological phases of casting production.	1st colloquium, auditory exercises, written and oral exam
2	Select the method of rapid prototyping.	1st colloquium, auditory exercises, written and oral exam
3	Optimization of the casting, models and tools construction using informatics technology.	Laboratory exercises/independent work, written and oral exam
4	Analyse cast product data.	Independent work, written and oral exam

1. COURSE DECRIPTION - GENERAL	URSE DECRIPTION – GENERAL INFORMATION ISVU CODE:				
1.1. Course teacher	Assist.Prof. Tahir Sofilić, PhD Assoc.Prof. Ivan Brnardić, PhD Assoc.Prof. Tamara Holjevac Grgurić, PhD	1.6. Year of study	2		
1.2. Name of the course	THE BEST AVAILABLE TECHNIQUES IN METALLURGY	1.7. Credit value (ECTS)	5		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	45+15+15+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	20		
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	To enable engineers for managing of the metallurgical processes, effective control of metallurgical industrial activities giving priority to intervention at source in the production plant, ensuring management of natural resources according to the economic situation and specific characteristics of the local area. Introduction to the industrial emissions directive 2010 / 75 / EU and to the integrated approach to prevention and control of environmental emissions, waste management as well as energy efficiency and accident prevention. Introduction to European reference documents (BREF) with review of the best available techniques (BAT) relating to the metallurgical processes, especially the processes of iron and steel production, as well as processes of production of non-ferrous metals.				
2.2. Enrolment requirements and required entry competences for the course					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Suggest solutions for the optimization of metallurgical processes. Formulate and suggest measures for increasing energy efficiency.				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Define and explain the concept of the best available techniques (BAT). To choose BAT in the metallurgical process. Use legislation of environmental protection in the metallurgical processes.				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (45): Trends of iron and steel production in the world, EU and Republic of Croatia. (1) Management of raw materials and energy in the iron and steel production. (1) Consumption of materials and energy in the metallurgical processes. (1) Emissions of pollutants. (2) Emissions into the air. (1) Materials management. (1) Storage and handling of raw materials. (1) Management of production residues. (1)				

Management of waste water. (1)
Introduction and selection of the best available techniques in metallurgy. (1)
Techniques to increase energy efficiency. (1)
Techniques for optimizing consumption of fuels. (1)
Techniques for improving effectivnes of waste heat. (1)
Consumption of materials and energy as well as emissions in iron production by blast furnace process. (1)
The emission of pollutants into the environment during blast furnace processes. (1)
The best available techniques in blast furnace process. (2)
The formation of slag and its treatment (granulation and pelletizing). (2)
Storage and handling of raw materials in electric arc furnace process (EAF). (1)
Pre-treatment of the ingot for steel production in the EAF. (1)
Pre-heating of the steel charge for steel production in the EAF. (1)
Smelting, refining and casting. (1)
Secondary metallurgy. (1)
The formation of slag (EP and LP) and its treatment. (1)
Consumption of materials and energy in the steel production process. (1)
Pollutant emissions from steel production process. (2)
Emissions of pollutants into the air from the steel production process. (1)
Water management and waste water from the steel production process. (1)
Noise emissions. (1)
Soil pollution. (1)
The best available techniques in the steel production process. (2)
Consumption of raw materials and energy as well as emissions from cooper production. (1)
The best available techniques in the cooper production process. (2)
Consumption of raw materials and energy as well as emissions from aluminium production. (1)
The best available techniques in the aluminium production process. (2)
Consumption of raw materials and energy as well as emissions from production of lead and tin. (1)
The best available techniques in the production of lead and tin. (1)
Consumption of raw materials and energy as well as emissions from production of zinc and cadmium. (1)
The best available techniques in the production of zinc and cadmium. (1)
SEMINAR(15):
Instructions for the preparation of the seminar. (2)
Presentation of seminar topics and the selection (1)
Individual students work under supervision. (6)
Written seminar and preparation of ppt presentation. (1)
Oral presentations. (5)
EXERCISES(15):
Auditory exercises - comparison of monitoring results of air emissions from the steel production process in Croatia with related
BREF data. (5)
A comparison of production waste (waste and / or by-products) from the steel production process in the Republic of Croatia with
related BREF data. (3)
Field work: metallurgical industrial processes and introduction to applied BAT. (7)

	PRELIMINARY EXAMS: 1.preliminary exam. Pollutant emissions in the production of iron and steel, emissions of pollutants into the air from the production processes, raw materials management, storage and handling of raw materials, waste residues, waste water management, review and selection of the best available techniques in metallurgy, techniques for increasing energy efficiency, techniques for optimizing the consumption of fuels, techniques for improving waste heat utilization, consumption of materials and energy as well as emissions in iron production by blast furnace process, the emission of pollutants into the environment during blast furnace processes, the best available techniques in blast furnace process, formation of slag and its treatment (granulation and pelletizing).						
	2.preliminary exam. Storage and handling of raw materials in electric arc furnace process (EAF), pre-treatment of the ingot for steel production in the EAF, pre-heating of the steel charge for steel production in the EAF, smelting, refining and casting, secondary metallurgy, formation of slag (ELP and LP), consumption of materials and energy in the steel production process, pollutant emissions from steel production process, emissions of pollutants into the from the steel production process, water management and waste water from the steel production process, noise emissions, soil pollution, the best available techniques in the steel production process.						
	3.preliminary exam. Consumption of raw materials and energy as well as emissions from cooper production, the best available techniques in the cooper production process, consumption of raw materials and energy as well as emissions from aluminium production, the best available techniques in the aluminium production process, consumption of raw materials and energy as well as emissions from aluminium production, the best available techniques in the aluminium production process, consumption of raw materials and energy as well as emissions from production of lead and tin, the best available techniques in the production of lead and tin, consumption of raw materials and energy as well as emissions from production of zinc and cadmium, the best available techniques in the production of zinc and cadmium.						
	Iectures	kshops	🛛 independen	t study	2.7.0	comments:	
2.6 Type of instruction	✓ seminars and workshops ✓ exercises						
	online in entirety		$\boxtimes$ work with th	e mentor			
	initial e-learning (other)						
2.8. Student responsibilities	Attendance to lectures min 70 %. Attendance to seminar min 70 %. Written seminar and oral presentation.						
2.9. Screening of student's work (specify	Class attendance		Research		Pract	ical training	
the proportion of ECTS credits for	Experimental work	1	Report				
each activity so that the total	Essay	2	Seminar essay	1		(Other-describe)	
the credit value of the course)):	Vritten exem	3	Drai exam Project			(Other_describe)	
2 10 Grading and evaluation of student	Seminar results of p	I reliminary	exams as well as	written and oral (	exams		1
work over the course of instruction and at a final exam							
2.11. Required literature (available at the library and via other media)	Title			Number of cop at the library	oies y	Availability via oth	er media

	Best Available Techniques (BAT) Reference Document for Iron and Steel Production		http://eippcb.jrc.ec.europa.eu/reference/BREF/IS_Adopt ed_03_2012.pdf		
	Best Available Techniques (BAT) Reference Document for the Non-Ferrous Metals Industries		http://eippcb.jrc.ec.europa.eu/reference/BREF/NFM_Fin al Draft 10 2014.pdf		
	DIREKTIVA 2010/75/EU EUROPSKOG PARLAMENTA I VIJEĆA od 24. studenoga 2010. o industrijskim emisijama (integrirano sprečavanje i kontrola onečišćenja)		http://eur-lex.europa.eu/legal- content/HR/TXT/?uri=celex:32010L0075		
2.12. Optional literature (at the time of the submission of the study programme proposal)	Sofilić T., Jendričko J., PCDDs/Fs Pollution from Metallurgical Processes in the Town of Sisak, Croatia, Archives of metallurgy and materials, <b>59</b> , 1 (2014) 293-297. Sofilić T., Unkić F., Direktiva IPPC (96/61/EC) i njezin značaj za hrvatske čeličane i ljevaonice, Ljevarstvo, <b>50</b> , 4 (2008) 107-117. Sofilić T., Rastovčan-Mioč A., Šmit Z., Čeličanska elektropeć kao izvor emisije polikloriranih dibenzo- <i>p</i> -dioksina i dibenzofurana u svijetlu Direktive Vijeća (96/61 EC) o cjelovitom sprječavanju i kontroli onečišćenja, Kem.Ind. <b>57</b> , 1 (2008) 9-18.				
2.13. Methods of monitoring quality that	Student survey input and output. Numerical analysis of tests and exams according to scoring task by task at the level of course.				
ensure acquisition of exit	Survey at the level of faculty and University.				
competences	Analyses provided in the system of quality assurance of the institution.				

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning		
number		outcomes		
1	Define and explain the concept of the best available techniques (BAT)	1st colloquium, seminar, written and oral exam		
2	To choose BAT in the metallurgical process	1st colloquium, 2nd colloquium, seminar, written and oral exam		
3	Use legislation of environmental protection in the metallurgical processes	2nd colloquium, 3rd colloquium, seminar, written and oral exam		
4	Understand application of the best available techniques to protect the environment	3rd colloquium, seminar, written and oral exam		

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Full Prof. Stoja Rešković, PhD Assoc.Prof. Natalija Dolić, PhD	1.6. Year of study	2			
1.2. Name of the course	SHAPING OF NON-FERROUS METALS AND THEIR ALLOYS	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	Assist.Prof. Ivan Jandrlić, PhD	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	20			
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	2., 10%			
2. COURSE DESCRIPTION	-					
2.1. Course objectives	<ol> <li>Introducing students to the properties of</li> <li>Introducing students with the theoretical</li> <li>Introducing students with technologies o</li> </ol>	non-ferrous metals bases of forming by deformation f shaping by deformation				
2.2. Enrolment requirements and required entry competences for the course	Passed exam from Theory of metal forming					
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Analyse and combine metal forming processes.</li> <li>Suggest solutions for the optimization of metallurgical processes.</li> </ol>					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Apply scientific principles important to the formation of non-ferrous metals,</li> <li>Create parameters of forming non-ferrous metal by deformation,</li> <li>Calculate and evaluate the influential parameters on forming process,</li> <li>Recommend individual metal forming processes.</li> </ol>					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ul> <li>LECTURES (30) AND EXERCISES (15):</li> <li>1. Introduction. Overview of methods for shaping non-ferrous metals. 2</li> <li>2. Technology of industrial casting of aluminum and aluminum alloys intended for further processing. Horizontal ("HDC") and vertical ("VDC") casting process by direct cooling water. 2.</li> <li>3. Homogenization of casted aluminium ingots. Technology for casting aluminum billets by direct cooling with water. 2</li> <li>4. Exercise: Control of casting parameters, constancy and repeatability casting parameters (examples from practice). 1</li> <li>5. I Colloquium, chapters 2, 3 and 4</li> <li>6. Properties of non-ferrous metals. 2</li> <li>7. Laboratory exercises. Determination of deformation resistance of non-ferrous metals and their alloys. 4</li> <li>8. Characteristics of the plastic processing and properties of Al and Al-alloys. 1</li> <li>9. Hot and cold deformation. 2</li> <li>10. The rolling of aluminium or aluminium alloy: Rolling of profiles. 2</li> </ul>					

	<ul> <li>12. New Al-alloys. 1</li> <li>13. Pressing with extrusion and flow. The deep drawing. 2</li> <li>14. Auditory exercises. Calculation of the rolling Al-alloys. 5</li> <li>15. II Colloquium, chapters 1, 6-13</li> <li>16. Characteristics of the plastic processing and properties of Cu and Cu-alloys. 2</li> <li>17. Hot and cold deformation. 2</li> <li>18. Rolling of Cu-alloys. 2</li> <li>19. Pressing with extrusion and flow. The deep drawing. 2</li> <li>20. Auditory exercises. Calculation of extrusion of Cu-alloy. 5</li> <li>21. Other non-ferrous metals and alloys. Overview on procedures of plastic deformation. 3</li> <li>22. Technical and economic indicators of the process of plastic processing of nonferrous metals and their alloys. 1</li> </ul>						
	☐ lectures ☐ independent study 2.7. Comments:						
2.6. Type of instruction	<ul> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>		<ul> <li>multimedia and the internet</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>				
2.8. Student responsibilities	Attendance at lectures 70%, ir before writing the colloquium of	ndividual wo	ork on all exercises and prepa n exam.	aration and sub	mission of reports from	field of colloquium	
2.9. Screening of student's work	Class attendance Resea		Research	0.5	Practical training		
(specify the proportion of ECTS	Experimental work	1	Report				
credits for each activity so that the	Essay Seminar essay (Otherd		(Otherdescri	be)			
total number of CTS credits is equal	Tests	1	Oral exam	1	(Other-descr	ibe)	
to the credit value of the course)):	Written exam		Project	0.5	(Other-descr	ibe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	The presence and activity of students on classes during the classes are evaluated. Students score participation on projects. Score on written colloquiums trough continuous monitoring (or written exam) and oral exam. Score of seminar paper.						
		Number of copies at the library	Availability via other media				
2.11. Required literature (available at	A. Tripalo, Tehnologija prerad	2					
the library and via other media)	D. G. Eskin, Physical Metallurgy of Direct Chill Casting of Aluminium Alloys, CRC 3 Press/Taylor and Francis Group, Boca Raton, 2008.						
	S. Rešković, Tehnologija oblik	10					
2.12. Optional literature (at the time of the submission of the study programme proposal)	S. Rešković, Teorija oblikovanja deformiranjem, Sveučilište u Zagrebu, Metalurški fakultet, Sisak 2014., peer reviewed lessons. Professional journals, Metallurgy, Mechanical Engineering - articles from this area.						
2.13. Methods of monitoring quality that	Survey on the level of faculty and University.						
ensure acquisition of exit	Analyses provided by quality assurance system of the institution.						

competences Analyses provided by quality assurance system and authorized office of the University.

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning outcomes
number		
1	Apply scientific principles important to the formation of non-ferrous metals.	1st colloquium, written and oral exam
2	Create parameters of forming non-ferrous metal by deformation.	Laboratory exercises, auditory exercises, project task
3	Calculate and evaluate the influential parameters on forming process.	2nd colloquium, auditory exercises, independent task, written and oral exam
4	Recommend individual metal forming processes.	3rd colloquium, seminar paper, written and oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:								
1.1. Course teacher	Full Prof. Mirko Gojić, PhD Assoc.Prof. Stjepan Kožuh, PhD	1.6. Year of study	2					
1.2. Name of the course	SURFACE TREATMENT	1.7. Credit value (ECTS)	3					
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+15+0+0					
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	20					
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%					
2. COURSE DESCRIPTION								
2.1. Course objectives	Knowing the basic physical-chemical princ Gaining insight into the many benefits of in Selecting the best methods for appropriate	iples of modification and coating. Idividual methods of surface treatment. application conditions of the machine parts an	d tools					
2.2. Enrolment requirements and required entry competences for the course	-							
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Design the properties of metallic materials. Compare the procedures of material treatment with microstructure and useful properties.							
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Define the concept of surface treatment material. Compare the most important characteristics of each surface treatment process. Categorize and interfuse the individual actions of modification and coating of metal surfaces. Create the appropriate procedure for the protection of materials and structures. Analyze the properties of metallic coatings. Compare methods for non-metallic inorganic and organic coatings.							
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Week 1 and 2: Classification of processes. Surface preparation for surface treatment (4 hours). Weeks 3 and 4: The mechanical modification (sandblasting). Methods for heat hardening of surfaces (flame, induction, surface hardening by laser and electron beam) (4 hours). Week 5-7: High temperature thermomechanical processes of surface hardening (carbonising, nitration, boronising) (6 hours). Week 8-10: Surface modification and deposition of thin films (ion implantation - 2 hours, deposition from the vapor phase by physical way-2 hours, the vapor phase deposition by chemical way- CVD and plasma CVD method-2 hours) Week 11: Hot dipping (galvanizing, application of aluminium and tin) (2 hours). Week 12: Electroplating (plating with zinc, nickel, tin, copper, chromium) (2 hours). Week 13: Enameling. Deposition of precious metals (Au, Ag, etc.) (2 hours). Week 14: Browing. Anodising. Phosphating. Deposition of chromium. Patination (2 hours). Week 15: Organic coatings (coloring, varnishing, plastification, bitumenizing) (2 hours). SEMINAR (15): The selection of topics and seminar work in writing form by a mentor system (10 hours). Preparation and							
	presentation of the seminar and discussions in relation to the topic of the present paper (5 hours).							
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2.6. Type of instruction	<ul> <li>lectures</li> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>		<ul> <li>independent study</li> <li>multimedia and the internet</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>		.7. Comments:			
2.8. Student responsibilities	Students must attend over 70	)% of lectur	res and are required to complete	e a seminar in wr	iting form and orally	present.		
2.9. Screening of student's work (specify	Class attendance	0.3	Research	P	ractical training			
the proportion of ECTS credits for	Experimental work		Report					
each activity so that the total number	Essay		Seminar essay	0.5	(Otherdescribe	)		
of CTS credits is equal to the credit	Tests	2.2	Oral exam		(Other-describe	e)		
value of the course)):	Written exam		Project		(Other-describe	e)		
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	<ul> <li>evaluation of students activi</li> <li>evaluation of written examin</li> <li>evaluation of seminar paper</li> </ul>	ties in cour ation (two r and prese	rse, colloquiums) through continuou entation.	is monitoring or fi	nal examination (writ	ten and oral),		
			Title		Number of copies at the library	Availability via other media		
2.11. Required literature (available at	M. Gojić, Površinska obradba	10						
the library and via other media)	I. Esih, Osnove površinske za	3						
	ASM Handbook.pdf					Electronic form		
2.12. Optional literature (at the time of the submission of the study programme proposal)	of T. Filetin, K. Grilec, Postupci modificiranja i prevlačenja površina, Hrvatsko društvo za materijale i tribologiju, Zagreb, 2004. D. Krumes: Površinske toplinske obrade i inženjerstvo površine, Strojarski fakultet Slavonski Brod, Sveučilište u Osijeku, Slavonski Brod, 2004.							
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Student survey input and output. Survey at the level of faculty and University. Analyses provided in the system of quality assurance of the institution. Analyses provided by quality assurance system and authorized office of the University.							

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of		
number		learning outcomes		
1	Define the concept of surface treatment material.	1st colloquium, written and oral exam		
2	Compare the most important characteristics of each surface treatment process.	1st colloquium, written and oral exam		
3	Categorize and interfuse the individual actions of modification and coating of metal surfaces.	1st colloquium, written and oral exam		
4	Create the appropriate procedure for the protection of materials and structures.	2nd colloquium, written and oral exam, seminar paper		
5	Analyze the properties of metallic coatings.	2nd colloquium, written and oral exam		
6	Compare methods for non-metallic inorganic and organic coatings.	2nd colloquium, written and oral exam, seminar paper		

1. COURSE DECRIPTION - GENERAL	INFORMATION	MATION ISVU CODE:					
1.1. Course teacher	Assoc.Prof. Ivan Brnardić, PhD	1.6. Year of study		2			
1.2. Name of the course	CORPORATE SOCIAL RESPONSIBILITY	1.7. Credit value (ECTS)		4			
1.3. Associate teachers	-	1.8. Type of instruction (number learning)	of hours L+S+E+e-	30+15+0+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the c	course	20			
1.5. Status of the course	compulsory	1.10. Level of use of e-learning percentage of instruction in (20% maximum)	(1, 2, 3 level), the course on line	1., 5%			
2. COURSE DESCRIPTION							
2.1. Course objective	<ol> <li>To introduce students with corporate</li> <li>To introduce with the strategy of susta</li> <li>To train for self-employment and / or sustainable development.</li> </ol>	socially responsible. ainable development in the Croatia. presentation of the institution / comp	oany / community / city	/ county in matters of			
2.2. Enrolment requirements and required entry competences for the course	Knowledge on industrial ecology and idea of sustainable development and work on computers.						
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Compare and choose individual technological process. Combine the skills necessary for lifelong learning, including continued professional training						
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Define and explain the concepts of corporate social responsibility.</li> <li>Know the guidelines of the strategy of sustainable development in Croatia.</li> <li>How to use regulations and scientific-technical literature in the field of corporate social responsibility.</li> <li>Understand and advocate for consistent application of the principles of corporate social responsibility.</li> <li>Design and implement environmentally, economically and socially responsible business practices and socially responsible investment community.</li> </ol>						
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): What is corporate social responsibility. 3h Sustainable Development Strategy of the Croatia. 8h Customer relationship management with the aim of fitting social responsibility in marketing activities. 4h Responsible operations in work safety improving. 2h The development and the state of CSR in Croatia and abroad. 3h Examples from practice. 10h SEMINAR (15): Example activities and proportion and proportion of comingr paper						
2.6. Type of instruction	Image: Sector secto	Image: Secondation of contract paper.         Image: Independent study         Image: I					

	mixed e-learning     field work			(other)					
2.8. Student responsibilities	Regular attendance of lectu	res (min. 70%	of the	e lectures) and s	semir	nar work	•		
2.9. Screening of student's work	Class attendance	0.5	Rese	earch			Practical training		
(specify the proportion of ECTS	Experimental work		Repo	ort					
credits for each activity so that the	Essay		Semi	inar essay	1.0		(Otherdescribe)		
total number of CTS credits is equal	Tests		Oral e	exam	1.2	5	(Other—describe)		
to the credit value of the course)):	Written exam	1.25	Proje	ect			(Other-describe)		
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Attendance on classes, pre and oral exam.	paration and p	oresen	ntation of semina	ar pap	per, con	tinuous monitoring – 2 prelim	inary exams or written	
	Title			Number of cop at the library	ies /		Availability via othe	er media	
	I. Brnardić, Lectures from C responsibility, Sisak, 2016.	orporate socia	al			Merlin	system for e-learning		
	L. Pavić Rogošić, Corporate social responsibility, Odraz, Zagreb, 2015.					<u>http://v</u>	http://www.odraz.hr/media/21845/dop.pdf		
	Strategy of sustainable development of the Republic Croatia, OG 110/07, RH, 2009.					http://narodne- novine.nn.hr/clanci/sluzbeni/2009_03_30_658.html			
2.11. Required literature (available at the library and via other media)	A. Glavočević, A. Radman Peša, Corporate social responsibility in CRM as a way of social responsibility integrating in marketing activity, Oeconomica Jadertina 2/2013.					https:// web&c hXG8F ak.srce XJcdlx	/www.google.hr/url?sa=t&rct= :d=1&cad=rja&uact=8&ved=0 RQKHcxPBRYQFggcMAA&u a.hr%2Ffile%2F170022&usg= :plamFJaP7CuEp6aYdRg	<u>-j&amp;q=&amp;esrc=s&amp;source=</u> )ahUKEwjCksTLxvzNA rl=http%3A%2F%2Fhrc =AFQjCNG-	
	V. Nikolić, S. Savić, J. Taradi, Models of corporate social responsibility in function of occupational safety and sustainable development, 14. International symposia on guality, Rovini, 2122.3.2013.					<u>https://</u> Taradi	<u>/bib.irb.hr/datoteka/623168.S</u> _HDMK-14.pdf	ijakovic_Savic_Nikolic_	
	M. A. Omazić et al., A collection of case studies of corporate social responsibility, Zagreb, 2012.					http://www.hup.hr/EasyEdit/UserFiles/Granske_udrug bal%20Compact/zbirka_studija_slucaja_DOP.pdf		<u>es/Granske_udruge/Glo</u> <u>caja_DOP.pdf</u>	
2.12. Optional literature (at the time of the submission of the study programme proposal)	Available scientific literature	e on the subjec	ct of c	corporate social r	respo	onsibility			
2.13. Methods of monitoring quality that	Survey on the faculty and U	Iniversity level.							
ensure acquisition of exit	Analysis predicted by system	ms for insurand	nce of	institution quality	у.				
competences	Analysis predicted by systems for insurance quality from authorized University office.								

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Define and explain the concepts of corporate social responsibility.	1st colloquium, seminar, written and oral exam
2	Know the guidelines of the strategy of sustainable development in Croatia.	1st colloquium, seminar, written and oral exam
3	How to use regulations and scientific-technical literature in the field of corporate social	2nd colloquium, seminar, written and oral exam
	responsibility.	
4	Understand and advocate for consistent application of the principles of corporate social	2nd colloquium, seminar, written and oral exam
	responsibility.	
5	Design and implement environmentally, economically and socially responsible business	2nd colloquium, seminar, written and oral exam
	practices and socially responsible investment community.	

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Assoc.Prof. Zoran Glavas	š, PhD	1.6. Year of study		2	
1.2. Name of the course	ANALYSIS OF CASTING	DEFECTS	1.7. Credit value (ECT)	S)	4	
1.3. Associate teachers	-		1.8. Type of instruction L+S+E+e-learning	(number of hours )	30+0+15+0	
1.4. Study programme (undergraduate, graduate, integrated)	graduate		1.9. Expected enrolme	nt in the course	10	
1.5. Status of the course	elective		1.10. Level of use of e level), percentage course on line (20	<ul> <li>learning (1, 2, 3 of instruction in th % maximum)</li> </ul>	e 1., 5%	
2. COURSE DESCRIPTION						
2.1. Course objectives	The ability to recognize a The ability to find the pos The ability to define meas	nd analyse c sible causes sures to prev	asting defects. of casting defects. ent the formation of casting	g defects.		
2.2. Enrolment requirements and required entry competences for the course						
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoret Plan and manage the cor Plan the production and o	cal knowledg npetences of casting proce	e in engineering practice. analysis and synthesis. sses of ferrous and non-fe	rrous metals.		
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Recognize the type of ca Describe the mechanism Analyse the causes of the Propose measures to pre Evaluate the success of r	sting defect. of the forma of formation o event the forn measures imp	ion of casting defect. f casting defect. nation of defect in casting. plemented to prevent the fo	ormation of defect	in casting.	
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Classif and detection of casting surface. (4); Incomplete of EXCERCISES (15): Exar results, determining the of	ication of cas defects. (4 casting. (1); lingles of cast ause of the c	ting defects. (1); The influ ; Metallic projections on acorrect dimensions or sha ngs with different defects. lefect and suggestions for	the castings. (2) ape. (1); Inclusion (3); Metallograph avoiding the form	n casting properties. (1); Me ; Cavities. (5); Discontinuiti s or structural anomalies. (10 c analysis of defects - interp ation of casting defects. (12)	thods for analysis es. (1); Defective ); retation of the
2.6. Type of instruction	<ul> <li>lectures</li> <li>seminars and workshow</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>	ops	<ul> <li>independent study</li> <li>multimedia and the international international international international internation (other)</li> </ul>	2.7. Comments:		
2.8. Student responsibilities	Conditions for signature:	Students mu	st attend the lectures and	exercises (> 70 %	).	
2.9. Screening of student's work	Class attendance	0.5	Research		Practical training	
(specify the proportion of ECTS	Experimental work		Report			
credits for each activity so that the	Essay		Seminar essay		(Otherdescribe)	
total number of CTS credits is equal	Tests	3.5	Oral exam		(Other—describe)	

to the credit value of the course)):	Written exam	Project			(Other-describe)				
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Exam of the course: Through continuous monitoring - student needs to pass 2 colloquiums. If the student has passed all colloquiums, the final score is determined as the average score of the colloquiums. Through the final exam: written exam for students who have not passed the exam through continuous monitoring or are not satisfied with the success that are achieved through continuous monitoring or have not decided on this method of examination. Conditions for access to the exam: -								
0.44. Described literature (overlights of	Title Copies the libr		Number of copies at the library	Availability via other media		dia			
2.11. Required literature (available at the library and via other media)	Z. Glavaš, Analiza greša the lectures placed or Metallurgy, Faculty of Me	aka na odljevcima, text of website of Faculty of tallurgy, Sisak, 2009.		<u>https://ww</u> <u>sveucilisn</u> <u>studija/an</u>	w.simet.unizg.hr/nastava/preda i-studij-metalurgija/2-godina-dip aliza-gresaka-na-odljevcima/vie	avanja/diplomski- plomskog- ew			
2.12. Optional literature (at the time of the submission of the study programme proposal)	<ul> <li>S. Hasse, Pogrješke na odljevcima, Croatian Foundry Association, Zagreb, 2003.</li> <li>M. T. Rowley, International Atlas of Casting Defects, AFS, 1990.</li> <li>, Aluminium Permanent Mold Handbook, AFS, Des Plaines, Illinois, 2001.</li> <li>W. G. Walkington, Die Casting Defects – Causes and Solutions, NADCA, USA, 2003.</li> </ul>								
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Survey on the level of faculty and university. Analysis predicted in the quality assurance system of institution. Analysis predicted in the quality assurance system and authorized office of the university.								

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of		
number		learning outcomes		
1	Recognize the type of casting defect.	1st and 2nd colloquium, written exam		
2	Describe the mechanism of the formation of casting defect.	1st and 2nd colloquium, written exam		
3	Analyse the causes of the formation of casting defect.	1st and 2nd colloquium, written exam		
4	Propose measures to prevent the formation of defect in casting.	1st and 2nd colloquium, written exam		
5	Evaluate the success of measures implemented to prevent the formation of defect in casting.	1st and 2nd colloquium, written exam		

1. COURSE DECRIPTION - GENERAL	SURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:					
1.1. Course teacher	Assoc.Prof. Zdenka Zovko Brodarac, PhD	1.6. Year of study	2			
1.2. Name of the course	MODERN TECHNOLOGIES OF METAL CASTING	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION	•	•				
2.1. Course objectives	Introduction to the procedures and the ch Introduction to modern concepts of indivi Awareness of the possibility of optimizing	naracteristics of the methods of casting metal dual and mass production of castings. g the castings production using computer mod	castings. Ieling			
2.2. Enrolment requirements and required entry competences for the course	-					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Analyse the development and application of new technologies Plan the production and casting processes of ferrous and non-ferrous metals					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Analyse of the casting and solidification permanent molds. Relate microstructure and performance of Select casting technology according to the select casting technology according t	process and on the basis of ferrous and non-fe characteristics of castings. The required properties of the castings.	errous metals in green sand and			
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction to the course content and modes of knowledge acquisition. (1) Review of modern technologies of metal casting The requirements and expectations placed in front of foundry industry, i.e. industry of transport vehicles - automobile, railway, shipbuilding. (2) Production process and casting tolerances. (4) The concept of near net shape casting. An integrated approach to the production of metal castings. The concept of simultaneous engineering. (6) Improving the quality of the castings through the development of technology of smelting and casting.Casting construction, optimization of cast components with FEM and CAD systems. (4) Optimization of molding processes, elimination method "trial and error". The use of computer modeling in order to optimize of casting and feeding processes. The concept of virtual production castings. Rapid prototyping: proceedings of FDM, SLS, SLA, DSPC, Modern technologies of metal casting: Replicast, Low-pressure casting in sand molds. Modern technologies of metal casting: Cosworth, FM, CLA procedures. (7) Modern technologies of metals castings: Direct and indirect "squeeze casting" process. Casting metal in semi-solid state - Devention and costing costing costing casting: Direct and indirect "squeeze casting" process.					

	EXERCISES (15): Visit to the relevant economic entities.							
2.6. Type of instruction	<ul> <li>lectures</li> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>		independent study2.multimedia and the internetlaboratorywork with the mentor(other)		2.7. 0	Comments:		
2.8. Student responsibilities	Attending the classes. 1. colloquium.							
	Class attendance	1	Research		Pract	ical training		
2.9. Screening of student's work (specify the proportion of ECTS	Experimental work		Report					
credits for each activity so that the total number of CTS credits is equal to the credit value of the course));	Essay		Seminar essay		(Otherdescribe)			
	Tests	3	Oral exam		(Other-describe)			
	Written exam		Project			(Other-describe)		
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Attending the classes. One test through continuous	monitoring	or final examination (writte	en and oral).				
			Title			Number of copies at the library	Av o	ailability via other media
2.11. Required literature (available at	Metals Handbook, Volume 15, CASTING, ASM International, Metals Park, Ohio, 1988.					1		
the library and via other media)	Aluminum ans Aluminum Alloys, ASM Speciality Handbook, ASM International, ed. J. R. Davis, Materials Park, Ohio, USA, 2002.					1		
	Metals Handbook, Volume 1, Properties and Selection: Irons and Steels, ASM International, Metals Park, Ohio, 1978.					1		
2.12. Optional literature (at the time of the submission of the study	-							
programme proposal)								
2.13. Methods of monitoring quality that	Survey at the Faculty and Ur	niversity lev	/el.					
ensure acquisition of exit	Analysis provided the quality assurance system of the institution.							
competences	Analysis provided the quality assurance system and authorized Office of the University							

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Analyse of the casting and solidification process and on the basis of ferrous and non-ferrous	1st colloquium, auditory exercises, written and oral exam
	metals in green sand and permanent molds.	
2	Relate microstructure and performance characteristics of castings.	1st colloquium, auditory exercises, written and oral exam
3	Select casting technology according to the required properties of the castings.	Independent work, written and oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Assoc.Prof. Vladimir Grozdanić, PhD	1.6. Year of study	2			
1.2. Name of the course	SOLIDIFICATION SIMULATION	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	compulsory	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION						
2.1. Course objectives	Introduction to mathematical modelling of s Using of numerical methods for solidification	solidification of castings of different geometry. on simulation by means of computer.				
2.2. Enrolment requirements and required entry competences for the course	-					
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge</li> <li>Recognize and apply scientific principle</li> <li>Plan and manage metallurgical process</li> <li>Design and apply the modelling of meta</li> </ol>	e in engineering practice. es important in the fild of metallurgy. ses. allurgical and other processes.				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Design methods of castings feeding wit</li> <li>Application of numerical methods of and</li> <li>Prediction of possibility of appearance of</li> <li>Formulation of model of solidification in</li> </ol>	h simulations, MPR programme. alysis of pouring and feeding of castings. of defects and measures of prevention. chosen programe language.				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ul> <li>LECTURES (30) AND EXERCISES (15):</li> <li>1. Introduction to mathematical modelling.</li> <li>2. Review of computer software for solidification simulation of castings.</li> <li>3. Computers and programme languages, numerical methods.</li> <li>4. Solidification simulation of castings of different geometrical complexity.</li> <li>5. Heat during solidification, energy balance, energy transfer for cast steel.</li> <li>6. Methods of modelling in foundry.</li> <li>7. Explicit and implicit methods of finite difference applied to periodical cooling and heating of slab.</li> <li>8. Implicit alternating direction method applied to solidification of L, T, H castings, blank gears, railway wheel.</li> <li>9. Saulyev explicit method applied to L-shaped castings and comparison with ADI method.</li> <li>10. 3D finite element method, derivation of Brians method in the case of L problem.</li> <li>11. Methods of prediction of casting defects on the basis of time of solidification and temperature gradient.</li> <li>12. Modelling of flow. Solve Navier-Stokes equation and application to flow of melt in ranning system.</li> <li>13. Heat and mass transfer.</li> <li>14. Principles of modelling and microstructure of castings.</li> <li>15. Computer optimization of design. Location and prevention casting defects with different ticknes of wall, inclination with</li> </ul>					

	☐ lectures		independent study			2.7. Comments:		
2.6. Type of instruction	serimats and workshops     exercises     online in entirety     mixed e-learning     field work		<ul> <li>multimedia and the internet</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>					
2.8. Student responsibilities	Conditions for signature: atte Conditions for taking:-	ndance to I	ectures and exercises min. 70%	6.				
2.9. Screening	Class attendance	0.4	Research		Pra	actical training		
of student's work (specify the	Experimental work		Report					
proportion of ECTS credits for each	Essay		Seminar essay			(Otherdescribe)	)	
activity so that the total number of	Tests	1.6	Oral exam	1.2		(Other-describe	e)	
value of the course)):	Written exam	Written exam 0.8 Project				(Other-describe	e)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Written exam: 50% Oral exam: 50%							
	Title Number of copies at the library					Avai oth	ilability via ner media	
2.11. Required literature (available at the	R. I. Esman, N. P. Žmakin, L. 1977.	. I. Šub, Ra	sčoti processov litja, Višejšaja š	škola, Minsk,		1		
library and via other media)	G. D. Smith, Numerical Solution of Partial Differential Equations, University Press, 1 Oxford, 1974.							
	B. Carnahan, H. A. Luther, J. O. Wilkes, Applied Numerical Methods, John Willey, New 1 York, 1969.							
2.12. Optional literature (at the time of the submission of the study programme proposal)	P. R. Sham, P. N. Hansen, N House, CIATF, Zurich, 1984.	lumerical S	imulation and Modelling of Cas	ting and Solid	ificat	ion Processes for F	oundr	y and Cast-
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Internal students survey, analysis anticipated by system of ensure quality. Survey of the level of faculty and University. Analysis anticipated by system of ensure quality of institution. Analysis anticipated by system of ensure quality and entitled office of University.							

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Design methods of castings feeding with simulations, MPR programme.	Colloquium
2	Application of numerical methods of analysis of pouring and feeding of castings.	Written exam
3	Prediction of possibility of appearance of defects and measures of prevention.	Oral exam
4	Formulation of model of solidification in chosen programe language.	Colloquium, written exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Assoc.Prof. Ljerka Slokar, PhD	1.6. Year of study	2			
1.2. Name of the course	POWDER METALLURGY AND SINTER MATERIALS	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION						
2.1. Course objectives	Presenting the historical development of production of powders and connectivity of technology to properties. Introduction to methods of production of sinterable metal powders, methods for their compaction and sintering, as well as studies of obtained compacts. Developing the ability to analyze and synthesize of the profession basic knowledge. The ability to apply acquired knowledge in practice.					
2.2. Enrolment requirements and required entry competences for the course						
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Analyse the development and application of new technologies. Design the properties of metallic materials					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Describe the methods of production of sinterable metal powders, and methods of compacting and sintering. Analyze the microstructure and properties of powders and compacts. Define the best technology to obtain certain sinterable powder.					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction. Brief history, definitions, applications and advantages of powder metallurgy. (2) Production of powder. Techniques: mechanical, electrolytic, chemical, atomization. Producing specific and ultradispersive metal powders. (4) Powder characterization: sampling, determining the particle size and particle size distribution, determining the shape of particles, the specific surface area, interparticulate friction, chemical characterization. (4) Preparation of powders and compacting: mixing powders, friction, phenomenology and theoretical foundations of compaction, conventional compacting, the influence of the characteristics of powder, technologies. (6) Sintered materials: definition, sintering theory, the impact of compaction on sintering, the sintering effect on the properties of sintered powder mixture, the atmospheres and the sintering furnaces. (4) Methods to achieve full density: the basics, benefits, procedures. (2) Characterization of the compact: microstructure, mechanical properties, surface, physical properties. (4) Application: powder-technology-properties dependence. (4) LABORATORY EXERCISES (15): Compacting powders. Characterization of powders and compacts. Compacting-sintering-					

	properties dependence	e.						
	Iectures				study		2.7.	Comments:
	seminars and workshops			multimedia and the internet				
2.6. Type of				⊠ laboratory				
Instruction				work with the	mentor			
				(other)				
2.8. Student responsibilities	Attending lectures and	l auditory ex	xercises (m	iin. 70%), labora	itory work	<pre>c performed</pre>	d 100% and comn	nitted to paper.
	Class attendance	0.5	Research	า		Р	ractical training	
0.0. Concerning of student's work (on seif.)	Experimental work		Report					
the proportion of ECTS credits for	Essay		Seminar	essay		de	(Other escribe)	
of CTS credits is equal to the credit	Tests	1.5	Oral exa	Oral exam		de	(Other— escribe)	
	Written exam	1.0	Project			de	(Other— escribe)	
2.10. Grading and evaluation of student	The final grade is dete	rmined by s	score of thr	ee colloquiums	and asse	ssments of	f written and oral e	exams respectively.
work over the course of instruction								
and at a final exam				Number of		r		
	T	itle		at the lib	orary		Availability v	via other media
	Lj. Slokar, Metalurgija	praha i sint	ter			https://wv	vw.simet.unizg.hr/	<u>'nastava/predavanja/diploms</u>
	materijali, Metalurški fa	akultet, Sisa	ak, 2015.			ki-sveucilisni-studij-metalurgija/2-godina-diplomsko		gija/2-godina-diplomskog-
						studija/M	<u>ETALURGIJA%2(</u>	PRAHA%201%20SINTER%
	P. M. Gormon, Powdo	r Motallura	v Science				RIJALI.pul/view	20
	Metal Powder Industri	a wetanury	y Science,				,	
	Federation.Princeton.	New Yerse	v. 1984.					
2.11. Required literature (available at the	D. Schulze, Powders a	and Bulk Sc	olids,				(	CD
library and via other media)	Springer, Berlin, 2008.		·					
	G. S. Upadhyaya, Pov	vder Metallu	urgy				(	CD
	Technology, Cambridg	ge Internatio	onal					
	Science Publishing, Ca	ambridge, 2	2002.					
	F. Thümmler, R. Ober	acker, An Ir	ntroduction				(	CD
	to Powder Metallurgy, The Institute of							
	Materials, London, 199	93.						
2.12. Optional literature (at the time of	ASM Handbook Volum	ne 7. Powde	er Metal Te	chnologies and	Application	ons, ASM I	nternational. 1998	3.
the submission of the study	B. D. Fahlan, Materials	s Chemistry	, Springer,	London, 2011.	FF	, · · <b>-</b> ···		
programme proposal)	B. S. Mitchell, An Introduction to Materials Engineering and Science, John Wiley & Sons, Inc. Hoboken, New Jersey, 2004.							

2.13. Methods of monitoring quality that	Survey on the level of the Faculty and University.
ensure acquisition of exit	Analysis provided the quality assurance system of the institution.
competences	Analysis provided the quality assurance system and authorized Office of the University.

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Describe the methods of production of sinterable metal powders, and methods of compacting and sintering.	1st colloquium, written and oral exam
2	Analyze the microstructure and properties of powders and compacts.	2nd colloquium, laboratory exercises, written and oral exam
3	Define the best technology to obtain certain sinterable powder.	3rd colloquium, laboratory exercises, written and oral exam
4	Assess the economic parameters to justify the selected technology.	3rd colloquium, written and oral exam

1. COURSE DECRIPTION - GENERAL IN	ISVU CODE:					
1.1. Course teacher	Assoc.Prof. Ljerka Slokar, PhD	1.6. Year of study	2			
1.2. Name of the course	ADVANCED METALLIC MATERIALS	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION	•		•			
2.1. Course objectives	Achieve an understanding of the basic Provide an overview of contemporary t	laws in the process of obtaining the a rends in their development.	advanced metallic materials.			
2.2. Enrolment requirements and required entry competences for the course						
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ul> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Design the properties of metallic materials.</li> <li>Compare the procedures of material treatment with microstructure and useful properties.</li> </ul>					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Explain the basic principles and methods for the preparation of advanced metallic materials. Evaluate, compare and select the advanced metallic materials for specific application conditions. Describe the types and examine metal composites and materials with shape memory effect.					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Magnetic materials. Diamagnetism, paramagnetism, ferromagnetism, antifero- and ferimagnetizam. The magnetic field, strength of the field, the flux density. Magnetostriction and magnetization. The soft and hard magnetic materials, use. (4) Metal foams. Porous metals and metal foams. The filling process. Methods of production. Mechanical and physical properties. The use of metal foams. (4) Amorphous metals (metallic glasses). The crystal and amorphous state. "Structure" and properties. Obtaining amorphous metals. Application. (4) Metal composites. Basic principles and definitions. The composites with particles, dispersion and filamentary materials. The laminar composites. The influence of the matrix. Reinforced plastic and metal matrix. The hybrid composites. (4) Materials with shape memory effect. Introduction. Martensitic transformation by twins formation and deformation. Pseudoelasticity and superelasticity. Types and characterization. Producing and training of the materials. Use. (4) Nanostructured materials. Nanoparticles and nano powders. Nanocapsules and nanoporous materials. Nano fibers, fullerenes and nanowires. Nano carbon tubes. (6) The development of metallic materials. Tendencies of development. Ecomaterials. Materials for secondary sources of energy. Materials with extreme physical and mechanical properties. (4)					

	LABORATORY EXERCISES (15): Determination of physical-chemical and mechanical properties of selected advanced metallic materials. The characterization of these materials by optical and electron microscopy						
2.6. Type of instruction	Interaine materials. The characterization of these materials by optical and electron in         Image: seminars and workshops         exercises         online in entirety         mixed e-learning         field work			2.7. Comments:	2.7. Comments:		
2.8. Student responsibilities	Attending classes (min. 70%), la	aboratory wor	k perforr	ned and comm	itted to pa	iper.	
2.9. Screening of student's work (specify	Class attendance	0.5	Rese	arch		Practical training	
the proportion of ECTS credits for	Experimental work		Repo	ort			
each activity so that the total number	Essay		Semi	nar essay		(Otherdescribe)	
of CTS credits is equal to the credit	Tests	1.5	Oral	exam	1.0	(Other-describe)	
value of the course)):	Written exam	1.0	Proje	ct		(Other-describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	The final grade is determined by score of three colloquiums and assessments of written and oral exams respectively.				ectively.		
	Title			Number of c at the libr	opies ary	Availability via other	media
	M. Oruč, R. Sunulahpašić, Suvremeni metalni materijali, Fakultet za metalurgiju i materijale, Zenica, 2005.			2			
2.11. Required literature (available at the library and via other media)	T. Filetin, I. Kramer, G. Marić: "Metalne pjene", Hrvatsko društvo za materijale i tribologiju, Zagreb, 2003.				<u>+</u> -	nttp://titan.fsb.hr/~tfiletin/pdf/met razvoj.pdf	alne_pjene_tehn
	R. E. Smallman, R. J. Bishop, Modern Physical Metallurgy and Materials Engineering, Oxford, 1999.			1	<u> </u> 1	http://www.sim.utcluj.ro/stm/dow nPhysicalMetallurgy.pdf	nload/Alba/Mode
2.12. Optional literature (at the time of the submission of the study programme proposal)	Lj. Slokar, Metalurgija praha i sinter materijali, Metalurški fakultet, Sisak, 2015. W. D. Callister, Materials Science and Engineering, J. Wiley & sons, New York, 1994.						
2.13. Methods of monitoring quality that	Survey on the level of the Faculty and University. Analysis provided the quality assurance system of the institution. Analysis provided the quality assurance system and authorized Office of the University.						

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Explain the basic principles and methods for the preparation of advanced metallic materials.	1st, 2nd and 3rd colloquium, laboratory exercises, written
		and oral exam
2	Evaluate, compare and select the advanced metallic materials for specific application	1st, 2nd and 3rd colloquium, laboratory exercises, written
	conditions.	and oral exam
3	Describe the types and examine metal composites and materials with shape memory effect.	2nd and 3rd colloquium, laboratory exercises, written and
		oral exam
4	Analyze the physical-chemical and mechanical properties of advanced metallic materials.	3rd colloquium, laboratory exercises, written and oral
		exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:					
1.1. Course teacher	Assoc.Prof. Stjepan Kožuh, PhD	1.6. Year of study	2		
1.2. Name of the course	MODERN TOOL STEELS	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+15+0+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10		
1.5. Status of the course	elective	<ul> <li>1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)</li> </ul>	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	Deepening knowledge in field related to tool steels. Understanding the basic principles of r	o the properties, microstructure and the naking, classifying and heat treatment	e application of modern of tool steel.		
2.2. Enrolment requirements and required entry competences for the course	-				
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ul> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Analyse the development and application of new technologies.</li> </ul>				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Analyze the properties of tool steels. Predict the type of tool steel for a particular practical application. Compare tool steels from the standpoint of wear and usability.				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Week 1: Basic terms and definitions. (2) Week 2: The work principle of the tool. The types of tools. Processing and properties of tool steels. The influence of alloying elements. (2) Week 3: Overview of making processes for tool steels. Special processes of remelting. (2) Week 4: Non-alloy (carbon) tool steels. (2) Weeks 5 and 6: Tool steels for cold work. Low alloyed tool steels (W-V steels, W-Cr (Si) -V steels, Cr steels, Mn-Cr-V and Mn- Cr-W steels). (4) Week 7: High alloyed tool steels (High-Carbon ledeburite steels, martensitic stainless steels). (2) Week 8 and 9: Tool steels for hot work (low alloy steels, high-alloyed Cr-Mo steels, high-alloyed W-Cr-V steel). High-speed steels. (4) Week 10: Maraging tool steels. Precipitation hardened tool steels. (2) Weeks 11 and 12: Sintered tool steels (steels: ISOMATRIX PM, ASP, CPM, martensitic steels, precipitation hardened steels). (4) Week 13: Other materials in the tool application. (2)				

	SEMINAR (15): The selection of topics and seminar work in writing form by a mentor system (10 hours). Preparation and presentation of the seminar and discussions related to the topic of the present paper (5 hours).						
2.6. Type of instruction	<ul> <li>lectures</li> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>	DS independent study multimedia and the laboratory work with the ment (other)			/ e interne itor	2.7. Comments:	
2.8. Student responsibilities	Students must attend over 70%	6 of lectur	es and are	required	to com	plete a seminar in writing form and orally prese	ent.
2.9. Screening of student's work (specify	Class attendance	0.5	Researc	ch		Practical training	
the proportion of ECTS credits for	Experimental work		Report				
each activity so that the total number	Essay		Semina	r essay	1.0	(Otherdescribe)	
of CTS credits is equal to the credit	Tests	2.5	Oral exa	am		(Other—describe)	
value of the course) <i>):</i>	Written exam		Project			(Other—describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	<ul> <li>evaluation of students activities in course,</li> <li>evaluation of written examination (two colloquiums) through continuous monitoring or final examination (written and c</li> <li>evaluation of seminar paper and presentation.</li> </ul>					and oral),	
	Title			Numl copies libr	per of at the ary	of the Availability via other media	
	M. Gojić, Metalurgija čelika, Sveučilište u Zagrebu Metalurški fakultet, Sisak, 2006.			1	5		
2.11. Required literature (available at the library and via other media)	M. Novosel, F. Cajner, D. Krumes, Alatni materijali, Sveučilište J. J. Strossmayera u Osijeku, Strojarski fakultet u Slavonskom Brodu, Slavonski Brod, 1996.				3		
	S. Kožuh, Specijalni čelici-skrip	ota, Sveuč	ćilište u			https://www.simet.unizg.hr/nastava/predavar	nja/diplomski-
	Zagrebu Metalurški fakultet, Si	sak, 2010				sveucilisni-studij-metalurgija/2-godina-diplom	nskog-
						studija/specijalni-celici/view	
	J. Pirš, Toplinska obrada metala, Tehnički fakultet Rijeka, Rijeka, 1992.			(	3		
	ASM Handbook.pdf					Electronic form	
2.12. Optional literature (at the time of the submission of the study programme proposal)	<ul> <li>V. Đukić, Alatni čelici, Naučna knjiga, Beograd, 1990.</li> <li>B. Jocić, Steels and cast Irons, BIO-TOP, Dobja Vas, 2008.</li> <li>C. R. Brooks, Principles of the Heat Treatment of Plain Carbon and Low Alloy Steels, ASM International, Materials Park, 1996.</li> <li>Scientific and professional papers in refereed journals and conference proceedings.</li> </ul>				s Park, 1996.		
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Questionnaire at Faculty and University.         The analyzes provided a system of quality assurance institutions.         The analyzes provided a system of quality and authorized office of University.						

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Analyze the properties of tool steels.	1st colloquium, written and oral exam
2	Predict the type of tool steel for a particular practical application.	2nd colloquium, written and oral exam, seminar paper
3	Compare tool steels from the standpoint of wear and usability.	1st and 2nd colloquium, written and oral exam
4	Differentiate a new types of tool steels.	2nd colloquium, written and oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Full Prof. Ladislav Lazić, PhD	1.6. Year of study	2			
1.2. Name of the course	METAL FORMING MACHINES	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	Assist.Prof. Ivan Jandrlić, PhD	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION						
2.1. Course objectives	<ol> <li>Knowing the names, division, chara</li> <li>Knowing the conditions that determ</li> <li>Knowing the type of transport mea</li> </ol>	acteristics and functions of machines and equip nine the shape, size and material of each elements ons, their function and basis of their design.	oments for plastic forming of metals. ent of a machine or device.			
2.2. Enrolment requirements and required entry competences for the course	Passed the exams of course Theory of metal forming.					
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Suggest new and improved technical and technological solutions.</li> <li>Design professional elaborates and professional projects in metallurgy.</li> <li>Formulate and suggest measures for increasing energy efficiency.</li> </ol>					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Analyze the characteristics of machines and equipment for plastic forming of metal materials in solving optimization problems of existing processes.</li> <li>Judge the characteristics of machines and equipment for plastic forming of metal materials in solving problems related to the use of new materials and technologies.</li> <li>Analyze and integrate metal forming processes and propose solutions to optimize metallurgical processes.</li> </ol>					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ul> <li>LECTURES (30):</li> <li>1. Systematization of machines and equipments for plastic forming of the metals. (2)</li> <li>2. Classification according to the cold and hot forming processes. (2)</li> <li>3. Hammer machines: air, steam and drop hammer, gag (eccentric) and friction presses, hydraulic presses. (6)</li> <li>4. Plastic forming tools. (4)</li> <li>1st colloquium - Submission of the program for the calculation of the power needed for the operation of a machine or device</li> <li>5. Strands of rolls: division according to the temperature rolling, rolling method and arrangement of machines and equipments. (4)</li> <li>6. Rolling machines: rolling stand parts (rolls, roll bearings, roll adjusting equipments, drive of rolling mills). (6)</li> <li>7. Conveying devices: Analysis of the material flow of internal and external transport, Auxiliary load-holding equipment, Constructive parts of transport systems, Drive of transport devices, Conveyors, Cranes, Winches. (6)</li> </ul>					

	2nd colloquium - Submission of the constructional calculation and drawing of the machine part for plastic forming or transport device								
	EXERCISES (15): The unders projects are selected so that the	EXERCISES (15): The understanding of the material exposed in lectures is facilitated by solving the given projects. The projects are selected so that they expand the presented theory and illustrate the application of theory to real problems.							
			independent study		2.7. Comments:				
2.6. Type of instruction	<ul> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> <li>mixed e-learning</li> <li>(other)</li> </ul>								
2.8. Student responsibilities	Attendance on Lectures and E	xercises >	70 %.						
2.9. Screening of student's work (specify	Class attendance	0.4	Research		Practical training				
the proportion of ECTS credits for	Experimental work		Report						
each activity so that the total number	Essay		Seminar essay		(Otherdescribe)				
of CTS credits is equal to the credit	Tests	1.0	Oral exam	1.6	(Other-describe	)			
value of the course)):	Written exam		Project	1.0	(Other-describe)				
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Projects – 30% Class attendance – 5% Written exam – 30% Oral exam – 35%								
	Title Number of copies at the library other						via a		
2.11. Required literature (available at the library and via other media)	H. Lippmann, O. Mahrenholtz, Springer Verlag, Berlin, 1967.	, 1							
	L. Lazić, ELEMENTI STROJE 2(075.8), Sisak, 2001.	13							
2.12. Optional literature (at the time of the submission of the study programme proposal)	E. G. Thomas, C. T. Yang, S. Kobayashi, Mechanics of Plastic Deformation in Metals Processing, Macmillan, New York, 1965. W. Beitz, K. H. Kuttner, Dubbel-Taschenbuch fur den Machinenbau, Springer Verlag, Berlin, 1986								
2.13. Methods of monitoring quality that	Survey at the level of faculty a	nd Universi	ty.						
ensure acquisition of exit	Analyses provided in the system of quality assurance of the institution.								
competences	Analyses provided in the system of quality assurance and an authorized office of the University.								

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Analyze the characteristics of machines and equipment for plastic forming of metal materials in	1st colloquium, oral exam
	solving optimization problems of existing processes.	
2	Judge the characteristics of machines and equipment for plastic forming of metal materials in	1st colloquium, oral exam
	solving problems related to the use of new materials and technologies.	
3	Analyze and integrate metal forming processes and propose solutions to optimize metallurgical	2nd colloquium, oral exam
	processes.	
4	Propose optimal choice and use of transport means.	2nd colloquium, project, oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Full Prof. Stoja Rešković, PhD	1.6. Year of study	2			
1.2. Name of the course	ROLL FORMING OF METALS	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	Tin Brlić, mag.ing.met.	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+15+0+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	2., 10%			
2. COURSE DESCRIPTION						
2.1. Course objectives	<ol> <li>To introduce students with the basic the students with the basic the students with the technor.</li> <li>To introduce students with the basic of the basic of the students with the basic of the students.</li> </ol>	heoretical knowledge formatting rollir logy of rolling. haracteristics of industrial plants.	ng.			
2.2. Enrolment requirements and required entry competences for the course	Passed exam from Theory of metal forming					
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Analyse the development and application of new technologies.</li> <li>Plan and manage metallurgical processes.</li> <li>Analyse and combine metal forming processes.</li> </ol>					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Understand today's situation and development trends of the procedures for processing of metallic materials by forming.</li> <li>To calculate the parameters of rolling process.</li> <li>Compare the various technologies of shaping by rolling.</li> <li>Monitor the development and application of new technologies.</li> </ol>					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ul> <li>LECTURES (30) AND SEMINAR (15): <ol> <li>The definition of the rolling process, the elements of the deformation zone, material griped by rolls. 2</li> <li>Computational parameters during rolling, deformation, strain rate. 2</li> <li>Seminar: Calculation of deformation zone elements. 2</li> <li>Acceleration (overtaking). 2</li> <li>Friction in the rolling process. Elements that affect the size of friction. 2</li> <li>Spread, the factors that influence the spread, calculation of spreading during rolling. 2</li> <li>Normal contact and tangential stresses. 2</li> <li>Metal pressure on the rollers: Distribution of specific pressure, differential equation of specific pressure. Middle specific pressure by Celikov and Ekelend. 2</li> <li>Seminar: Calculation of constants of continuous production lines for the production of the strips. Technology of rolling strips on continuous line (rolling pre-strip on the block rolling mill, rolling strips on the final line). 2</li> <li>I colloquium, Chapters 1-10</li> <li>Elements of the rolling line. Types of rolling mills in relation to the number of rollers, space distribution of rolling mills. 2</li> </ol></li></ul>					

	<ul> <li>and small profiles. 4</li> <li>14. Rolling wire and manufacturing of special profiles. 4</li> <li>15. Rolling flat profiles: Hot rolling of thick and medium plates. Rolling thin sheets. Cold rolling of sheets and strips. Rolling sheets and strips of non-ferrous metals. 4</li> <li>16. Specifics of rolling metals. 2</li> <li>17. 4. Seminar: Assignments of calibration of rollers: Concept and distribution of calibres. Calibres system. Calibration of rollers for rolling profiles. 2</li> <li>18. 5. Seminar: Calibration rollers - flat profiles. 2</li> <li>19. 6. Seminar: Checking calibration in practice. 5</li> <li>20. IL Colloquium, chapters 12-19</li> </ul>							
	⊠ lectures		🛛 independer	nt studv			2.7. Comments:	
2.6. Type of instruction	Seminars and workshops       Independent study       Entrementer         exercises       multimedia and the internet         online in entirety       laboratory         mixed e-learning       (other)							
2.8. Student responsibilities	Attendance at lectures min. 70%, individual work on all exercises and preparation and submission of reports from field of colloquium before writing the colloquium or the written exam.							
2.9 Screening of student's work (specify	Class attendance		Research	0	.2	Practio	cal training	
the proportion of ECTS credits for	Experimental work	0.2	Report	0	.2			
each activity so that the total	Essay		Seminar essay 0.2		.2		(Otherdescribe)	
number of CTS credits is equal to	Tests	2.0	Oral exam	1	.0		(Other-describe)	
the credit value of the course)):	Written exam		Project	0	.2		(Other-describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	The presence and activity of stud Students score participation on p Score on written colloquium troug Score of seminar paper.	lents on clas rojects. gh continuou	sses during the c us monitoring (or	vlasses ar	e evaluate	oral exam	ı.	
				Numb	er of			
	Title			copies libra	at the ary		Availability via other r	nedia
	Čaušević, Obrada metala valjanjem, Veselin Masleša, Sarajevo, 1983.			2				
2.11. Required literature (available at the library and via other media)	S. Rešković, Teorija oblikovanja u Zagrebu, Metalurški fakultet, Si lessons.	deformiranje isak 2014., j	em, Sveučilište peer reviewed			attps://www ar/nastava sveucilisn liplomsko itudija/S% DBLIKOV/	w.simet.unizg. /predavanja/diplomski i-studij-metalurgija/1-g g- 20Reskovic%20TEOR ANJA%20DEFORMIR/	odina- IJA%20 NJEM.pdf/view
	S. Rešković, Tehnologija oblikova nastavna građa, Sisak, 2011.	anja deform	iranjem-	10				
2.12. Optional literature (at the time of	R. Križanić, Valjanje metala, Fac	ulty of Meta	llurgy, Sisak, 198	35. (intern	nal script).			

the submission of the study	Professional journals, articles from this area.
programme proposal)	
2.13. Methods of monitoring quality that	Survey on the level of faculty and University.
ensure acquisition of exit	Analyses provided by quality assurance system of the institution.
competences	Analyses provided by quality assurance system and authorized office of the University

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of		
number		learning outcomes		
1	Understand today's situation and development trends of the procedures for processing of	1st colloquium, seminar paper, written and oral exam		
	metallic materials by forming.			
2	To calculate the parameters of rolling process.	1st colloquium, seminar paper, written and oral exam		
3	Compare the various technologies of shaping by rolling.	seminar paper, independent task, oral exam		
4	Monitor the development and application of new technologies.	2nd colloquium, seminar paper, oral exam		

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:							
1.1. Course teacher	Full Prof. Stoja Rešković, PhD	1.6. Year of study	2				
1.2. Name of the course	METAL TUBE AND PROFILE FORMING	1.7. Credit value (ECTS)	4				
1.3. Associate teachers	Assist.Prof. Ivan Jandrlić, PhD	st.Prof. Ivan Jandrlić, PhD 1.8. Type of instruction (number of hours 30+15+0+0 L+S+E+e-learning)					
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course 10					
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	2., 10%				
2. COURSE DESCRIPTION							
2.1. Course objectives	<ol> <li>To introduce students to the importance and the trend of production of pipes and profiles in the world.</li> <li>Introduce students with the technologies of production of pipes and profiles</li> <li>Students become familiar with the basic characteristics of industrial plants</li> <li>Acquired knowledge will enable the student to recognize and solve problems in the process of production of pipes and profiles</li> </ol>						
2.2. Enrolment requirements and required entry competences for the course	Passed exam from Theory of metal forming						
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge in engineering practice</li> <li>Suggest new and improved technical and technological solutions</li> <li>Analyse and combine metal forming processes</li> </ol>						
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Analyze the situation and development trends of the manufacturing processes of pipes and profiles,</li> <li>Calculate the technological parameters of the production of seamless pipes,</li> <li>Calculate the technological parameters of the production of welded pipes,</li> <li>Calculate the technological parameters of the production of profiles,</li> <li>Calculate the technological parameters of the production of profiles,</li> <li>Apply the scientific principles relevant to the manufacturing process of pipes.</li> </ol>						
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30) AND SEMINAR (15): 1. Introduction. The present state and development trends in the technologies of production of pipes and profiles. 2 2. The technology of production of seamless pipes. 2 3. Preparation and heating of bilet. Production of semi-hollow. 2 4. Hot rolling tubes. 2 5. Finishing tubes. Special procedures. 2 6. 1. Seminar: The rolling seamless pipes. 5 7. 1. Colloquium, chapters 1-6 8. The technology of production of welded tubes. 2 9. Welding in furnace. High-frequency welding. 2 10. Final treatment of tubes. Special procedures. 2 11. 2. Seminar: The rolling of welded pipes. 5 12. Cold rolling of tube. 4						

	<ul> <li>13. 2. Colloquium, chapters 8-12</li> <li>14. Profile rolling. Simple profiles. The complex profiles. 4</li> <li>15. Special features of the profile calibration. 4</li> <li>16. 3. Seminar: The profile rolling programs. 5</li> <li>17. Modern technologies in the production of pipes and profiles. 2</li> <li>18. 3. Colloquium, chapters 14-17</li> </ul>							
2.6. Type of instruction	○ lectures       ○ independent study         ○ seminars and workshops       ○ multimedia and the i         ○ online in entirety       ○ laboratory         ○ mixed e-learning       ○ (other)			e internet tor	2.7.	Comments:		
2.8. Student responsibilities	Attendance at lecture colloquium before wi	es min. 70 riting the o	0%, individual work on a colloquium or the writter	II exercises exam	and prepar	ation and submission of I	reports from field of	
2.9. Screening of student's work (specify the proportion of ECTS credits for	student's work (specify n of ECTS credits for         Class attendance         Research				Prac	ctical training		
each activity so that the total number of CTS credits is equal to the credit value of the course)):	Essay Tests Written exam	2.0	Oral exam Project	0.2		(Otherdescribe) (Otherdescribe) (Otherdescribe)		
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	The second of th							
	Title			Number of copies at the library		Availability	via other media	
2.11. Required literature (available at the library and via other media)	S. Rešković, Teorija oblikovanja deformiranjem, Sveučilište u Zagrebu, Metalurški fakultet, Sisak 2014., peer reviewed lessons,					https://www.simet.uniz nastava/predavanja/dij sveucilisni-studij-metal godina-diplomskog- studija/S%20Reskovic %20OBLIKOVANJA%2 DEFORMIRANJEM.pc	<u>g.hr/</u> <u>olomski-</u> <u>urgija/1-</u> %20TEORIJA 20 If/view	
	I. Mamuzić, V. M. Drujan, Teorija, materijali, tehnologija čeličnih cijevi, Hrvatsko metalurško društvo, Zagreb 1996.			;	34			
	B. Iharoš, Proizvodn 1987. (interna skripta	ja čeličnih a).	n cijevi, MF, Sisak,		16			
	S. Rešković, Tehnologija oblikovanja deformiranjem- nastavna građa, Sisak, 2011.				10			

2.12. Optional literature (at the time of the submission of the study programme proposal)	Professional journals, articles from this area.
2.13. Methods of monitoring quality that	Survey on the level of faculty and University.
ensure acquisition of exit	Analyses provided by quality assurance system of the institution.
competences	Analyses provided by quality assurance system and authorized office of the University.

Ordinal number	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning outcomes
1	Analyze the situation and development trends of the manufacturing processes of pipes and	1st colloquium, seminar paper, written and oral exam
-	profiles.	
2	Calculate the technological parameters of the production of seamless pipes.	1st colloquium, seminar paper, independent task
3	Calculate the technological parameters of the production of welded pipes.	2nd colloquium, seminar paper, independent task
4	Calculate the technological parameters of the production of profiles.	3rd colloquium, seminar paper, independent task, written
		and oral exam
5	Apply the scientific principles relevant to the manufacturing process of pipes.	Seminar paper, project task, oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Assoc.Prof. Tamara Holjevac Grgurić, PhD	1.6. Year of study	2			
1.2. Name of the course	NANOSTRUCTURED MATERIALS	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION						
2.1. Course objectives	Acquisition of knowledge about nanomaterial Defining the effect of processing parameters Introducing with specific areas of application	ls; structure, properties and processing. on structure and properties. of nanomaterials.				
2.2. Enrolment requirements and required entry competences for the course						
2.3. Learning outcomes at the level of the study programme to which the course contributes	Design the properties of metallic materials. Suggest appropriate methods for material qu Use the acquired theoretical knowledge in er	ality analysis. ngineering practice.				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	To define and classify nanomaterials. To apply different processing methods. To define one-dimensional, two-dimensional and special nanostructured materials. To analyse different types of metal based nanocomposites. To understand relationship between structure and properties of nanocomposites.					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction to nanoscience. Defining of nanostructures. Physical-chemical properties of solids. (2) Synthesis of nanostructured materials. Metal matrix nanomaterials. (3) Nanocomposites by Metal Alloying. (2) Nanocomposites from Sol-Gel Synthesis. (2) Nanocomposites by Thermal Spray Synthesis. (2) 1.preliminary exam Nanocomposites with metal matrix. Ceramic-metal nanocomposites. (4) One-dimensional and two-dimensional nanomaterials. Nanowires and thin films. (2) Special nanomaterials. Carbon nanotubes. (3) Polymer-metal nanocomposites. Nanoscale fillers. (2) 2.preliminary exam Structure and dispersion of nanoparticles in matrix, nucleation mechanisms and stabilisation. (2) Stabilisation of interphase. Characterisation of nanomaterials. Properties of nanomaterials. (2) Nanocomposites for ontical and electric applications. (2)					

	Nanocomposites for biomedical application. (2) 3.preliminary exam						
	LABORATORY EXERCISES (15).						
	Image: Seminars and workshops       independent study       2.7. Comments:						
2.6. Type of instruction	<ul> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>	in entirety e-learning vork					
2.8. Student responsibilities	Attendence to lectures min. 7	0 %. Atten	dance to lab	practice 100	) % (comp	esation of 2 exercies). Lab reports.	
2.9. Screening of student's work (specify	Class attendance	1	Research			Practical training	
the proportion of ECTS credits for	Experimental work	1	Report				
each activity so that the total	Essay		Seminar e	ssay		(Otherdescribe)	
number of CTS credits is equal to	Tests		Oral exam		1	(Other-describe)	
the credit value of the course)):	Written exam	1	Project			(Other-describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Lab reports, results of preliminary exams as well as written and oral exams. Written exam could be replaced with successful preliminary exams.						
	Title			Number copies at library	of the	Availability via other media	
2.11. Required literature (available at the library and via other media)	P. M. Ajayan, L. S. Schadler, P. V. Braun, Nanocomposite Science and Technology, Wiley- VCH, Verlag, 2003.			1			
	L. Nicolais, N. Carotenuto, Metal-Polymer Nanocomposites, John Wiley &Sons, 2005.				<u>http</u> sam	http://samples.sainsburysebooks.co.uk/9780471695424_ sample_386645.pdf	
2.12. Optional literature (at the time of the submission of the study programme proposal)	Kohler, W. Fritzsche, Nanotechnology, Wiley-VCH Verlag, 2005.						
2.13. Methods of monitoring quality that	Internal student survey. Analysis of attendance to lectures and exercises, results of preliminary exams as well as oral exams.						
ensure acquisition of exit							
onouro acquionion or oxit	Survey on the level of faculty	and Unive	rsity.				

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of		
number		learning outcomes		
1	To define and classify nanomaterials.	1st colloquium, written and oral exam		
2	To apply different processing methods.	1st colloquium, written and oral exam		
3	To define one-dimensional, two-dimensional and special nanostructured materials.	2nd colloquium, written and oral exam		
4	To analyse different types of metal based nanocomposites.	2nd and 3rd colloquium, written and oral exam, exercises		
5	To understand relationship between structure and properties of nanocomposites.	3rd colloquium, written and oral exam, exercises		
6	To correlate content, structure and properties of nanomaterials with application requirements.	1st, 2nd and 3rd colloquium, written and oral exam		

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:					
1.1. Course teacher	Assoc.Prof. Natalija Dolić, PhD	1.6. Year of study	2		
1.2. Name of the course	SEMICONTINUOUS CASTING OF ALUMINIUM ALLOYS	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10		
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	<ol> <li>Introduce students with the technologies of industrial casting of aluminum and aluminum alloys intended for further processing: horizontal ("HDC") and vertical ("VDC") casting process by direct cooling with water.</li> <li>Acquisition of knowledge of solidification and microstructure evolution in ingot and billet during "DC" casting.</li> <li>Define the basic casting parameters.</li> <li>Train students how to recognize the basic phases in the microstructure of aluminum alloys on the optical microscope in cast and homogenized condition and how to determine the grain size.</li> </ol>				
2.2. Enrolment requirements and required entry competences for the course	-				
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Combine social, ethical and business principles and norms in the professional field. Plan and manage metallurgical processes. Compare the procedures of material treatment with microstructure and useful properties.				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Outline the Al-Mg phase diagram.</li> <li>Analyze the basic casting parameters in production of aluminium ingots by "VDC" process.</li> <li>Detect the surface defects on semicontinuously cast aluminum semi-finished products.</li> <li>Identify the individual phases in the microstructure of samples cast by "VDC" process on the optical microscope.</li> <li>Demonstrate the principle of determining the grain size by semiautomatic method on aluminum alloys</li> </ol>				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction to the plan course and the time schedule for the colloquium. Aluminium and their alloys. System for designation aluminium and wrought aluminium alloys. Basic standards (2). Technology of industrial casting of aluminum and aluminum alloys intended for further processing. Horizontal ("HDC") and vertical ("VDC") casting process by direct cooling with water (2). Direct-Chill ("DC") casting: solidification and structure patterns (2). Microsegregation. Macrosegregation in "DC" casting of aluminium alloys. Porosity (3). Grain refinement in aluminium alloys (2). The solidification of Al-Mg system (2). Tests I				

-									
		Melting. Alloying. Purification of the melt. Inclusions in aluminum (5).							
		Casung, Casung parameters (2). Surface defects in indots casting by "VDC" process. Possible causes, measures for removal (2)							
		Homogenization of aluminium ingots (2).							
		Technology for casting alumin	um billets b	y direct cooling with water (6).					
		Test II.							
		EXERCISES (15):							
		the optical microscope (ovami	aluminium	samples series 5xxx (EIN 508)	3, EN 5754, EN 3	0052) at cast and nor	nogeniz	zed state at	
		Measuring the effectiveness of	f removing i	inclusions and decassing melt	t (examples from	practice) (2)	u) (0).		
		Control and influence of castir	na paramete	ers (examples from practice) (5	5).	praotico) (2).			
		Tests I, II (2).	01		,				
				independent study 2.		.7. Comments:			
		Seminars and workshops		multimedia and the inter	net				
2.6. T	ype of instruction			🛛 laboratory	⊠ laboratory				
				work with the mentor					
				Seminar paper					
2.8.	Student responsibilities	Regular attendance (> 70 %).							
2.9.	Screening of student's work	Class attendance		Research	F	Practical training		1	
	(specify the proportion of ECTS			Depert	(	optical microscope)			
	credits for each activity so that								
	is equal to the credit value of the	Essay		Oral aver	1	(Other descri	h a)		
	course)):	Vritten even	2	Draiexam		(Other_descri	be)		
		Continuous monitoring and	2 avaluation			(Other-desch	be)		
		Continuous monitoring and evaluation of student:							
		right to take not passed test one more time. Both positive evaluation tests release the student of laving the final evan. At each							
0.40	Creding and evolution of student	tests student can achieve a maximum of 10 points for the question, number of questions is 5. For satisfactory accomplishment							
2.10.	work over the course of	in each tests student must collect more than 30 % of points for each question. If student fails the examination by tests, laying							
	instruction and at a final exam the final exam (written + oral).								
		Continuous monitoring and evolution of students							
		Tests $(1 + 11)$ writing + oral: 3 ECTS							
		Practical training (optical microscopy): 1 ECTS							
		Title				Number of			
0.44						copies at the	Avai	lability via	
2.11.	the library and via other media)					library	U		
the library and via other media)		D. G. Eskin, Physical Metallurgy od Direct Chill Casting od Aluminium Alloys, CRC							
		Press/Taylor and Francis Grou	up, Boca Ra	aton, 2008.		· ·			

2.12. Optional literature (at the time of the submission of the study	
programme proposal)	
2.13. Methods of monitoring quality that	Examination of students who have finished study.
ensure acquisition of exit	Survey on the faculty and University level.
	Analysis predicted by systems for insurance of institution quality.
competences	Analysis predicted by systems for insurance quality from authorized University office.

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of		
number		learning outcomes		
1	Outline the AI-Mg phase diagram.	1st colloquium, written and oral exam exam		
2	Analyze the basic casting parameters in production of aluminium ingots by "VDC" process.	2nd colloquim, written and oral exam		
3	Detect the surface defects on semicontinuously cast aluminum semi-finished products.	2nd colloquim, written and oral exam		
4	Identify the individual phases in the microstructure of samples cast by "VDC" process on the	Laboratory exercises, written and oral exam		
	optical microscope.			
5	Demonstrate the principle of determining the grain size by semiautomatic method on aluminum	Laboratory exercises		
	alloys.			

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Assoc.Prof. Ivan Brnardić, PhD	1.6. Year of study	3			
1.2. Name of the course	THE LIFE CYCLE OF METAL PRODUCTS	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+15+0+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	e 1., 5%			
2. COURSE DESCRIPTION						
2.1. Course objectives	To introduce students with the general princip To able to set goals, scope, limits and to colle To acquire knowledge about explaining and r	ples, concepts, frameworks, methods and o ect data for the analysis of the life cycle. research evaluation of lifecycle.	challenges of life cycle analysis.			
2.2. Enrolment requirements and required entry competences for the course	Knowledge on metal materials, work on computers and with computer applications.					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Analyse the present situation, identify problems, formulate and recommend the optimal technological solution by using the knowledge acquired. Manage metallurgical production residues.					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Explain and understand the importance of the life cycle analysis.</li> <li>Know how to set goals, scope and limits for assessments.</li> <li>Able to collect quality data for analysis.</li> <li>Able to explain the results and derive conclusions of the analysis.</li> </ol>					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction, objectives, scope and methodology for life-cycle analysis. 3h Introduction to the project of the life cycle of world steel production and objectives. Description of the project through a description of the system, setting boundaries, data collection, explanation and critical review. The quality of the data, the life cycle model, evaluation of results, analysis, explanations and conclusions. 9h Introduction to LCA methodology for metals, objectives and scope. The boundaries of the system, by-products, recycling and impact of life cycle. 9h Example of a comparative analysis of the life cycle on truck wheel from aluminum and steel. The objectives, scope and lifecycle analysis and assessment. Explanation of analysis results. 9h SEMINAR (15): Example seminar work and preparation and presentation of seminar paper.					
2.6. Type of instruction	<ul> <li>Iectures</li> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> </ul>	<ul> <li>independent study</li> <li>multimedia and the internet</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>	2.7. Comments:			
	field work					
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2.8. Student responsibilities	Regular attendance of lecture	es (70% of the	lectures) and seminar	work.		
2.9. Screening of student's work (specify	Class attendance	0.5	Research		Practical training	
the proportion of ECTS credits for	Experimental work		Report			
each activity so that the total	Essay		Seminar essay	1	(Otherdescribe)	
number of CTS credits is equal to	Tests		Oral exam	1.25	(Other-describe)	
the credit value of the course)):	Written exam	1.25	Project		(Other-describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Attendance on classes, prepared and oral exam.	aration and pre	sentation of seminar p	aper, continuous n	nonitoring – 2 preliminary e	xams or written
2.11. Required literature (available at the library and via other media)	Title	Title Nu		Availability via other media		a
	I. Brnardić, Lectures from The life cycle of metal products, Sisak, 2016.			Merlin system for e-learning		
	Life Cycle assessment methodology report, World steel association, Brussels, Belgium, 2011.			https://www.worldsteel.org/dms/internetDocumentList/books hop/LCA- MethodologyReport/document/LCA%20Methodology%20Re port.pdf		
	Comparative Life Cycle Assessment of Aluminum and Steel Truck Wheels, PE INTERNATIONAL, Inc., Boston, USA, 2012.			http://www.alcoay pdf/Alcoa_Compa atement.pdf	wheels.com/alcoawheels/no arative LCA of Truck Wh	orth_america/en/ eels_with_CR_st
	Harmonization of LCA Methodologies for Metals, PE INTERNATIONAL, Inc., Boston, USA, 2014.			https://www.icmn	n.com/document/6657	
2.12. Optional literature (at the time of the submission of the study programme proposal)	Available scientific literature on the subject of life-cycle analysis.					
2.13. Methods of monitoring quality that	Survey on the faculty and Un	iversity level.				
ensure acquisition of exit	Analysis predicted by system	is for insurance	of institution quality.			
competences	Analysis predicted by systems for insurance quality from authorized University office.					

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of			
number		learning outcomes			
1	Explain and understand the importance of the life cycle analysis.	1st colloquium, seminar paper, written and oral exam			
2	Know how to set goals, scope and limits for assessments.	1st colloquium, seminar paper, written and oral exam			
3	Able to collect quality data for analysis.	2nd colloquium, seminar paper, written and oral exam			
4	Able to explain the results and derive conclusions of the analysis.	2nd colloquium, seminar paper, written and oral exam			

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:					
1.1. Course teacher	Assoc.Prof. Ivan Brnardić, PhD Assist.Prof. Tahir Sofilić, PhD	1.6. Year of study	2		
1.2. Name of the course	UTILIZATION OF METALLURGICAL PRODUCTION REMAINS	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+15+0+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	aduate1.9. Expected enrolment in the course10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
2.1. Course objectives	<ol> <li>To acquaintance students with remains disposal.</li> <li>To get insight in creation and maintenar</li> </ol>	s from metallurgical production, its occurrent	ce, possibilities of recovery until the final		
2.2. Enrolment requirements and required entry competences for the course	Knowledge of metallurgical processes, waste management and work on computer.				
2.3. Learning outcomes at the level of the study programme to which the course contributes	Apply logical conclusion and precision in data processing. Compare and choose individual technological process. Manage metallurgical production residues.				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Define waste according to properties and place of origin in metallurgical processes.</li> <li>Enumerate ways for waste treatment.</li> <li>Enumerate actions for avoiding and reducing of waste and reducing its dangerous properties.</li> <li>Propose resource recovery from waste.</li> <li>Suggest recycling of water and reagents in processes.</li> </ol>				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Introduction. Waste characterization. 4h Physical and physico-chemical processes, material preparation, gravity separation processes, Magnetic and electrostatic separation, shredding systems. 4h Uses hydrometallurgical processes, biotechnological and pyrometallurgical processing. 7h Metal recycling, ferrous and non-ferrous. Metallurgical slags, dust and fumes. By-product processing and utilization. 7h Resource recovery from process wastes. Recycling of water and reagents. Emerging new technologies. 8h SEMINAR (15): Example seminar work and preparation and presentation of seminar paper.				
2.6. Type of instruction	6. Type of instruction       Image: Seminar work and preparation and presentation of seminar paper.       2.7. Comments:         6. Type of instruction       Image: Seminar seminar work and preparation and presentation of seminar paper.       2.7. Comments:         6. Type of instruction       Image: Seminar seminar work and workshops       Image: Seminar seminar paper.       2.7. Comments:         Image: Seminar seminar work and workshops       Image: Seminar seminar paper.       Image: Seminar seminar paper.       2.7. Comments:         Image: Seminar seminar seminar seminar seminar paper.       Image: Seminar seminar seminar seminar seminar paper.       2.7. Comments:         Image: Seminar se				

	field work						
2.8. Student responsibilities	Regular attendance of lectu	ular attendance of lectures (70% of the lectures) and seminar work.					
2.9 Screening of student's work (specify	Class attendance	0.5	Research		Pract	tical training	
the proportion of ECTS credits for	Experimental work		Report				
each activity so that the total number	Essay		Seminar essay	1.0		(Otherdescribe)	
of CTS credits is equal to the credit	Tests		Oral exam	1.25		(Other-describe	)
value of the course) <i>):</i>	Written exam	1.25	Project			(Other-describe	)
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Attendance on classes, prep and oral exam.	Attendance on classes, preparation and presentation of seminar paper, continuous monitoring – 2 preliminary exams or written and oral exam.					exams or written
			Title			Number of copies at the library	Availability via other media
2.11. Required literature (available at the library and via other media)	S. Ramachandra Rao, Resource recovery and recycling from metallurgical wastes, 1 Elsevier, Amsterdam, The Netherlands, 2006.						
	I. Brnardić, T. Sofilić, Predavanja iz Iskorištavanje metalurških proizvodnih ostataka, Sisak, 2016.						
	I. Brnardić, T. Sofilić, Preda Sisak, 2016.	vanja iz Isko	rištavanje metalurških proi:	zvodnih ostataka,			Merlin system for e-learning
0.40 Online of the store of	I. Brnardić, T. Sofilić, Preda Sisak, 2016.	vanja iz Isko	rištavanje metalurških proi:	zvodnih ostataka,			Merlin system for e-learning
2.12. Optional literature (at the time of	I. Brnardić, T. Sofilić, Preda Sisak, 2016. Available scientific literature	vanja iz Isko	rištavanje metalurških proi: ect of utilization of metallur	zvodnih ostataka, gical production rer	nains.		Merlin system for e-learning
2.12. Optional literature (at the time of the submission of the study programme proposal)	I. Brnardić, T. Sofilić, Preda Sisak, 2016. Available scientific literature	vanja iz Isko	rištavanje metalurških proi: ect of utilization of metallur	zvodnih ostataka, gical production rer	nains.		Merlin system for e-learning
<ul> <li>2.12. Optional literature (at the time of the submission of the study programme proposal)</li> <li>2.13. Methods of monitoring quality that</li> </ul>	I. Brnardić, T. Sofilić, Preda Sisak, 2016. Available scientific literature Survey on the faculty and U	vanja iz Isko on the subje niversity leve	rištavanje metalurških proi: ect of utilization of metallur el.	zvodnih ostataka, gical production rer	nains.		Merlin system for e-learning
<ul> <li>2.12. Optional literature (at the time of the submission of the study programme proposal)</li> <li>2.13. Methods of monitoring quality that ensure acquisition of exit</li> </ul>	I. Brnardić, T. Sofilić, Preda Sisak, 2016. Available scientific literature Survey on the faculty and U Analysis predicted by system	on the subje niversity leve ns for insura	rištavanje metalurških proi: ect of utilization of metallur el. ince of institution quality.	zvodnih ostataka, gical production rer	nains.		Merlin system for e-learning

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of			
number		learning outcomes			
1	Define waste according to properties and place of origin in metallurgical processes.	1st colloquium, seminar, written and oral exam			
2	Enumerate ways for waste treatment.	1st colloquium, seminar, written and oral exam			
3	Enumerate actions for avoiding and reducing of waste and reducing its dangerous properties.	1st colloquium, seminar, written and oral exam			
4	Propose resource recovery from waste.	2nd colloquium, seminar, written and oral exam			
5	Suggest recycling of water and reagents in processes.	2nd colloquium, seminar, written and oral exam			

1.1. Course teacher	Full Prof. Ladislav Lazić, PhD	1.6. Year of study	2			
1.2 Nome of the source		ull Prof. Ladislav Lazić, PhD   1.6. Year of study   2				
1.2. Name of the course	RENEWADLE ENERGI SOURCES	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	ective 1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)				
2. COURSE DESCRIPTION						
2.1. Course objectives	<ol> <li>Provide students the informations about implementation, and impact on the enviloant Obtain basic knowledge for independent alternative energy sources.</li> <li>Acquire the ability to solve problems in</li> </ol>	It renewable energy sources and issues related ironment. In critical thinking about technical and economi the field of application and use of renewable e	d to their development, ic perspective of the applicability of energy sources.			
2.2. Enrolment requirements and required entry competences for the course	Passed the exams of course Energy management.					
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Analyse the development and application of new technologies.</li> <li>Suggest new and improved technical and technological solutions.</li> <li>Formulate and suggest measures for increasing energy efficiency.</li> </ol>					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Distinguish between conventional and non-conventional energy sources.</li> <li>Estimate the amount of energy which can give a source.</li> <li>Evaluate the harmful impact of a source of energy on the environment.</li> <li>Determine the safety and economic profitability of using a source of energy.</li> </ol>					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ol> <li>LECTURES (30):         <ol> <li>Conventional and non-conventional energy: Geothermal energy: Geothermal energy aspects. (4)</li> <li>Solar energy: Solar radiation, Solar co Bioconversion. (10)</li> </ol> </li> <li>Wind energy: Wind speed, Energy and 5. Biomass and biofuels: Classification of fermentation, Anaerobic digestion, Pro ecological aspects. (10)</li> </ol>	ergy sources. Renewable energy sources. (2) gy of underground fluids, Geothermal energy o llectors, Storage of solar energy, Solar heating 1st colloquium f power of wind, Wind turbine, Wind power plan f biofuels, Production of biomass, Direct combine cessing of waste and scrap, Vegetable oils an	f dry rocks, Social and ecological g and cooling, Producing of electricity, nts. (4) ustion, Pyrolysis, Alcoholic d biodiesel, Sociological and			

	EXERCISES (15): Solution of practical problems. The understanding of the material exposed in lectures is facilitated by solving the two programs. The programs are selected so that they expand the presented theory and illustrate the application of theory to real problems.							
			independent study	2	2.7. Comments:	7. Comments:		
2.6. Type of instruction		<ul> <li>multimedia and the internet</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>						
2.8. Student responsibilities	Attendance on Lectures and E	xercises > 7	70 %					
2.9. Screening of student's work (specify	Class attendance	0.4	Research		Practical training			
the proportion of ECTS credits for	Experimental work		Report					
each activity so that the total number of CTS credits is equal to the credit	Essay		Seminar essay	Seminar essay				
	Tests	1.0	Oral exam	(Other-describe	(Other—describe)			
value of the course)):	Written exam		Project	(Other-describe)				
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Projects – 30% Class attendance – 5% Written exam – 30% Oral exam – 35%							
2.11. Required literature (available at	Title			Number of copies at the library	Avai oth	ilability via ner media		
the library and via other media)	V. Knapp, P. Kulišić, Novi izvori energije, Školska knjiga Zagreb, 1985.				1			
	P. Kulišić, Novi izvori energije,	1						
2.12. Optional literature (at the time of the submission of the study programme proposal)	A. V. da Rosa, Fundamentals of renewable energy resources, Elsiver, Amsterdam, 2005. J. Twidel, T. Weir, Renewable energy resources, Taylor & Francis, London and New York, 2006.							
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Survey at the level of faculty and University. Analyses provided in the system of quality assurance of the institution. Analyses provided in the system of quality assurance and an authorized office of the University.							

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Distinguish between conventional and non-conventional energy sources.	1st colloquium, oral exam
2	Estimate the amount of energy which can give a source.	1st colloquium, oral exam
3	Evaluate the harmful impact of a source of energy on the environment.	2nd colloquium, oral exam
4	Determine the safety and economic profitability of using a source of energy.	2nd colloquium, seminar paper, oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:					
1.1. Course teacher	Assist.Prof. Tahir Sofilić, PhD	1.6. Year of study	2		
1.2. Name of the course	CIRCULAR ECONOMY	1.7. Credit value (ECTS)	4		
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+15+0+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10		
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%		
2. COURSE DESCRIPTION					
Familiar students with the existing development level of productive capacity and development level of the system values, in which material resources are not always exploited sufficiently. Students need to understand that it is therefore necessary to improve the use of resources, as in the production process and their management especially part which is in the production process not embodied in the product, which is commonly referred as waste. Explain to students that the process, in which the economy of the primitive accumulation of capital and the industrial revolution grew on the principle: "take, make, consume, throw" and that was a so-called "Linear model" which is based on the assumptior of unlimited and easy availability of resources, it is necessary to achieve a program for successful using of material resources the context of the Europe 2020 strategy, which only can ensure sustainable and inclusive growth.					
2.2. Enrolment requirements and required entry competences for the course					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Combine social, ethical and business principles and norms in the professional field. Suggest solutions for the optimization of metallurgical processes. Combine the skills necessary for lifelong learning, including continued professional training.				
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Describe the differences between linear and circular economy. Explain the circular economy as alternative for a linear economy guided by principles,, take, make, consume, throw ". Illustrate the possibility of increasing the productivity of resources in parallel with the existing EU policy objectives such as the reduction of carbon dioxide emissions, increase energy efficiency etc. Propose to increase resource productivity while reducing potential adverse effects on the environment and greenhouse gas emissions on the example of a metallurgical process				
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Limiting the negative environmental footprints of the existing European economy and reducing the cost of the economic activities with a aim of economic growth; 2h The Europe 2020 strategy, 2h. The priorities of the 2020 strategy (Smart growth: development of the economy based on knowledge and innovation Sustainable growth: promoting an economy that effectively exploit resources, which is greener and more competitive, an Inclusive growth: fostering economy with high employment that delivers social and territorial cohesion), 4h. The concept of a linear economy, 1h The transition of the linear concept to circular economy. 1h				

The transition from the existing linear to circular economy ie. economic model that ensures sustainable management of resources and the extension of the lifetime of materials and products as a central part of the 2020 strategy; 2h
Circular economy as a kind of alternative to worn model of a linear economy guided by principles ,, take, make, consume, throw", 2h
Fundamentals of smart, sustainable and inclusive economy growth (efficient use, recovery, recycling and reuse of resources), 2h
Systematically reducing the generation of waste to a minimum during the entire life cycle of the product and its components, 2h. The importance and contribution to waste management policy in terms of reducing its creation through the development of a functioning waste management system, which aims to use waste as a valuable resource, 2h.
In systems of circular economy added value of products is retain as long as possible and does not create waste, 2h. Resources retain in the economy after the end of the product lifetime how they can be productively used, thus creating new value, 2h.
Possibilities of improving ongoing sustainable growth by increasing integration of economic growth, resource use and its effects, 2h.
Increasing the productivity of resources parallel with the existing EU policy objectives (reduction in carbon dioxide emissions, increase energy efficiency, ensuring access to raw materials), 2h.
Increasing resource productivity while reducing potential adverse effects on the environment and greenhouse gas emissions, 2h.
SEMINAR (15):
Instructions for the preparation of the seminar (2h)
I opics presentation and selection (1n)
Making PPT of seminar work and preparing for presentation (1b)
Presentation of seminar work (5b)
TESTS.
1. Test
Limiting the negative environmental footprints of the existing European economy and reducing the cost of the economic activities with aim for economic growth; The Europe 2020 strategy, smart growth and the development of economy based on knowledge and innovation; Sustainable growth and promote a more efficient economy that exploits resources, inclusive growth and nurturing of the economy with high employment that delivers social and territorial cohesion; The concept of a linear transition in circular economy.
The transition from the existing linear to circular economy ie. economic model that ensures sustainable management of resources and the extension of the lifetime of materials and products as a central part of the 2020 strategy; Circular economy as a kind of alternative to old model of a linear economy, basic postulates of smart, sustainable and inclusive economy growth (efficient use, recovery, recycling and reuse of resources), systematically reducing the generation of waste to a minimum during the entire life cycle of the product and its components, significance and contribution to waste management policy in terms of reducing its creation through the development of a functioning waste management system.
3. Lest Systems of circular economy in the function of keeping resources in the economy after the end of the product lifetime to be used productively, thus creating new value, Possibilities of improving ongoing sustainable growth by increasing integration of economic growth, resource use and its effects, increase in productivity of resources in parallel with the existing EU policy objectives (reduce carbon emissions, increase energy efficiency, ensuring access to raw materials), increasing resource

	productivity while reducing potential adverse effects on the environment and greenhouse gas.						
2.6. Type of instruction	Image: Section of the section of t		lependent study ultimedia and the internet poratory ork with the mentor (other)		2.7. Comments:		
2.8. Student responsibilities	Students must attend more that	an 70% of lect	tures	and make seminar		•	
2.9. Screening of student's work (specify the proportion of ECTS credits for	Class attendance Experimental work		Res Rep	earch ort		Practical training	
each activity so that the total	Essay		Sem	ninar essay	1	(Otherdescribe)	
number of CTS credits is equal to	Tests	3	Ora	exam		(Other-describe)	
the credit value of the course)):	Written exam		Proj	ect		(Other-describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Seminar work, continuous monitoring through 3 mid tests or written and oral exam.			· · · ·			
	Title			Number of copie at the library	S	Availability via other medi	а
	M. Krišto, Kružna ekonomija za brži razvoj, Gospodarstvo i okoliš, Hrvatski poslovni savjet za održivi razvoj <b>11</b> , 41 (2015).		1				
2.11. Required literature (available at	TOWARDS THE CIRCULAR ECONOMY- Economic and business rationale for an accelerated transition, vol <b>1</b> (2013).				https://www oads/public Towards-th	v.ellenmacarthurfoundation.org cations/Ellen-MacArthur-Found ne-Circular-Economy-vol.1.pdf	/assets/downl ation-
the library and via other media)					http://www centre.com Economy.p	<u>.c2c-</u> //sites/default/files/Towardsa%/ odf	20Circular%20
	Europska Komisija, EUROPA 2020 - Europska strategija za pametan, održiv i uključiv rast, Bruxelles, 2010.		oska rast,		http://www ments/Stru <u>3%202020</u>	http://www.strukturnifondovi.hr/UserDocsImages/Docu ments/Strukturni%20fondovi%202014.%20%E2%80%9 3%202020/eu_hr.pdf	
2.12. Optional literature (at the time of the submission of the study programme proposal)	T. Sofilić, ZDRAVLJE I OKOLIŠ, skripta, Sveuči T. Sofilić, ODRŽIVO GOSPODARENJE OTPAL COMMUNICATION FROM THE COMMISSIC ECONOMIC AND SOCIAL COMMITTEE AND waste programme for Europe, Brussels, economy/pdf/circular-economy-communication.		eučilis PADC SSION AND els, tion.po	šte u Zagrebu, Met DM, skripta, Sveuči I TO THE EURO THE COMMITTEE 2.7.2014 COM(2 df	alurški fakultet, lište u Zagrebu PEAN PARLIA OF THE REG 014) 398 fin	2015. , Metalurški fakultet, 2015. MENT, THE COUNCIL, THE JONS, Towards a circular eco al, <u>http://ec.europa.eu/enviror</u>	E EUROPEAN pnomy: A zero nment/circular-

2.13. Methods of monitoring quality that	Students survey input and output. Numerical analysis of tests and exams by scoring task by task at the course level.
ensure acquisition of exit	Survey on the faculty and University level.
competences	Analysis predicted by systems for insurance of institution quality.

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Describe the differences between linear and circular economy.	1st colloquium, seminar, written and oral exam
2	Explain the circular economy as alternative for a linear economy guided by principles,, take,	1st and 2nd colloquium, seminar, written and oral exam
	make, consume, throw ".	
3	Illustrate the possibility of increasing the productivity of resources in parallel with the existing EU	2nd and 3rd colloquium, seminar, written and oral exam
	policy objectives such as the reduction of carbon dioxide emissions, increase energy efficiency	
	etc.	
4	Propose to increase resource productivity while reducing potential adverse effects on the	3rd colloquium, seminar, written and oral exam
	environment and greenhouse gas emissions on the example of a metallurgical process.	

1. COURSE DECRIPTION - GENERAL	INFORMATION ISVU CODE:					
1.1. Course teacher	Assist.Prof. Tahir Sofilić, PhD Assoc.Prof. Ivan Brnardić, PhD	1.6. Year of study	2			
1.2. Name of the course	ENVIRONMENTAL LAW	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+e-learning)	30+15+0+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION			•			
2.1. Course objectives	Acquaintance students with directions of sustainable development which insure economic development of community and at the same time insure conditions for protection of natural resources. Explain and adopt the principle of environmental law and environmental law sources in the Republic of Croatia. Acquaintance with skills which will conciliate opposite interests of industry and economy in global with requirements for the environmental protection.					
2.2. Enrolment requirements and required entry competences for the course	-					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Combine social, ethical and business principles and norms in the professional field. Combine the skills necessary for lifelong learning, including continued professional training. Recognize and apply scientific principles important in the field of metallurgy.					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	To define base questions regarding the environmental law. To distinguish quality of life from life quality. To describe ecology policy and ecology strategy. To illustrate protection of specially worth parts of nature. To compare the environmental protection in Croatian legal system with EU. To explain obligations of economic operators toward current legislation					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	To explain obligations of economic operators toward current legislation. LECTURES (30): Base questions regarding environmental law, current situation in environmental law, 2h Principals of the environmental law, sources of the environmental law in RH, 2h Ecology policy and ecology strategy, quality of life and life quality, 2h Normative institutions of ecology policy, institutions for the environmental protection in RH, 2h The environmental protection in Croatian legal system, 2h The protection of specially worth parts of the nature, 2h The protection and improvement of forests, 2h The protection of agricultural soil, 2h					

	The protection of soil, 2h The protection of water, 2 The protection from noise The protection from radiat The environmental protect Economic operators and t SEMINAR (15): Instructions for the prepar Topics presentation and s Creating individual seminar Making PPT of seminar w Presentation of seminar w TESTS 1. TEST: Basic issues of environr environmental law in the Normative institutions of legal system. 2. Test: Protection of particularly w air, soil protection, water p 3. Test:	h , 2h. ion, 2h tion in inter heir obligat ation of the election (1l ar work, su ork and pre rork (5h) mental law Republic c environment valuable na protection, n	national legal syste ions toward current e seminar (2h) h) pervision and corre- paring for presenta of Croatia, environn ntal policy, environ tural areas, protect noise protection, ra	m, 2h legislation ctions (6) tion (1h) e of envir nental pol mental ins on and pro	n., 2h ronmental I icy and env stitutions in omotion of t	aw, principles of enviro /ironmental strategies, qu Croatia, Environmental forests, protection of agri	onmental law, sources of uality of life or life quality, Protection in the Croatian icultural land, protection of
2.6. Type of instruction	legislation, examples of s applicable regulations. ☐ lectures ☐ seminars and worksho ☐ exercises	ps	independent study multimedia and the laboratory	internet		2.7. Comments:	he type of activity and the
2.8. Student responsibilities	in mixed e-learning field work	e than 70%	work with the ment (other)	or oke semin	ar		
	Class attendance		Posoarch			Practical training	
2.9. Screening of student's work (specify	Experimental work		Report		I		
the proportion of ECTS credits for each activity so that the total	Esperimental work			1		(Other-describe)	
number of CTS credits is equal to	Toete	3	Oral evam			(Other describe)	
the credit value of the course)):	Written exam	5	Project			(Other describe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Seminar work, continuous monitoring through 3 mid tests or written and oral exam.						
2.11. Required literature (available at the library and via other media)	Titl	e	Nu co	nber of pies at		Availability via ot	her media

		the library					
	T. Sofilić, PRAVO OKOLIŠA, skripta, Sveučilište u Zagrebu, Metalurški fakultet, 2015.		https://www.simet.unizg.hr/nastava/predavanja/preddiplomsk i-sveucilisni-studij-metalurgija				
2.12. Optional literature (at the time of the submission of the study programme proposal)	O. Lončarić-Horvat, L. Cvitanović, I. Gliha, T. Jos dopunjeno izdanje, Ministarstvo zaštite okoliša i pr Deklaracija o zaštiti okoliša u Republici Hrvatskoj Zakon o zaštiti prirode (NN br. 80/13) Zakon o zaštiti okoliša (NN br. 80/13) Zakon o zaštiti zraka (NN br. 130/11,47/14) Zakon o održivom gospodarenju otpadom (NN br. Zakon o vodama (NN br. 153/09,63/11,130/11,56/ Zakon o kemikalijama (NN br.18/13)	sipović, D. Medv rostornog uređen (NN br. 34/92) 94/13) 13,14/14),	vedović, J. Omejec, M. Seršić, Pravo okoliša, 3. izmijenjeno i nja i Organizator (Pub), Zagreb, 2003.				
2.13. Methods of monitoring quality that	Students survey input and output. Numerical analy	sis of tests and	exams by scoring task by task at the course level.				
ensure acquisition of exit	Survey on the faculty and University level.						
competences	Analysis predicted by systems for insurance of ins	titution quality.					

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	To define base questions regarding the environmental law.	1st colloquium, seminar, written and oral exam
2	To distinguish quality of life from life quality.	1st colloquium, seminar, written and oral exam
3	To describe ecology policy and ecology strategy.	2nd colloquium, seminar, written and oral exam
4	To illustrate protection of specially worth parts of nature.	2nd colloquium, seminar, written and oral exam
5	To compare the environmental protection in Croatian legal system with EU.	3rd colloquium, seminar, written and oral exam
6	To explain obligations of economic operators toward current legislation.	3rd colloquium, seminar, written and oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Full Prof. Ladislav Lazić, PhD	1.6. Year of study	2			
1.2. Name of the course	LOW – EMISSION COMBUSTION	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	1.8. Type of instruction (number of hours L+S+E+elearning)	30+0+15+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION						
2.1. Course objectives	<ol> <li>Acquire knowledge about certain types of pollutant emissions into the environment as a result of the combustion process.</li> <li>Acquire knowledge on the mechanisms of formation of NO<sub>X</sub> and methods of their reduction during the combustion process.</li> <li>Acquire knowledge on the mechanisms of formation of SO<sub>2</sub> and methods of their reduction during the combustion process.</li> <li>Acquiring knowledge about the formation and reduction of CO<sub>2</sub> emissions.</li> </ol>					
2.2. Enrolment requirements and required entry competences for the course	The acquired knowledge from the course of graduate study: Industrial furnaces, Heating technology of industrial furnaces, Energy management.					
2.3. Learning outcomes at the level of the study programme to which the course contributes	<ol> <li>Use the acquired theoretical knowledge in engineering practice.</li> <li>Suggest new and improved technical and technological solutions.</li> <li>Analyse the development and application of new technologies.</li> <li>Combine the skills necessary for lifelong learning, including continued professional training.</li> </ol>					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Identify the type of pollutant emissions into the environment as a result of the combustion process.</li> <li>Analyse the causes of certain pollutant emissions.</li> <li>Suggest a method for reducing certain pollutant emission.</li> <li>Evaluate the effectiveness of applied method.</li> </ol>					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	<ol> <li>LECTURES (30):         <ol> <li>Ecological noxiousness of combustion product. (2)</li> </ol> </li> <li>Formation of the emission of nitrogen oxides in the course of combustion: Thermal nitrogen oxide, Prompt nitrogen oxide, Nitrogen oxide generated by N<sub>2</sub>O, Fuel nitrogen oxide generated by fuel, Mechanisms of the formation of fuel NO, Formation of NO<sub>2</sub>, Generalisation of the description of the mechanisms of NO<sub>x</sub> generation. (4)</li> <li>Abatement of the emission of nitrogen oxides in the course of combustion: Staged combustion, Supply of ammonia or urea to the combustion chamber, Decrease of temperature in the combustion zone, High Temperature Air Combustion (HITAC) technology of flameless combustion, The influence of the fundamental operation parameters on the emission of NO<sub>x</sub>. (6)         <ul> <li>1st colloquium</li> </ul> </li> </ol>					
	4. Formation and reduction of SO <sub>2</sub> emission in the course of combustion (4): Sulphur compounds in fuels, Transformation of fuel sulphur in the course of rapid preheating, Oxidisation of sulphur compounds in the flame, High-temperature binding of SO <sub>2</sub> in the course of combustion with a shortage of oxygen, Influence of additives on the degree of binding of SO <sub>2</sub> in					

	<ul> <li>the combustion gases. (4)</li> <li>Emission of carbon oxide, Mechanisms of the formation and oxidation of CO. (2)</li> <li>Emission of carbon dioxide: Greenhouse effect, Formation and decrease of CO<sub>2</sub> emission. (2)</li> <li>Formation and emission of combustible solid particles. (2)</li> <li>Formation and emission of polycyclic aromatic hydrocarbons during the combustion. (2)</li> <li>Noxious substance occurring in minute quantities in combustion processes: Chlorine and fluorine compounds, Heavy metal compounds. (2)</li> <li>EXERCISES (15): Solving the practical problems. The understanding of the material exposed in lectures is facilitated by solving the two program tasks. The program tasks are selected so that they expand the presented theory and illustrate the application of theory to real problems.</li> </ul>							
2.6. Type of instruction	<ul> <li>lectures</li> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>		<ul> <li>independent study</li> <li>multimedia and the interr</li> <li>laboratory</li> <li>work with the mentor</li> <li>(other)</li> </ul>	net	2.7	7. Comments:		
2.8 Student responsibilities	Attendance on Lectures and	Exercises >	> 70 %					
2.0. Careening of student's work (anality	Class attendance	0.4	Research		Pra	Practical training		[
2.9. Screening of student's work (specify the proportion of ECTS credits for	Experimental work	0.1	Report			dottoar training		
each activity so that the total number	Essav		Seminar essav			(Otherdescribe)		
of CTS credits is equal to the credit	Tests	1.0	Oral exam	1.6		(Other-describe	e)	
value of the course)):	Written exam	1.0	Project		1	(Other-describe	) e)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Projects – 30% Class attendance – 5% Written exam – 30% Oral exam – 35%							
2.11. Required literature (available at the	Title					Number of copies at the library	Ava otl	ilability via ner media
library and via other media)	Ryszard Kazimierz, Low-emission combustion, Wydawnictwo Politechniki Slaskiej,       1         Gliwice, 2002.       1							
2.12. Optional literature (at the time of the submission of the study programme proposal)								
2.13. Methods of monitoring quality that ensure acquisition of exit competences	Survey at the level of faculty Analyses provided in the syst Analyses provided in the syst	and Univers tem of quali tem of quali	sity. ity assurance of the institution. ity assurance and an authorize	d office of the	Univ	versity.		

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of
number		learning outcomes
1	Identify the type of pollutant emissions into the environment as a result of the combustion	1st colloquium, oral exam
	process.	
2	Analyse the causes of certain pollutant emissions.	1st colloquium, oral exam
3	Suggest a method for reducing certain pollutant emission.	2nd colloquium, oral exam
4	Evaluate the effectiveness of applied method.	2nd colloquium, seminar paper, oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Full Prof. Mirko Gojić, PhD Assoc.Prof. Stjepan Kožuh, PhD	1.6. Year of study	2			
1.2. Name of the course	PACKAGING MATERIALS	1.7. Credit value (ECTS)	4			
1.3. Associate teachers	-	<ol> <li>Type of instruction (number of hours L+S+E+e-learning)</li> </ol>	30+15+0+0			
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment in the course	10			
1.5. Status of the course	elective	1.10. Level of use of e-learning (1, 2, 3 level), percentage of instruction in the course on line (20% maximum)	1., 5%			
2. COURSE DESCRIPTION						
2.1. Course objectives	Introducing students with different type Clarification of interaction product-pack Training for risk avoidance or basic mis	es of packaging materials and methods of their kage-environment. stakes during the selection and application of c	making. ertain packaging materials.			
2.2. Enrolment requirements and required entry competences for the course	-					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Use the acquired theoretical knowledge in engineering practice. Analyse the development and application of new technologies.					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	Define the terms, types and classification of packaging materials. Predict the function of packaging. Compare the properties of different packaging materials. Predict of packaging for various products. Analyze the properties of metal packaging. Explain the role of recycling packaging materials.					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30): Week 1: Introduction. Definitions, importance and role of packaging. The legislation for packaging. (2) Week 2: Elements of packaging design. The classification of packaging. (2) Week 3: Packaging functions (security, storage-transport, marketing, usage, environmental). (2) Week 4: The properties of packaging materials. Classification of packaging materials. (2) Weeks 5 and 6: Metal packaging (steel, tinplate, galvanized steel, aluminum). Analysis of metal packaging. (4) Week 7: Tree. Paper, paperboard and cardboard. Glass. (2) Week 8: Polymeric materials. (2) Week 9: Surface treatment of packaging materials. (2) Week 10: Laminates (composites). Biodegradable materials. New materials. (2) Week 11: Packaging forms (cover, boxes, cans, jars etc.). (2) Week 12: The technology of production of packaging. Design and innovation. (2) Week 13: Packaging and environmental protection. Environmentally friendly packaging. (2)					

	SEMINAR (15): The selection of topics and seminar work in writing form by a mentor system (10 hours). Preparation and presentation of the seminar and discussions related to the topic of the present paper (5 hours).						
2.6. Type of instruction	<ul> <li>lectures</li> <li>seminars and wo</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>	rkshops	<ul> <li>☐ independent study</li> <li>☐ multimedia and the internet</li> <li>☐ laboratory</li> <li>☑ work with the mentor</li> <li>(other)</li> </ul>		2.7. Comments:		
2.8. Student responsibilities	Students must atten	d over 70%	6 of lectures and are required	d to complete	a seminar in writing	form and	l orally present.
2.9. Screening of student's work (specify the proportion of ECTS credits for	Class attendance Experimental work	0.5	Research Report		Practical training		
each activity so that the total number	Essay		Seminar essay	1.0	(Otherdesc	ribe)	
of CTS credits is equal to the credit	Tests	2.5	Oral exam		(Other-des	cribe)	
value of the course) <i>):</i>	Written exam		Project		(Other-des	cribe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	<ul> <li>evaluation of stude</li> <li>evaluation of writte</li> <li>evaluation of semir</li> </ul>	<ul> <li>evaluation of students activities in course,</li> <li>evaluation of written examination (two colloquiums) through continuous monitoring or final examination (written and oral),</li> <li>evaluation of seminar paper and presentation</li> </ul>					
	Title				Number of copies at the library	Availability via other media	
	M. Gojić, Metalurgija Sisak, 2006.	veučilište u Zagrebu Metalurš	15				
	M. Gojić, Površinska Sisak, 2010.	Sveučilište u Zagrebu Metalur	10				
2.11. Required literature (available at the	N. Stipanelov Vrand Kemijsko tehnološki	alaža-interna skripta, Sveučili plit, 2010.			Electronic form		
	K. Galić, N. Ciković, Priručnici Sveučilišta	K. Galić, N. Ciković, K. Berković, Analiza ambalažnog materijala-skripta, Priručnici Sveučilišta u Zagrebu, HINUS					<u>/w.hinus.hr/wp-</u> knjige/2011/10/ANALIZA- AZNOG- JALA.pdf
	W. D. Callister, D.G. introduction, John W	n, Materials Science and Eng s, inc. Hoboken, 2010.	1				
						×	7 . 7
2.12. Optional literature (at the time of the submission of the study programme proposal)	<ul> <li>S. Rocco, Upravljanje proizvodom, kreiranje marke i dizajn-elektronska skripta, Visoka poslovna škola Zagreb, Zagreb, 2015.</li> <li>N. Stričević, Suvremena ambalaža 2, Školska knjiga, Zagreb, 1983.</li> <li>N. Stričević, Suvremena ambalaža 1, Školska knjiga, Zagreb, 1982.</li> <li>I. Budak, J. Hodolič, M. Stević, Đ. Vukelić, B. Kosec, B. Karpe, Označavanje proizvoda o zaštiti životne sredine, Fakultet tehničkih nauka Univerziteta u Novom sadu, Novi Sad, 2009.</li> <li>Scientific and professional papers in referred journals and conference proceedings.</li> </ul>						

2.13. Methods of monitoring quality that	Input and output students survey. Survey at the level of faculty and University.
ensure acquisition of exit competences	Analyses provided in the system of quality assurance of the institution.
•	Analyses provided in the system of quality assurance and an authorized office of the University.

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning
number		outcomes
1	Define the terms, types and classification of packaging materials.	1st colloquium, written and oral exam
2	Predict the function of packaging.	1st colloquium, written and oral exam
3	Compare the properties of different packaging materials.	1st colloquium, written and oral exam, seminar paper
4	Predict of packaging for various products.	1st and 2nd colloquium, written and oral exam, seminar paper
5	Analyze the properties of metal packaging.	2nd colloquium, written and oral exam
6	Explain the role of recycling packaging materials.	2nd colloquium, written and oral exam

1. COURSE DECRIPTION – GENERAL INFORMATION ISVU CODE:						
1.1. Course teacher	Assoc.Prof. Ivan Brnardić, PhD Assoc.Prof. Tamara Holjevac Grgurić.	, PhD 1.6. Year of study	1.6. Year of study 2			
1.2. Name of the course	LIGHT POLLUTION	1.7. Credit value (ECTS)		4		
1.3. Associate teachers	-	1.8. Type of instruction (n hours L+S+E+e-learn	umber of ing)	30+15+0+0		
1.4. Study programme (undergraduate, graduate, integrated)	graduate	1.9. Expected enrolment i	n the course	10		
1.5. Status of the course	elective	1.10. Level of use of e-lea level), percentage of the course on line (20	arning (1, 2, 3 instruction in 0% maximum)	1., 5%		
2. COURSE DESCRIPTION						
2.1. Course objectives	<ol> <li>To explain the key concepts of light pollution, with particular emphasis on light pollution from industrial entities.</li> <li>To introduce the principles of protection from light pollution and tasks of entities that carry out protection and resource conservation.</li> </ol>					
2.2. Enrolment requirements and required entry competences for the course	Knowledge on environment protection and work on computers.					
2.3. Learning outcomes at the level of the study programme to which the course contributes	Analyse the development and application of new technologies. Combine social, ethical and business principles and norms in the professional field. Combine the skills necessary for lifelong learning, including continued professional training.					
2.4. Expected learning outcomes at the level of the course (4-10 learning outcomes)	<ol> <li>Define the light pollution.</li> <li>Explain the impact of light pollution on wildlife, human health and workers.</li> <li>Explain the principles of protection against contamination by light.</li> <li>Explain the obligations of economic operators according to the legislation.</li> <li>Select procedures to avoid and reduce light pollution.</li> </ol>					
2.5. Course content broken down in detail by weekly class schedule (syllabus)	LECTURES (30):         Light as a source of environmental pollution, light pollution from urban areas and industry. 3h         The impact of light pollution on wildlife and the environment and to human health. 3h         The link between light pollution and health of workers in the industry. 3h         Principles of light pollution protection. 3h         The Law on the light pollution protection. 3h         I Entities to ensure the implementation of light pollution protection. 3h         Standards of illumination and mandatory methods to illuminate. 3h         Protection measures, restrictions and prohibitions. 3h         Planning, construction, maintenance and reconstruction of lighting, the responsibility of manufacturers of products that serve the clarification. 3h         Administrative supervision and inspection. 3 hours         SEMINAR (15):         Example seminar work and preparation and presentation of seminar paper.					
2.6. Type of instruction	⊠ lectures	independent study	2.7. Commen	its:		

	<ul> <li>seminars and workshops</li> <li>exercises</li> <li>online in entirety</li> <li>mixed e-learning</li> <li>field work</li> </ul>		multimedia and the internet laboratory work with the mentor (other)			nternet			
2.8. Student responsibilities	Regular attendance of lectures, written and orally presented seminar work.						_		
2.9. Screening of student's work (specify the proportion of ECTS credits for each activity so that the total number	Class attendance	0.5	Research				Practical training		
	Experimental work		Report						
	Essay		Seminar essa	ay	1.0		(Otherdea	scribe)	
of CTS credits is equal to the credit	Tests		Oral exam		1.25		(Other-de	escribe)	
value of the course)):	Written exam	1.25	Project				(Other-de	escribe)	
2.10. Grading and evaluation of student work over the course of instruction and at a final exam	Attendance on classes, pr and oral exam.	reparation a	and presentatio	n of seminar paper, continuous monitoring – 2 preliminary exams or wr			ry exams or written		
2.11. Required literature (available at the library and via other media)	Title		Nun cop at t	nber of oies the rary	Availability via other media		lia		
	I. Brnardić, Lectures from Light pollution, Sisak, 2016.				Internet – Merlin system for e-learning			earning	
	The Law on the light pollution protection, OG 114/11, RH, 2011.					http://narodne- novine.nn.hr/clanci/sluzbeni/2011_10_114_2221.html			
	T. Sofilić, Pravo okoliša, Skripta, Metalurški fakultet, Sisak, 2015.					http://www.simet.unizg.hr/nastava/predavanja/preddiplomski- sveucilisni-studij-metalurgija/3-godina-preddiplomskog- studija/pravo-okolisa/view			
	D. Božičević, The influence of light pollution on the envirionment and human, Rad u zborniku sa znanstvenog skupa Lječilišna medicina i turizam, Veli Lošinj, 2010., 56-61.				https://www.google.hr/url?sa=t&rct=j&q=&esrc=s&source=web &cd=3&cad=rja&uact=8&ved=0ahUKEwiEvv7kyPzNAhVBsBQ KHY9XCwoQFggsMAI&url=http%3A%2F%2Fhrcak.srce.hr%2 Ffile%2F107383&usg=AFQjCNHfFxpuFedTnyfkzicxkP0i25No 7w				
	Ž. Andreić, K. Korlević, D. Andreić, A. Bonaca, P. Korlević, M. Kramar, Light pollution in Croatia, Građevinar 63 (2011) 8, 757-764.				https://w &cd=1& QKHdsF %2Ffile No7w	ww.google.hr/url cad=rja&uact=88 PDQQQFggcMA/ %2F107383&usg	i?sa=t&rct=j&q=&e: &ved=0ahUKEwjcg A&url=http%3A%2f =AFQjCNHfFxpuF	src=s&source=web NKCyfzNAhUDXB 5%2Fhrcak.srce.hr edTnyfkzicxkP0i25	
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2.12. Optional literature (at the time of	Available scientific literature on the subject of light pollution.			
the submission of the study				
programme proposal)				
2.13. Methods of monitoring quality that	Survey on the faculty and University level.			
ensure acquisition of exit	Analysis predicted by systems for insurance of institution quality.			
competences	Analysis predicted by systems for insurance quality from authorized University office.			

Ordinal	Expected learning outcomes at the level of the course (4-10)	Methods for monitoring of the achievement of learning
number		outcomes
1	Define the light pollution.	1st colloquium, seminar, written and oral exam
2	Explain the impact of light pollution on wildlife, human health and workers.	1st colloquium, seminar, written and oral exam
3	Explain the principles of protection against contamination by light.	1st colloquium, seminar, written and oral exam
4	Explain the obligations of economic operators according to the legislation.	2nd colloquium, seminar, written and oral exam
5	Select procedures to avoid and reduce light pollution.	2nd colloquium, seminar, written and oral exam